| 1 | Growth media affect the volatilome and antimicrobial activity against <i>Phytophthora infestans</i> |
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| 2 | in four <i>Lysobacter</i> type strains |
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23 ABSTRACT

24 Bacterial volatile organic compounds (VOCs) play important ecological roles in soil microbial 25 interactions. Lysobacter spp. are key determinants of soil suppressiveness against phytopathogens 26 and the production of non-volatile antimicrobial metabolites has been extensively characterised. 27 However, the chemical composition and antagonistic properties of the Lysobacter volatilome have 28 been poorly investigated. In this work, VOC emission profiles of four Lysobacter type strains grown 29 on a sugar-rich and a protein-rich medium were analysed using solid-phase microextraction gas 30 chromatography-mass spectrometry and proton transfer reaction-time of flight-mass spectrometry. 31 Lysobacter antibioticus, L. capsici, L. enzymogenes and L. gummosus type strains were recognised 32 according to their volatilome assessed using both headspace mass spectrometry methods. Moreover, 33 the chemical profiles and functional properties of the Lysobacter volatilome differed according to the 34 growth medium, and a protein-rich substrate maximised the toxic effect of the four Lysobacter type 35 strains against Phytophthora infestans. Antagonistic (pyrazines, pyrrole and decanal) and non-36 antagonistic (delta-hexalactone and ethanol) VOCs against Ph. infestans or putative plant growth 37 stimulator compounds (acetoin and indole) were mainly emitted by Lysobacter type strains grown on 38 protein- and sugar-rich media respectively. Thus nutrient availability under soil conditions could 39 affect the aggressiveness of Lysobacter spp. and possibly optimise interactions of these bacterial 40 species with the other soil inhabitants.

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Keywords: volatile organic compounds, *Lysobacter* spp., biological control, *Phytophthora infestans*,
SPME/GC-MS analysis, PTR-ToF-MS analysis

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• Four *Lysobacter* type strains can be recognised according to their volatilome

• Composition and properties of the *Lysobacter* volatilome differ according to the growth media

• Protein-rich media increase the mission of antagonistic compounds by *Lysobacter* spp.

49 **1. Introduction**

50 Microorganisms produce a wide variety of secondary metabolites, including antibiotics, toxins, 51 pigments and volatile organic compounds (VOCs). Volatile organic compounds are molecules of high 52 vapour pressure and low molecular weight that readily diffuse through water- and gas-filled pores in 53 soil environments (Schmidt et al., 2015). VOCs emitted by bacteria belong to different chemical 54 classes (alcohols, aldehydes, alkenes, benzenoids, ethers, lactones, ketones, terpenoids and sulphur 55 compounds) and are generated by complex metabolic pathways (Audrain et al., 2015; Schmidt et al., 56 2015; Schulz, 2007). Bacterial VOCs play essential ecological roles in communications with soil 57 microorganisms, nematodes, insects and plants (Effmert et al., 2012; Kai et al., 2009). Notably, bacterial VOCs can inhibit spore germination and mycelial growth of several phytopathogens (De 58 59 Vrieze et al., 2015; Kai et al., 2007; Weisskopf, 2013), promote plant growth (Blom et al., 2011; Ryu 60 et al., 2003) and induce plant resistance (Lee et al., 2012; Ryu et al., 2004). The chemical composition 61 of the bacterial volatilome is defined by genetic determinants and can be used as a chemotaxonomic 62 marker in standardised conditions (Peñuelas et al., 2014). However, composition and functional 63 properties of the bacterial bouquet are influenced by the nutrient source where bacteria are grown 64 (Asari et al., 2016; Blom et al., 2011; Bruce et al., 2003; Fiddaman and Rossall, 1994; Garbeva et al., 2014; Weise et al., 2012), indicating metabolic changes in VOC production according to nutrient 65 66 availability and growth conditions in the soil (Insam and Seewald, 2010).

67 Bacteria belonging to the Lysobacter genus are frequently found in soil and increased disease 68 suppression of soil phytopathogens correlated significantly with increased populations of L. antibioticus, L. capsici and L. gummosus (Postma and Schilder, 2015; Postma et al., 2008). The 69 70 Lysobacter genus (Christensen and Cook, 1978) includes species that are efficient antagonists of 71 phytopathogens and potential candidates for biological control of crop diseases (Hayward et al., 2010; Kobayashi and Yuen, 2007). In particular, L. antibioticus DSM 2044^T (ATCC 29479), L. enzymogenes 72 DSM 2043^T (ATCC 29487) and L. gummosus DSM 6980^T (ATCC 29489) were described as 73 74 Lysobacter type strains by Christensen and Cook (1978), and the antagonistic mechanisms of these

75 species have been extensively characterised (Folman et al., 2004; Folman et al., 2003; Ko et al., 2009; Qian et al., 2009; Yu et al., 2007). For example, L. antibioticus HS124 produced lytic enzymes and a 76 77 toxic compound against Phytophthora capsici (Ko et al., 2009). Likewise, the production of lytic 78 enzymes and antibiotics was shown for L. enzymogenes 3.1T8 (Folman et al., 2004; Folman et al., 79 2003), L. enzymogenes C3 (Yu et al., 2007) and L. enzymogenes OH11 (Qian et al., 2009) against 80 Fusarium graminearum, Pythium aphanidermatum, Py. ultimum, Ph. capsici, Rhizoctonia solani and 81 Sclerotinia sclerotiorum. The antagonistic properties of L. gummosus were associated with proteolytic 82 degradation of biofilm (Gokcen et al., 2014) and biosynthesis of antifungal metabolites (Meyers et al., 1985). Furthermore, the type strains L. capsici DSM 19286^T (YC5194) (Park et al., 2008) and L. 83 84 capsici AZ78 (Puopolo et al., 2014a; Puopolo et al., 2014b; Puopolo et al., 2016) produced secondary 85 metabolites that inhibit the growth of phytopathogenic fungi (Botrytis cinerea, Colletotrichum gloeosporioides, F. oxysporum and R. solani) and oomycetes (Ph. infestans, Plasmopara viticola and 86 Pv. ultimum) respectively. 87

