Effect of Powder Characteristics on Flash Sintering of Hydroxyapatite

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Abstract

A comparative study on flash sintering of hydroxyapatite (HA) with two different powder characteristics is presented in this work. Electric fields up to 2000 V cm⁻¹ were tested, resulting in sintering at temperatures as low as 850°C and densit<u>y in excess toies</u>. 90%. XRD and FTIR analysis show no decomposition of HA into other phases for conventional as well as for flash sintering. In general, the conductivity curve for HA is found to be different from that of other materials₇ and this is attributed to the microstructural changes taking place due to the dehydroxylation reaction <u>happening-occurring</u> during sintering. Powder characteristics are shown to influence the conductivity curve and also the onset temperature for flash, although the overall character of the curve remains similar. Sintering conditions that lead to the partial melting of the samples exhibit (002) preferred oriented microstructure.

Key words

Flash sintering, Hydroxyapatite, Powder characteristics, Biomaterial

Introduction

Hydroxyapatite (HA) is a commonly used bio-ceramic because of its chemical and structural similarities with bone tissue resulting in superior biocompatibility [1] [2] [3] [4]. HA is typically produced as a powder that is formed by many different ceramic processing techniques into a low density, low strength 'green' body which is then densified by heat treatment. To obtain a high degree of densification and high mechanical strength, the material must be subjected to high temperature, typically over 1200°C. Such a sintering process improves the mechanical properties but can activate other processes resulting in grain growth, phase separation and decomposition [5].

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To enhance <u>densification and the</u>-mechanical properties <u>and densification of</u>+ HA, several sintering techniques have been <u>studiedinvestigated</u>. The addition of small amounts of a low melting sintering aids increases diffusion rates and improves the sinterability of HA [6]. Other sintering methods such as hot pressing (HP) [7] [8] and hot isostatic pressing (HIP) [9] deployed to enhance the densification of HA use externally applied pressure and thermal energy. Other approaches, known as Field Assisted Sintering Techniques (FAST), use electromagnetic fields or microwaves to enhance diffusion rates [9] [10]. Microwave sintered HA possesses a higher density and exhibits much lower grain growth compared to the <u>onethe material</u> sintered conventionally. HA has also been sintered at lower temperature by spark plasma sintering (SPS) where electric field, thermal energy and pressure are combined [11].

Powder characteristics have been shown to considerably influence the flash phenomena on 3 mol% yttria-stabilized zirconia [20], and therefore it is worth studying the effect of powder characteristics on flash sintering of HA. In this work, we present a study on flash sintering phenomena at different applied E-fields on two HA powders with different powder characteristics.

Materials and Methods

Two commercially available HA powders were used in the present work. One is a spray-dried powder (Fluidinova, S.A., Engenharia de Fluidos, Portugal), here on called HA-A. The other is an agglomerated powder (Eurocoating, Italy), here on called HA-B. Each powder was mixed

with 3 wt% polyacrylate binder (Duramax B-1000, Rohm and Haas France SAS, BP, France) and uniaxially pressed into dog-bone shaped pallets similar to those shown in [21] [22]. Pressing-Uniaxial pressing with 120 MPa within a steel mould (REF) resulted in an initial relative green density of~44% (theoretical density for HA = 3.16 g cm⁻³) which was obtained from dimensional parameters and weight-mass of the sample. The effective dimensions of the rectangular section of the dog bone specimen were 21 mm between the electrodes x 3 mm width x 1.80 ± 0.03 mm thickness.

The field-assisted sintering setup and procedure were similar to those describedin [22]. The sample was suspended in the hot zone of a sealed horizontal tubular furnace (HTRH 100-300/18, GERO Hochtemperaturöfen GmbH, Germany) by two platinum wires (diameter = 0.5 mm) run through the holes in the handles of the dog bone_<u>specimen</u>. The electrodes served to hold the sample freely and to apply the necessary voltages and required currents. The sample shrinkage was monitored by a CCD camera through a silica window at one end of the alumina tube. The furnace temperature was monitored by an S-type thermocouple kept in the hot zone of the furnace, at about 10 mm from the sample, to minimize the error due to <u>Joule_heating</u> of sample <u>resulting from the Joule heating-[11]</u>. Artificial dry air was fluxed through the furnace chamber for the duration of each experiment to minimize arcing issues and to maintain a dry atmosphere. All treatments were carried out with a constant heating rate of 10°C min⁻¹ up to 1400°C or until a flash sintering event occurred. After sintering, the density was determined by the Archimedes' method.