88 Although the production of extracellular lytic enzymes (proteases, glucanases, chitinases and 89 cellulases) and antimicrobial compounds (pyrazines, tetramic acid-containing macrolactams and 90 other antifungal factors) has been widely characterised in Lysobacter spp. (Puopolo et al., 2014a; Xie 91 et al., 2012), the possible contribution of VOCs in antagonistic processes has been poorly 92 investigated. The limited studies available (Sang et al., 2011; Zou et al., 2007) suggest a significant 93 potential for VOC-mediated antagonistic processes. In particular, the VOCs emitted by L. gummosus 94 KCTC 12132 and L. enzymogenes ISE13 inhibited mycelial growth of nematicidal fungi (Paecilomyces lilacinus and Pochonia chlamydosporia) (Zou et al., 2007) and phytopathogenic 95 96 microorganisms (C. acutatum and Ph. capsici) (Sang et al., 2011) respectively.

97 The aim of this study was to elucidate the antagonistic potential of *Lysobacter* spp., based on a 98 better understanding of the emission profiles and functional properties of VOCs. Therefore, we used 99 four *Lysobacter* type strains (*L. antibioticus* DSM 2044^T, *L. capsici* DSM 19286^T, *L. enzymogenes* 100 DSM 2043^T and *L. gummosus* DSM 6980^T) as representative of the biocontrol *Lysobacter* spp. (Hayward et al., 2010; Kobayashi and Yuen, 2007; Postma and Schilder, 2015; Postma et al., 2008)
and we assessed both volatilome composition and antagonistic effects against *Ph. infestans*, the causal
agent of late blight of potato and tomato plants (Fry, 2008). The VOCs produced by the type strains
on two growth media were analysed by solid-phase microextraction gas chromatography-mass
spectrometry (SPME/GC-MS) and proton transfer reaction-time of flight-mass spectrometry (PTRToF-MS) to precisely analyse the chemical composition and rapidly monitor the emission profiles,
respectively (Jordan et al., 2009b).

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110 **2. Materials and methods**

111 2.1. Propagation of the Lysobacter type strains and the plant pathogenic oomycete

The Lysobacter type strains L. antibioticus DSM 2044^T, L. capsici DSM 19286^T, L. enzymogenes 112 DSM 2043^T and L. gummosus DSM 6980^T were grown on Luria-Bertani Agar (LBA, Sigma-Aldrich, 113 St. Louis, MO, USA) for 72 h at 27 °C and cell suspensions of each strain were prepared by flooding 114 115 LBA dishes with 5 ml of sterile isotonic solution (0.85% NaCl). Bacterial cells were scraped from the 116 medium surface with a sterile spatula and collected in a sterile 15 ml-tube. The resulting cell suspensions were centrifuged $(4,300 \times g \text{ for } 15 \text{ min})$; pelleted cells were suspended in sterile isotonic 117 solution to a final optical density of 0.1 at 600 nm (OD₆₀₀), corresponding to 1×10^8 cells/ml (Puopolo 118 119 et al., 2016).

The *Ph. infestans* isolate (kindly provided by M. Finckh and A. Butz, University of Kassel,
Germany) was grown on pea agar medium (PAM, 12.5% frozen peas and 1.2% agar in distilled water)
at 17 °C, as described by Puopolo et al. (2015).

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- 126 2.2. Bacterial growth conditions for headspace analysis of volatile organic compounds

For headspace VOC analysis, 5 ml of sterilised nutrient agar (NA, Oxoid, Basingstoke, United 127 Kingdom) or potato dextrose agar (PDA, Oxoid) were poured into sterile 20 ml headspace vials (HS 128 129 vials, Supelco, Sigma-Aldrich) and they were left open under a laminar flow for 2 h at room temperature to avoid condensation. Each HS vial was then inoculated with 20 µl of the cell suspension 130 of a Lysobacter type strain $(1 \times 10^8 \text{ cells/ml})$ and left to dry under a laminar flow for 1 h at room 131 temperature. Each HS vial was tightly sealed with a sterilised 18 mm screw metal cap assembled with 132 133 silicone/PTFE septa of 1.3 mm (Supelco, Sigma-Aldrich). Additional HS vials containing non-134 inoculated NA or PDA (Uninoculated) were used as controls to exclude VOCs released from the 135 culture medium in the absence of bacteria (Kluger et al., 2013). HS vials were incubated at 25 °C for 136 ten days to accumulate VOCs before the headspace VOC assessment by SPME/GC-MS and PTR-ToF-MS analysis. This time point was selected because it showed the greatest antagonism of the 137 138 Lysobacter type strains against P. infestans.

The number of *Lysobacter* cells developed in each inoculated HS vial was assessed one day after headspace VOC analysis (11 days after inoculation). Each HS vial was flooded with 4 ml of sterilised isotonic solution (0.85% NaCl) and bacterial cells were scraped from the medium surface by vigorous vortexing for 30 sec. The cell concentration of the resulting suspension was assessed by converting the OD₆₀₀ values, with OD₆₀₀ = 0.1 corresponding to 1×10^8 cells/ml (Puopolo et al., 2016), and the quantity of *Lysobacter* cells was then calculated for each HS vial.

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146 2.3. Headspace analysis of volatile organic compounds using solid-phase micro extraction gas
147 chromatography-mass spectrometry (SPME/GC-MS) analysis

Headspace VOC analysis was carried out with SPME/GC-MS using an Auto System XL gas chromatograph coupled with a Turbo Mass Gold Mass spectrometer (Perkin Elmer, Norwalk, CT, USA). For measurement automatisation and standardisation, the instrument was coupled with a thermostated autosampler (CTC CombiPAL, CTC Analytics, Zwingen, Switzerland) and HS vials were kept at 25 °C. After equilibration for 30 min, VOCs were extracted and pre-concentrated with 153 solid phase microextraction (SPME) using 2 cm PDMS/DVB/CAR fibre (Supelco, Bellafonte, PA, USA), according to Endrizzi et al. (2012). The fibre collected VOCs from the headspace for 30 min 154 155 and desorbed them into the GC injector for 5 min at 250 °C. The chromatographic separation was 156 performed via an HP-Innowax fused-silica capillary column (length 30 m, inner diameter 0.32 mm, 157 film thickness 0.5 µm; Agilent Technologies, Palo Alto, CA, USA). The GC oven temperature 158 program was the following: 40 °C for 3 min, raised from 40 °C to 220 °C at 4 °C/min, 220 °C for 1 min, increased from 220 °C to 250 °C at 10 °C/min and 250 °C for 1 min. The carrier gas was helium 159 160 with a constant column flow rate of 1.5 ml/min. The transfer line temperature was maintained constant 161 at 220°C. Upon exiting the column, compounds were ionised via electron impact at 70 eV and 162 detected with a quadrupole mass spectrometer with a mass/charge ratio (m/z) ranging from 30 to 300 163 Thomson. Spectra analysis was carried out using TurboMass 5.4.0 software (Perkin Elmer, Norwalk, 164 CT; USA). Mass measure parameters were: background subtraction with a polynomial order of 1 and 165 a below curve of 33%, smooth mode with a peak width of 0.75 Da, minimum peak width at half 166 height of 4. Compound annotation was achieved by comparing the spectra with the NIST-98/Wiley 167 library (National Institute of Standards and Technology, www.nist.gov) using a mass spectrum 168 similarity greater than 85%, and by matching retention indices (RI) of authentic reference standards 169 computed under the same chromatographic conditions with the C7-C30 n-alkane series (Supelco, 170 Sigma-Aldrich) using a maximum tolerance of 4% RI deviation. The VOC content of each sample 171 was reported as the absolute peak area obtained with the TurboMass 5.4.0 software. Five replicates 172 (HS vials) of L. enzymogenes and four replicates of L. antibioticus, L. capsici and L. gummosus were 173 used for each media.

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175 2.4. Headspace analysis of volatile organic compounds using proton transfer reaction-time of flight176 mass spectrometry (PTR-ToF-MS)

177 Rapid headspace VOC analysis was carried out using a commercial PTR-TOF 8000 instrument
178 (Ionicon Analytik GmbH, Innsbruck, Austria) (Jordan et al., 2009a). The instrument was coupled

179 with an adapted thermostated autosampler (MPS Multipurpose Sampler, Gerstel) and HS vials were 180 kept at 25 °C. During VOC headspace measurement, 40 sccm of zero air were injected into the HS 181 vial through a needle heated to 40 °C, and the outflow going through a second heated needle (40 °C) 182 was delivered via Teflon fittings to the PTR-ToF-MS. Zero air was produced via a catalytic VOC 183 scrubber (GCU unit, Ionicon Analytik, Innsbruck, Austria). HS vials were measured in random order 184 and each measurement lasted for 3 min, with a waiting time of 5 min between samples to avoid memory effects. The PTR-ToF-MS was operated in H₃O⁺ primary ion mode. The following 185 186 conditions were set in the instrument drift tube: 2.3 mbar drift pressure, 480 V drift voltage, 110 °C 187 drift tube temperature, leading to an E/N value (E corresponding to electric field strength and N to gas number density) of about 120 Townsend (Td; 1 Td=10⁻¹⁷ Vcm²). The primary and product ions 188 189 exiting the drift tube region were detected using a time-of-flight (ToF) mass spectrometer operated 190 with its standard configuration (V mode). Each acquisition consisted of 350,000 channels with a 191 sampling time of 0.1 ns per channel of ToF acquisition, resulting in a mass spectrum ranging up to m/z = 400. Each individual spectrum was the sum of about 28,600 acquisitions lasting for 35 µs, 192 193 resulting in a time resolution of 1 s. Because the analysis time for each sample was set to 3 min, 180 194 spectra were acquired for each vial during each measurement.