Experiments were carried out with and without the application of an electric field (E-field). The required voltage (V) to achieve the desired E-field (E) based on the initial electrode separation ($L_0 = 21$ mm) and determined as E=V/L₀, was applied at 600°C after complete burnout of <u>the</u> binder from the samples. Voltage The voltage was maintained constant throughout the sintering treatment. The current was monitored by a digital multimeter and current density (J) was determined based on the initial sample cross-section (J=I/(thickness x width)). The volumetric power dissipation was calculated as the product of E-field and current density (P=E x J). When the current density abruptly increased in the "flashing" dog-bone sample, the power supply and furnace were turned off and the sample was allowed to cool at the furnace rate.

——The morphology of starting powder and microstructures of sintered samples were examined by SEM (JEOL JSM-5500E). The particle size distribution of the powders was

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obtained from the analysis of SEM images using Image J software. Crystalline phases were analyszed before and after sintering by XRD (Rigaku DMax - Bragg-Brentano configuration). The samples were exposed to Cu K-α radiation with steps of 0.02° with hold time of 6 s at each scan of 2 theta from 20° to 55°. Fourier Transformation Infra-Rred (FTIR) spectrometry was used to compare the dehydration of HA before and after sintering (Nicolet Avatar 330, Thermo Electo Corporation, Waltham, MA.); crushed samples were mixed with KBr and scanned at an average rate of 64 scans s⁻¹ with 4 cm⁻¹ resolution.

Results and Discussion

Powder Properties

The morphology of two-HA powders are shown in figure 1. HA-A powder has a spherical morphology, whereas HA-B powder consists of hard agglomerates which are irregular in shape. Figure 2 shows the particle size distribution for the two powders. HA-A has a narrower <u>particle</u>-size distribution with an average particle size of 1.33 μ m (standard deviation = 0.47 μ m)_{ir} HA-B on other hand <u>has-is characterized by</u> a wider particle size distribution with an average particle size distribution = 1.11 μ m).

Sintering:

Shrinkage strain as a function of furnace temperature at different applied E-fields for the HA compacts is shown in figure 3. For conventional sintering, it can be seen that the two powders show different sintering behaviors. Dependence—The dependence of sintering behavior (for conventional sintering) of HA on powder morphology has been <u>already</u> shown in previous works [23] [24]. HA-A shows better sinterability <u>for-upon</u> conventional sintering with a final shrinkage strain of -24.8 % in comparison to HA-B <u>with-characterized by</u> a final shrinkage strain of 23.1 %, <u>as</u> <u>-this is</u>-reflected <u>in-by</u> the<u>ir</u> final density reported in table 1. The characteristics of the powders used in this study are comparable to <u>the onesthose</u> studied in [23]. It was shown that <u>the HA</u> powder with spherical morphology, smaller particle size and a narrower size distribution ended up with higher final density when sintered conventionally at 1400°C, <u>this is</u> in agreement with the current study.

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In <u>the</u> presence of an applied E- field, the sintering process occurs at a faster rate. The onset of sintering is shifted to a-lower temperature with the applied E-field for compacts of both the powder<u>s</u>-types. For the applied<u>Under an</u> E-field of 500 V cm⁻¹, although the sintering rate increases, the character of the sintering curve is similar to <u>that-ofthe</u> conventional <u>sintering one</u> for both <u>the</u>-powder morphologies. For HA-B, the extremely high sintering rate (wherein the densification occurs within seconds) is observed for the applied E-field of 1500 V cm⁻¹ and above_{ir} however, for HA-A, such extremely high sintering rate is observed only at <u>the</u>-applied E-field of 2000 V cm⁻¹.