195 PTR-ToF-MS spectra were processed according to the methodology reported by Cappellin et al. 196 (2011a), with slight modifications. As the first data processing step, signal distortions related to 197 detector dead time were calculated using a correction approach based on Poisson statistics, according 198 to Cappellin et al. (2011b). Because the external calibration provided by the acquisition software did 199 not achieve sufficient mass accuracy, internal mass calibration was carried out according to Cappellin 200 et al. (2011b) and a mass accuracy of greater than 0.001 Th was obtained. Subsequent data processing 201 of noise reduction, baseline removal and peak intensity extraction were carried out according to 202 Cappellin et al. (2011b) using modified Gaussians to fit spectral peaks. Headspace VOC 203 concentrations, expressed as parts per billion by volume (ppbv), were estimated from the integrated 204 signal over the 3 min of spectra acquisition using the formula described by Lindinger et al. (1998),

205 considering hydronium H_3O^+ as primary ion and a constant reaction rate coefficient of 2×10^{-9} cm³/s 206 in the calculations. This approach introduces a systematic deviation of up to 30% that can be 207 accounted for if the actual rate coefficient is known (Cappellin et al., 2012b). Four replicates (HS 208 vials) of *L. enzymogenes* and five of *L. antibioticus*, *L. capsici* and *L. gummosus* were grown on PDA. 209 Five replicates of *L. capsici* and four of *L. antibioticus*, *L. enzymogenes* and *L. gummosus* were grown 200 on NA.

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212 2.5. Functional analysis of bacterial volatile organic compounds against Phytophthora infestans 213 Split Petri dishes (92 mm of diameter) with two compartments and ventilation cams (Sarstedt, Nümbrecht, Germany) were used to analyse the effect of VOCs emitted by Lysobacter type strains 214 215 on Ph. infestans growth. Sterilised NA or PDA (5 mL) were poured into one half of the split dish 216 (Lysobacter-growth side) and 5 ml of sterilised PAM were poured into the other half (Phytophthoragrowth side). Once dried, 50 μ l of the cell suspension of the *Lysobacter* type strain (1 × 10⁸ cells/ml) 217 218 were spread onto the Lysobacter-growth side of the split dish containing NA or PDA using sterile 219 spatulas. As a control, split dishes containing non-inoculated NA or PDA (Uninoculated) on the 220 Lysobacter-growth side were used. Dishes were sealed with Parafilm tape (Beims, Neenah, WI, USA) 221 and incubated at 25 °C in the dark for 72 h. Subsequently, Ph. infestans plugs (5 mm) were cut from 222 the edge of ten-day-old colonies grown on PAM, as described by Puopolo et al. (2016), and a plug 223 was placed at the centre of the *Phytophthora*-growth side of each split dish. Inoculated dishes were 224 sealed with Parafilm tape and mycelial growth was evaluated by measuring the diameter (parallel to 225 the edge of the dish) of the Ph. infestans colony after seven days of incubation in the dark at 20 °C, 226 corresponding to ten days after Lysobacter spp. inoculation. Each Ph. infestans plug exposed to VOCs 227 of Lysobacter type strains grown on NA or PDA was then transferred to fresh PAM dishes and the colony diameter was measured after seven days of incubation in the dark at 20 °C. Seven replicates 228 229 (split dishes) were analysed for each Lysobacter type strain and each growth medium and the functional assay against Ph. infestans was carried out twice. 230

231 VOCs were selected according to their emission profiles; pure 2,5-dimethyl pyrazine, 2-232 methoxy-3-methyl pyrazine, decanal, delta-hexalactone, ethanol and pyrrole were purchased (Sigma-233 Aldrich) and tested against Ph. infestans. Sterilised PAM was poured into one half of a split dish 234 (Phytophthora-growth side) and a pure VOC was applied to a filter paper disk placed into the other 235 half (VOC side) at a concentration of and 190 mg/L (VOC-treated) of air volume, which is a dosage 236 compatible for VOC-mediated functional assays (De Vrieze et al., 2015; Fernando et al., 2005). As control, distilled water was applied to a filter paper disk into the VOC side of control dishes. Each 237 238 dish was sealed with Parafilm tape, incubated at 25 °C in the dark for 72 h and inoculated with a Ph. 239 infestans plug (5 mm) into the Phytophthora-growth side. The diameter (parallel to the edge of the dish) of each Ph. infestans colony was measured after seven days of incubation in the dark at 20 °C 240 241 and the inhibition of Ph. infestans growth (percentage) was calculated according to the following 242 formula: (growth in control dishes — growth in VOC-treated dishes) / (growth in control dishes) \times 243 100. Seven replicates (split dishes) were analysed for each treatment and the experiment was carried 244 out twice.

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246 2.6. Statistical analysis

To obtain background-corrected headspace VOC concentration, the background signal (the signal corresponding to the mean signal for empty HS vials) was subtracted from VOC emission values of both SPME/GC-MS and PTR-ToF-MS analysis. Emitted VOCs were identified as peaks with a background-corrected headspace concentration significantly greater than the corresponding signal for uninoculated HS vials for at least one strain and growth medium, according to the Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$).

Volatile emission data of SPME/GC-MS and PTR-ToF-MS analysis were analysed using inhouse routines written in R (www.r-project.org), including the Agricolae package (https://cran.rproject.org/web/packages/agricolae/index.html) and Glmnet package (https://cran.rproject.org/web/packages/glmnet/index.html). Data exploration with principal component analysis (PCA) was carried out using in-house routines written in R, on normalised variables that were
obtained by subtracting the mean and dividing by the standard deviation, to obtain more homogeneous
variables and prevent variance from being concentrated in few variables, affecting the results of PCA
(Afifi et al., 2011).

261 Bacterial class prediction models based on VOC data were developed with both SPME/GC-MS 262 and PTR-ToF-MS datasets, using the least absolute shrinkage and selection operator (LASSO) method described by Tibshirani (1996). Briefly, a linear model can be represented by the following 263 264 equation: $Y = X \times B + E$, where Y is the matrix of the properties to be predicted (dependent variable), X is the matrix of the measurements to be employed in the prediction (independent variable), B is the 265 266 matrix of regression coefficients to be estimated in the model optimisation procedure and E is the 267 matrix of residuals. In the LASSO method, the model can be represented by the following equation: $Y = X \times B + E + \lambda \times |B|$, which includes a penalisation term ($\lambda \times |B|$) of the absolute values of 268 269 coefficient B, multiplied by a factor λ , corresponding to the penalty coefficient to be optimised. 270 During model optimisation, the size of the penalty coefficient λ needs to be optimised and cross-271 validation is used for this purpose (Tibshirani, 1996). Models for all possible λ values were calculated 272 simultaneously as an ordinary linear regression, as reported by Hastie et al. (2009). The considered 273 study was a multiclass problem (i.e. growth media and strain classes) and a LASSO model for each 274 class was therefore developed to predict whether a sample belonged to the class (model value 1) or 275 not (model value 0). Performance evaluation of the classification methods was carried out using a 276 leave-one-out (LOO) procedure and confusion matrices (Cappellin et al., 2012a; Westerhuis et al., 2008). 277

VOC emission values, *Ph. infestans* colony diameters and *Lysobacter* cell numbers (log_{10} transformed) were analysed using Statistica 13.1 software (Dell, Round Rock, TX, USA) and a Kruskal-Wallis test with Bonferroni correction was applied to detect significant differences ($p \le 0.05$) among *Lysobacter* type strains and growth media.

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3. Results

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3.1. Profiles of volatile organic compounds differed according to the Lysobacter type strain and
growth media

VOC emission profiles measured by SPME/GC-MS analysis varied in Lysobacter type strains 288 289 and growth media, with the first and sixth principal components (explaining 36.7 and 3.8% of 290 variance respectively) of PCA analysis discriminating samples according to the growth medium (Fig. 291 1A). Moreover, marked similarities of VOC emission profiles occurred between HS vials belonging 292 to the same Lysobacter type strain, independently of the growth medium (Fig. 1B). Specifically, the 293 second and sixth principal components of PCA (explaining 18.6 and 3.8% of variance respectively) 294 discriminated Lysobacter type strains, and HS vials of the same type strain grown on the two media clustered together. 295

296 A total of 77 VOCs were detected by the SPME/GC-MS analysis, and the emission profiles of 297 70 of them differed according to the growth medium and type strain (Fig. 2 and Table S1). No 298 differences in bacterial growth were found between PDA and NA, except for L. antibioticus (Table 299 1), and VOC differences were mainly related to bacterial metabolism rather than to the growth rate. 300 Specifically, the emission of 17 VOCs was higher for all *Lysobacter* type strains on PDA as compared 301 with NA (PDA-specific VOCs, Cluster 1, Table S1), such as 3-methyl-2-buten-1-ol, 1-tridecanol, 1-302 tetradecanol, 1-pentadecanol, according to the Kruskal-Wallis test with Bonferroni correction ($p \leq$ 303 0.05). PDA-specific profiles were also found for three ketones [delta-hexalactone (Fig. 3A), dihydro-304 5-pentyl-2(3H)-furanone and 2-hexadecanone], an organosulfur compound (methyl thiolacetate), a 305 heterocyclic compound (indole; Fig. 3B) and five unknown compounds. Likewise, higher emission 306 of methyl 2-methyl butanoate, 4-methyl-1-pentanol and (Z)-3-decen-1-ol was measured for all 307 Lysobacter type strains grown on PDA as compared with NA. The emission profiles of PDA-specific 308 VOCs differed for Lysobacter type strains, and the emission of 1-tridecanol, 1-tetradecanol, 1pentadecanol, delta-hexalactone, 2-hexadecanone, methyl thiolacetate, five unknown compounds, 4methyl-1-pentanol and (Z)-3-decen-1-ol by *L. enzymogenes* was higher as compared with the other
type strains on PDA. Moreover, 3-methyl-2-buten-1-ol and indole were emitted mainly by *L. gummosus* and *L. capsici* respectively, and dihydro-5-pentyl-2(3H)-furanone was mainly emitted by
both *L. gummosus* and *L. capsici*.

314 The emission of 16 VOCs was higher for some Lysobacter type strains on PDA as compared with NA (Cluster 2, Table S1). For example, the emission of the n-undecanoic acid methyl ester was 315 316 specific for three type strains grown on PDA (L. capsici, L. enzymogenes and L. gummosus) and it 317 was not detected in other HS vials. These three type strains also showed higher emission of 1-318 propanol, 2-furanmethanol and 3-methyl-1-hexanol on PDA as compared with NA. The emission of 319 2-undecanol and methyl isobutyrate was higher on PDA as compared with NA for three type strains 320 (L. antibioticus, L. capsici and L. enzymogenes) and two type strains (L. antibioticus and L. 321 enzymogenes) respectively. On PDA, emission of 1-propanol by L. capsici and L. enzymogenes was 322 higher as compared with L. gummosus, while that of methyl isobutyrate by L. antibioticus and L. 323 enzymogenes was higher as compared with L. capsici and L. gummosus. Five compounds (2-324 tridecanone, acetone, 2-butanol and two unknown compounds) were emitted mainly by one PDA-325 grown Lysobacter type strain. Specifically, L. enzymogenes was characterised by the highest emission 326 of 2-tridecanone and two unknown compounds, while L. capsici was characterised by the most 327 significant emission of 2-butanol as compared with the other PDA-grown strains. Moreover, higher 328 emission of isoamyl alcohol and phenyl ethyl alcohol by L. enzymogenes, ethanol by L. capsici and 329 acetoin by L. antibioticus and L. capsici was found on PDA as compared with NA.