Table 1 shows the final density of all the sintered samples. It should be noted that the variation in densities for samples sintered in the presence of applied E-field is likely a result of the uncontrolled finishing conditions. The final density is a function of the final current densityies and the time for which the sample is kept in the flash sintering conditions [25]. In our case, these conditions were not controlled; therefore, even a few seconds more in flash conditions even at lower applied E-field can result in higher densities due to the extremely high diffusion rate.

Microstructure

The micrographs of a fractured surface of sintered samples are-is_shown in figure 4. It can be seen that a nearly dense microstructure with few pores and grains in the micron range is obtained for all the samples regardless of-the powder type and the applied E-field. HoweverNevertheless, HA-A sintered at an applied E-field of 500 V cm⁻¹ shows a unique microstructure that is different from the rest of the other flash sintered samples. The unique nature of the microstructure could be very likely because accounted for by of-partial melting, this at is discussed in more detail in the furthera following section of-(XRD analysis). Again, the variations in microstructure for sintering at different applied E-field can be attributed to the uncontrolled finishing conditions, as previously discussed-earlier.

Power / electrical conduction:

——Figure 5 shows the power density curves plotted in Arrhenius format for flash sintered HA sample from 700°C onwards. The power curves in flash sintering studies offor HA flash sintering are different from the onesthose recorded onof other materials [17]. In the present study, two different types of power curves are seenobserved. For the samples

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sintered with-under_high applied-E-field (for HA-A sintered with-under_2000 V cm⁻¹ and for HA-B sintered with-under_2000 V cm⁻¹ and -1500 V cm⁻¹), there is a constant <u>rise-increase</u> in conductivity along with the heating process, followed by a sudden <u>conductivity increase</u> <u>intensification in conductivity at a point</u> where the flash event occurs. For <u>the rest of the the</u> <u>other</u> combinations of <u>HA</u> powder type and applied E-field the conductivity increases with the thermal process, <u>howeveralthough</u>, the curves show a 'wavy' nature beginning from~850°C, <u>which is then</u> followed by <u>the a</u> sudden <u>conductivity</u> increase <u>in conductivity</u>-leading to the flash event.

Power density plots in Arrhenius-like diagram format for the two powder types at the same applied E-fields are shown for the two powders in figure 6. It can be seen that the temperature for the flash onset of flash-is lower for HA-B for the same applied E-field, except for 2000 V cm⁻¹; (this is represented in figure 7). The fact that the power density plot does not coincide for the two powders (although the character of the curves is similar), infers that the conductivity observed is not only dependent on the intrinsic resistance of the material but relies also depends on the compact resistance of the compact [20]. In the sense thatother words, powders with different powder characteristics will beare packed differently in the green sample component, this leading to different electrical resistance. Initially, at low temperatures, the compacts of HA-B have-are more resistanceresistive. We speculate that the powder characteristics-HA-B powder is such that, as the temperature increases and as the sintering process continues resulting in ainduces changes in the microstructure, the resistance of the <u>HA-B</u> compacts of <u>HA-B</u> decreases. As a result of this <u>resistance</u> decrease in resistance, the conductivity of HA-B increases (the 'wavy' nature begins) and exceeds that of HA-A at the particular temperature, that is, and a crossover takes place as can be seen shown in figure 6. The temperature at which the cross-over takes place is seen to decreases with the increasingthe applied E-field, this being probably related to faster which could be because sintering is faster at higher applied E-field which induces quicker, the higher degree of sintering means the microstructural changes will take place faster. This Said crossover causes the temperature for the onset of flash to be lower for HA-B for the same applied E-field. For the applied E-field of At 2000 V cm⁻¹, the flash for HA-A occurs at such a low temperature that the crossover does not take place.

From the analysis of figures 5 and figure 6_7 and from the above discussion the following points in view of flash sintering of HA can be inferred: 1) Regardless regardless of the powder

type and the applied E-field, all the power curves would have a similar character (constant increase in conductivity followed by 'wavy' nature) if the flash event had not to occur, 2) The the resistance of the compact influences the power curve, although the character of the curve remains similar, 3) The the powder characteristics play an important role in deciding determining the temperature for the flash onset of flash.

Thermal runaway and the generation and <u>movement of defects formation/diffusion</u> in the material are <u>typically considered as</u> the main reasons for the occurrence of flash event. With the increase in<u>As</u> temperature <u>rises</u>, the defects concentration also increases, and, when the defects concentration exceeds <u>the a</u> threshold value, the current flows through the material, <u>which causesthis causing</u> the generation of heat <u>inside within</u> the sample. The further increase in temperature due to the <u>generated</u> heat <u>generated</u> further increases the defects concentration, resulting in a synergistic effect of increasing heat <u>generationproduction</u>. Such a chain reaction causes the flash event [25] [18]. For HA, the major <u>fundamental</u> defects which contribute to electrical conduction are hydroxide (OH⁻) and hydrogen <u>ions</u> (H⁺) <u>ions</u> [26] [27] [28] [29]. The dehydroxylation reaction occurring at high temperatures (at 800°C or above) [30] in HA is considered to be the source for the generation of <u>these-such</u> defects [26]. <u>This Such</u> dehydroxylation reaction can <u>be given asexpressed as</u>:

$Ca_{10}(PO_4)_6(OH)_2 \rightarrow Ca_{10}(PO_4)_6(OH)_{2-2x}O_x(\Box)_x + xH_2O_{(g)} \uparrow (1)$

Where where x<1 and (\Box) is the vacancies at the hydroxyl site. From Eq.uation 1, follows that for every two moles of released H₂O-released, one mole of H⁺ vacancy-vacancies and one mole of OH⁻ vacancy-vacancies is are created. Although it is not very clear which of the two defects is the major contributor for electrical conduction at high temperatures for HA, it is obvious that dehydroxylation is an important reaction as far as the generation of these defects is concerned [19] [29]. As the-temperature increases, more and more of these defects are generated due to dehydroxylation reaction and when the threshold of these defects concentrations is exceeded, the current flows, leading to the chain reaction and the flash event to occurs, as discussed earlier [18]. However, it should be kept in mind that the high sintering rates observed in flash sintering can not be explained exclusively by the temperature effect, that is, a higher diffusion rate due to an increase in temperature can not alone explain such high sintering rates by itself[11] [31]_{i7} therefore, there should be a fundamental change in sintering mechanism causing the <u>observed</u> high sintering rates-observed. Referring to figures 5 and figure-6 and to the following reported discussion-on it, the question that remains to be answered is: what is the reason for the microstructural change to occur which leads to the 'wavy' nature of the conductivity curve, which otherwise should be linear in that temperature range as reported in a previous work [29]?- Note-One can observe that the conductivity measured up to 1000°C in [29] were-corresponds to on the dense and porous HA ceramics obtained after sintering at 1300°C; therefore, so-it is expected that no microstructural changes happen-occur while measuring the conductivity up to 1000°C, giving leading to a the linear increase in conductivity with the temperature.

Liu et al. [30] prepared-produced_dense (relative density of = 99.8%) HA ceramic by sintering-at 925°C by spark plasma sintering technique. Thermal annealing carried out on these such samples at 1000°C and 1100°C revealed a considerable density decrease-reduction (-~2.5% and ~14% decrease at 1000°C and ~14% decreaseand at 1100°C, respectively) accompanied with no weight loss. The microstructure of the samples was found to become porous, with the-porosity increasing with annealing temperature. The decrease in density (accompanied with no weight loss) and the pores formed-formation were attributed to the dehydroxylation reaction (Eq.equation 1) which takes place at high temperatures leading to the formation of water vapour. Therefore,So according to Liu et al, at the annealing temperatures, the dehydroxylation reaction leads to pore formation with water vapour being trapped in the closed pores, and, as the cooling occurs, rehydration takes place (water enters the lattice) keeping the pores as it is they are.