All the *Lysobacter* type strains showed higher emission of ten VOCs on NA as compared with PDA (NA-specific VOCs, Cluster 3, Table S1), such as 2,5-dimethyl pyrazine (Fig. 3C), 2,6-dimethyl pyrazine, 2,4,6-trimethyl pyridine, 1-(2-furanyl)-ethanone and an unknown compound, with consistent emission profiles for all NA-grown *Lysobacter* type strains. NA-specific profiles were also found for 2-butanone, pyrrole (Fig. 3D), 2-methoxy-3-(1-methyl-propyl) pyrazine and two unknown compounds. Emission of these VOCs differed for NA-grown *Lysobacter* type strains: 2-butanone and
pyrrole were mainly emitted by *L. capsici*, while 2-methoxy-3-(1-methyl-propyl) pyrazine and two
unknown compounds were mainly emitted by *L. enzymogenes*, as compared with the other NA-grown
type strains.

339 The emission of 15 VOCs was higher on NA as compared with PDA for some type strains 340 (Cluster 4, Table S1), as in the case of 1-butanol and 2-nonanone emission by L. antibioticus and L. 341 gummosus respectively. Dihydro-3-methyl-2(3H)-furanone, decanal (Fig. 3E) and an unknown 342 compound were specifically emitted by more than two Lysobacter type strains grown on NA, with 343 the highest emission by L. enzymogenes. Emission of 3-methoxy-2,5-dimethyl pyrazine by L. capsici 344 and L. gummosus was higher on NA as compared with PDA, while its emission by L. antibioticus and 345 L. enzymogenes was comparable on NA and PDA medium. Moreover, 2-methoxy-3-methyl pyrazine 346 (Fig. 3F) and 2-methoxy-6-methyl pyrazine were emitted exclusively by L. enzymogenes, with lower 347 and no emission by L. enzymogenes on PDA respectively. Likewise, four unknown compounds 348 (named from 13 to 16) were mainly emitted by NA-grown L. antibioticus and L. enzymogenes.

349 The emission profiles differed for Lysobacter type strains on both media for 14 VOCs (Cluster 350 5, Table S1). For example, the emission of methyl 2-methyl butanoate and methyl 3-methyl butanoate 351 was higher by L. antibioticus was higher as compared with the other type strains on both growth 352 media. On NA, the emission of 2-methyl-1-propanol by L. antibioticus and L. enzymogenes was 353 higher as compared with L. capsici and 3-octanol emission by L. gummosus was higher as compared 354 with L. antibioticus. Moreover, 2-methoxy-3-(1-methylethyl) pyrazine was mainly emitted by L. 355 enzymogenes on both media and to a lower extent by L. antibioticus on NA, while methyl isobutyl 356 ketone was mainly emitted by L. capsici as compared with L. antibioticus and L. gummosus on PDA. 357 Consistent emission of seven VOCs (2-pentanone, dimethyl disulfide, 2-heptanone, 6-methyl-2-358 heptanone, 5-methyl-2-heptanol, and two unknown compounds) was detected in *Lysobacter* type 359 strains and growth media (Cluster 6, Table S1) and they were possibly produced by constitutive 360 metabolic pathways.

362 3.2. Lysobacter type strains and their growth media can be recognised by modelling the profiles of
363 volatile organic compounds

364 Differences in VOC emission among Lysobacter type strains grown on the two growth media 365 were used to predict the medium on which the bacteria were grown. Optimisation of the LASSO 366 model corresponded with a linear combination of the original variables (VOCs annotated by SPME/GC-MS analysis) with a coefficient of zero, except for the coefficient associated with the 367 368 2,4,6-trimethyl pyridine variable (Table S1). This result highlighted that a simple univariate model, 369 built on the SPME/GC-MS emission data of 2,4,6-trimethyl pyridine alone, was sufficient to predict 370 the growth substrate on the basis of only one bacterial VOC. The prediction performance was assessed 371 with a LOO procedure and the success rate of growth media prediction using LASSO was 100%, 372 meaning that the growth media could be predicted with high level of accuracy (Table 2).

373 PTR-ToF-MS data (Table S2) confirmed the marked differences in VOC emissions by Lysobacter 374 type strains grown on PDA and NA. In agreement with the SPME/GC-MS analysis, the growth of 375 Lysobacter type strains was comparable on PDA and NA (except for L. antibioticus; Table 1). LASSO 376 modelling based on PTR-ToF-MS data resulted in prediction performance of the growth media 377 comparable to that obtained with SPME/GC-MS data and cross-validation using a LOO procedure 378 provided a prediction success rate of 100% (Table 2). The LASSO procedure was able to predict 379 growth media on the basis of only two bacterial VOCs associated with peaks at m/z of 68.050 and 380 129.091 (Table S2). Although annotation of the compounds associated with PTR-ToF-MS spectral peaks is difficult (Cappellin et al., 2011a), the peak at m/z of 68.050 corresponds with the C₄H₆N⁺ 381 382 ion, which is consistent with a fragment ion of 2,4,6-trimethyl pyridine reaction with H_3O^+ .