<u>On this basis, We propose the following towe can</u> explain the non-linear increase and then <u>successive</u> the decrease of conductivity in the 'wavy' region of the <u>conductivity curveas</u> follows. As the temperature increases, more and more water vapour is released due to the dehydroxylation reaction. When the sample achieves sufficient density due to sintering, the conductivity increases; <u>but</u>, <u>however</u>, at the stage when the sample is dense enough, the closed pores containing water vapour start forming <u>inside within</u> the sample <u>which this</u> causing <u>aes conductivity</u> the decrease <u>in conductivity</u>. Further <u>density</u> increase <u>in density</u> due to sintering <u>again</u> increases <u>again</u> the conductivity <u>which and</u> then decreases due to pores formation, giving the next wave, <u>--</u><u>T</u><u>t</u><u>h</u> is process continuinges <u>till-until</u> the flash event occurs.

X-ray diffractometry (XRD) analysis

The XRD patterns of as-received powders and sintered samples for both the powder types are shown in figure 8. The patterns of both the powders and all their sintered samplespeaks account for HA (JCPDS file 09-0432), only, and. No-no traces of other phases of calcium phosphates phases are observed [17], this indicating that no phase-transformation occurred for conventional sintering as well as for flash sintering at all the applied E-fields. These observations are in agreement with previous works on conventional and flash sintering of HA [23] [24] [18]. One can observe that, regardless of the powder type and applied E-field, the sintering is responsible for narrower peaks indicating which point out the formation of well-developed micron-size grains as <u>already</u> shown before in figure 4. The broad peaks for the powders indicate instead that both the powdersthese are poorly crystalline <u>______narrow</u> diffraction peaks for powder would indicate crystalline material which is obtained due to calcination/heat treatment of the powder [24].

(211) diffraction peak is the most prominent for HA. Some diffraction peaks for different sintering conditions show intensities that are not in good agreement with that of standard HA with randomly oriented and grown grains. (002) peak for HA-A sintered at-under applied E-field of -500 V cm⁻¹ is one such peak which showsing a very largecertain deviation deviation; the intensity ratio of intensities of (002) and (211) peaks (I_{002}/I_{211}) is seen to be 1.9 which-while actually it should be ~0.4 for randomly oriented grains [32]. Such high relative intensity for the (002) peak suggests the (002) preferred orientation and the crystallite c-axis perpendicular to the sample surface [33] [32] [34] [35]. Figure 9 shows the I₀₀₂/I₂₁₁ for samples sintered at different applied E-field for both the powder types. It can be seen that only HA-A sintered at applied E-field of under 500 V cm⁻¹ shows (002) preferred orientation. Previous works on flash sintering of HA did not show any such (002) preferred orientation [17] [18]. There are two important differences between the previous and the present current study. 1) In the previous studies, once the flash event occurrsed, the power supply was is set to switch to current control mode when the current reachesed the pre-set value, and the samples were is kept in those such conditions for a maximum of 10 seconds. In the present study, no such pre-set current values were decided; the power setup was turned off manually once the flash event occurred. 2) The maximum sample temperature reached due to the flash event was ~1250°C in previous studies [REF], whereas in this the present work study the maximum furnace temperature for on-set of the flash event is 1317°C which is(-for HA-A sintered with applied E-field of under 500 V cm⁻¹-). Note It is important to point out that the temperature

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reported for this-studythe present study is the furnace temperature and not the sample temperature. The actual temperature of the sample will-beis definitely higher than 1317°C due to Joule heating which will further rise instantaneously much above during stage II [17] [18]. Coming Going back, referring to figure 5, we one can see that the maximum power dissipation was allowed recorded for HA-A sintered at applied E-filed of 500 V cm⁻¹ and, a the combination of such-high current flow and high temperature (as said, above 1317°C, and which will further rise instantaneously) can cause partial melting of HA sample [18]. We One speculate that partially melted HA recrystallized with (002) preferred orientation for HA-A sample sintered with applied E-field of 500 V cm⁻¹. Such (002) preferred orientation-has been already reported previously [32] [34]. The possible reason for this could be that the C-plane of hexagonal crystal has a closed-packed atomic arrangement and hence C-plane terminated surface is preferred to curtail the surface free energies [35].