LASSO modelling was used for *Lysobacter* type strain prediction on the basis of VOC emission profiles. Since the medium prediction had a 100% success rate, the different growth media were modelled as separate classes and the success rate of strain prediction was 97 and 90% with LASSO modelling based on SPME/GC-MS and PTR-ToF-MS data respectively (Table 2). Specifically, HS

vials belonging to L. enzymogenes and L. antibioticus grown on both PDA and NA were correctly 387 388 classified by SPME/GC-MS analysis, as well as L. capsici and L. gummosus grown on PDA. HS vials 389 of L. gummosus were confused only once with L. capsici when grown on NA. Likewise, all 390 Lysobacter type strains grown on PDA were correctly classified according to their volatilome 391 assessed with PTR-ToF-MS analysis. On NA, HS vials of L. gummosus were confused only once 392 with L. antibioticus and those of L. enzymogenes were confused twice with L. capsici. The LASSO modelling based on SPME/GC-MS and PTR-ToF-MS data associated non-zero coefficients with 11 393 394 VOCs (Table S1) and 12 peaks (Table S2) to distinguish Lysobacter type strains respectively, 395 indicating that a linear model built using only these compounds was sufficient to discriminate the 396 bacterial strains tested. Specifically, the emission profiles of methyl 2-methyl butanoate and methyl 397 3-methyl butanoate were characteristic for L. antibioticus on PDA, while those of 2-butanol, methyl 398 thiolacetate and 3-methyl-2-buten-1-ol specified the emission of L. capsici, L. enzymogenes and L. 399 gummosus respectively. On NA, the volatilome of L. antibioticus was characterised by the emission 400 profiles of 2-furanmethanol and unknown compound 14, and L. capsici, L. enzymogenes and L. 401 gummosus were specified by acetone, pyrazine (2-methoxy-3-methyl pyrazine and 2-methoxy-6-402 methyl pyrazine) and 3-octanol emission respectively.

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406 3.3. Volatile organic compounds emitted by Lysobacter type strains grown on nutrient agar and not
407 on potato dextrose agar inhibit Phytophthora infestans growth

408 VOCs emitted by *L. antibioticus*, *L. capsici*, *L. enzymogenes* and *L. gummosus* grown on NA 409 inhibited the mycelial growth of *Ph. infestans* (Fig. 4A and 4B). Conversely, VOCs produced by the 410 four *Lysobacter* type strains grown on PDA did not affect *Ph. infestans* growth. When transferred to 411 new PAM dishes, the growth of *Ph. infestans* plugs previously exposed to VOCs produced by NA-412 grown *Lysobacter* type strains was compromised as compared with *Ph. infestans* plugs exposed to 413 uninoculated NA (Fig. 4C). Moreover, the growth of plugs exposed to VOCs produced by PDA-414 grown *Lysobacter* type strains was comparable (Kruskal-Wallis test, p > 0.05) with those exposed to 415 uninoculated NA and PDA (data not shown).

Functional assays demonstrated that pure 2,5-dimethyl pyrazine, 2-methoxy-3-methyl pyrazine, decanal and pyrrole inhibited the *Ph. infestans* growth (Table 3) and they were mainly emitted by the NA-grown *Lysobacter* type strains. Conversely, pure delta-hexalactone and ethanol, that were mainly emitted by the PDA-grown *Lysobacter* type strains, did not significantly inhibit the *Ph. infestans* growth.

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423 **4. Discussion**

424 In the last decade, increasing attention has been paid to the functional roles of bacterial VOCs in 425 soil microbial interactions (Effmert et al., 2012; Kai et al., 2009). Strains belonging to L. antibioticus, 426 L. capsici and L. gummosus play a major role in the soil suppressiveness against R. solani (Postma 427 and Schilder, 2015; Postma et al., 2008) and strains belonging to L. enzymogenes are involved in the 428 biocontrol of several phytopathogens (Hayward et al., 2010; Kobayashi and Yuen, 2007). The 429 incidence of Lysobacter spp. in soil is influenced by soil type, plant cover, seasonal factors and 430 organic amendments (Hayward et al., 2010; Postma et al., 2008), but little is known about the 431 ecological role of VOCs emitted by strains belonging to this genus. Although the synthesis of non-432 volatile antimicrobial metabolites against phytopathogens has been widely studied (Puopolo et al., 433 2014a; Xie et al., 2012), the possible contribution of Lysobacter VOCs to antagonistic processes has 434 been poorly investigated (Sang et al., 2011; Zou et al., 2007). In this work we analysed the VOC 435 profiles emitted by four Lysobacter type strains grown on PDA and NA using two types of headspace 436 analysis. The growth media had a different nutrient composition: PDA (sugar-rich media) contained 437 a high sugar content (20 g/l glucose), while NA (protein-rich media) mainly contained proteins (5 g/l 438 peptone and 2 g/l yeast extract) with a low sugar content (Fiddaman and Rossall, 1994). VOC 439 emission by the Lysobacter type strains changed radically according to the growth medium, and the 440 comparable growth rate on both media indicated that volatilome differences were mainly related to 441 metabolic changes instead of biomass formation. The composition and functional properties of the 442 bacterial volatilome are known to be influenced by the growth substrate (Asari et al., 2016; Blom et 443 al., 2011; Bruce et al., 2003; Garbeva et al., 2014; Weise et al., 2012), indicating metabolic adaptation 444 of VOC production according to nutrient availability in the soil (Insam and Seewald, 2010). However, 445 the composition of the bacterial bouquet is defined by genetic determinants (Peñuelas et al., 2014) 446 and can be used to identify the four Lysobacter type strains grown on PDA with LASSO modelling. 447 Specifically, butanoates (methyl 2-methyl butanoate, methyl 3-methyl butanoate), 2-butanol, methyl 448 thiolacetate and 3-methyl-2-buten-1-ol specified the emission of L. antibioticus, L. capsici, L. 449 enzymogenes and L. gummosus on PDA respectively. On NA, 2-furanmethanol and two pyrazines (2-450 methoxy-3-methyl pyrazine and 2-methoxy-6-methyl pyrazine) discriminated the emission of L. 451 antibioticus and L. enzymogenes respectively. Interestingly, Lysobacter type strains emitted some 452 strain-specific VOCs independently of the growth media. For example, L. antibioticus emitted higher 453 amounts of two methyl esters (methyl 2-methyl butanoate and methyl 3-methyl butanoate) as 454 compared with the other type on both media.