FT-IR Analysis

The stability of HA under conventional and flash sintering was also-confirmed <u>also</u> by FT-IR analysis. Figure 10 shows IR spectra of as-received powders and samples sintered at <u>under</u> different applied E-fields. As expected, the IR spectra of both <u>the</u>-powder types are the same. The sharp bands at around 3570 cm⁻¹ correspond to the stretching of isolated OH species that are coordinated in HA. Weak signals at around 1635 cm⁻¹ account for the bending of OH species. The broad band between 3000 cm⁻¹ and 3700 cm⁻¹ observed for both <u>the</u> powders is assigned to physisorbed water molecules_{*x*} and it can be seen that this band weakenings or <u>even</u> disappearings for sintered samples. Strong absorption bands are <u>seen</u> <u>visible</u> below 1500 cm⁻¹ which these being are due to asymmetric stretching (γ_3), symmetric stretching (γ_1) and asymmetric bending (γ_4) of PO4⁻³ unit [35]. The broadening of $\gamma_3 PO4^{-3}$ signal for sintered samples compared with that of powder indicates that the structure of PO4⁻³ is disordered or <u>that</u> oxygen-deficient PO_x (x<3) <u>are present</u> in sintered samples [36]. Overall, the IR analysis confirms that regardless of the powder type, HA does not show any type of major structural rearrangement or decomposition upon conventional and flash sintering.

Conclusion

HA with two different powder characteristics werewas successfully densified by flash sintering starting from with two different powders, considerably reducing the sintering temperature and the processing time. HA is found to be stable for-during both conventional as wells as and flash sintering conditions, as and no decomposition of HA into other phases is observed, as indicated by XRD and FT-IR analysis. For the samples sintered with aunder high applied E-field, the conductivity curve for HA-is found to be different from that of other materials. At lower temperatures, the conductivity increases constantly with the heating processtemperature, howeverbut, at a temperature above ~850°C the curves shows a 'wavy' region. The 'wavy' region in the conductivity curves This can be explained by considering the microstructural changes taking place due to the dehydroxylation reaction happening occurring during sintering. The onset temperature for the flash event is found to be different for the two powders_types-sintered at the same applied E-field. This is because the powder characteristics influence the conductivity curves, although the overall character of the curves remains similar. A combination of high furnace temperature and high current densitives passing through the sample (which will further increase the sample temperature) is shown to cause partial melting of the samplematerial which, - The samples solidified after partial meltingupon the successive solidification, is shown to possess grow along (002) preferred oriented-microstructure.

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Tables

Electric field (V cm ⁻¹)		0	500	1000	1500	2000
Relative	HA-A	96	94	93	95	92
density (%)	HA-B	91	93	90	90	95

Table 1. Final density of samples sintered at different applied E-field for HA-A and HA-B

Figures



Figure 1. Micrographs of (a) HA-A and (b) HA-B powders.



Figure 2. Particle size distribution for HA-A and HA-B powders.



Figure 3. Shrinkage strain as a function of furnace temperature at different applied E-fields for (a) HA-A and (b) HA-B.



Figure 4. Microstructures of HA-A sintered at (a-e) 0-2000 Vcm⁻¹ and of HA-B sintered at (f-j) 0-2000 V cm⁻¹ applied E-field.





Figure 5. Arrhenius plots of power density of flash sintered samples under different applied E-fields for (a) HA-A and (b) HA-B.

Figure 6. Arrhenius plot of power density for HA-A and HA-B flash sintered at applied E-field of (a) 500 V cm^{-1} (b) 1000 V cm^{-1} (c) 1500 V cm^{-1} (d) 2000 V cm^{-1} .





Figure 7. Temperature for onset of flash for different applied E-field for HA-A and HA-B.

Figure 8. XRD spectra of as received powders and samples sintered at different applied E-field for (a)HA-A and (b) HA-B.







Figure 10. Infrared spectra of (a) HA-A and (b) HA-B at different applied E-fields.