455 The functional properties of the Lysobacter volatilome changed radically according to the growth 456 medium and the VOC-mediated biocontrol effects of the four Lysobacter type strains against Ph. 457 infestans were enhanced on the protein-rich medium as compared with the sugar-rich medium. 458 Likewise, a protein-rich medium (tryptone soya agar) increased the VOC-mediated antagonism of Serratia spp. against five sapstain fungi as compared with a sugar-rich medium (malt extract agar) 459 460 (Bruce et al., 2003). In our experiments, higher emission of six pyrazines [2,5-dimethyl pyrazine, 2,6-461 dymethyl pyrazine, 2-methoxy-3-(1-methyl-propyl) pyrazine, 3-methoxy-2,5-dimethyl pyrazine, 2methoxy-3-methyl pyrazine and 2-methoxy-6-methyl pyrazine] was found by NA-grown as 462 463 compared with PDA-grown Lysobacter type strains and two pure pyrazines (2,5-dimethyl pyrazine, 464 2-methoxy-3-methyl pyrazine) inhibited the Ph. infestans growth. Pyrazines are synthesised by

alanine, valine, leucine and isoleucine (Dickschat et al., 2005) and addition of amino acids to the 465 466 growth medium increased the bacterial production of pyrazines (Beck et al., 2003; Bungert et al., 467 2001), in agreement with higher emission by the NA-grown as compared with the PDA-grown Lysobacter type strains. Pyrazines and related heterocyclic compounds were involved in 468 469 antimicrobial activities (Baldwin et al., 2013; Beck et al., 2003). Specifically, 2,5-dimethyl pyrazine 470 and 2-ethyl-3,5-dimethyl pyrazine were emitted by a biocontrol strain of Bacillus pumilus and pure 2,5-dimethyl pyrazine showed antagonistic activity against Ph. infestans (De Vrieze et al., 2015) and 471 472 Phaeomoniella chlamydospora (Haidar et al., 2016). Likewise, the VOC-mediated antimicrobial 473 activity of the antagonist B. megaterium BP17 (Munjal et al., 2016) and Pseudomonas putida BP25 474 (Sheoran et al., 2015) was attributed to the emission of pyrazines, and four pure molecules (2,5-475 dimethyl pyrazine, 2-ethyl-3-methyl pyrazine, 2-ethyl pyrazine and 2-methyl pyrazine) showed 476 inhibitory activities against *Ph. capsici*, *Ralstonia solanacearum* and *Magnaporthe oryzae* (Munjal 477 et al., 2016). As shown for pyrazine emission by Paenibacillus spp. (Rybakova et al., 2016), three 478 pyrazines [2-methoxy-3-(1-methyl-propyl) pyrazine, 2-methoxy-3-methyl pyrazine and 2-methoxy-479 6-methyl pyrazine] were species-specifically emitted, indicating the involvement of some species-480 specific biosynthetic pathways.

481 All the Lysobacter type strains showed higher emission of 2,4,6-trimethyl pyridine when grown 482 on NA as compared to PDA and this VOC was also produced by the antagonistic strains Collimonas 483 fungivorans Ter331 and C. pratensis Ter91 (Garbeva et al., 2014), suggesting its contribution to 484 bacterial biocontrol processes. Pyrrole was also emitted by NA-grown Lysobacter type strains and 485 inhibited the Ph. infestans growth. Likewise, pyrrole derivatives exhibited antagonistic activity 486 towards phytopathogens, such as pyrrolnitrin [3-chloro-4-(20-nitro-30-chlorophenyl) pyrrole], which 487 is a broad spectrum antifungal metabolite produced by several bacterial species (Saraf et al., 2014). 488 The emission of 1-butanol by L. antibioticus was higher on NA as compared with PDA and the pure 489 compound inhibited mycelial growth of F. oxysporum and Moniliophthora perniciosa (Chaves-López 490 et al., 2015). Likewise, 1-butanol derivatives (3-methyl-1-butanol, 2-methyl-1-butanol and 1-butanol, 491 3-methyl-acetate) inhibited the mycelial growth of Py. ultimum, R. solani and S. sclerotiorum (Fialho 492 et al., 2011; Strobel et al., 2001), suggesting their contribution to the antagonistic activity against 493 phytopathogens. Dihydro-3-methyl-2(3H)-furanone and decanal were mainly emitted by the NA-494 grown L. enzymogenes and the last one inhibited the Ph. infestans growth, indicating their potential 495 biocontrol activities as already shown for 3-(1-Hexenyl)-5-methyl-2-(5H)-furanone (Paulitz et al., 496 2000) and decanal (Fernando et al., 2005). Nine VOCs specifically emitted by the NA-grown 497 Lysobacter type strains were found as unknown substances and more sensitive chromatographic 498 techniques are required to better characterise these compounds.

499 A blend of 17 VOCs was specifically emitted by all the Lysobacter type strains grown on PDA 500 and not on NA, such as three aliphatic alcohols (1-tridecanol, 1-tetradecanol and 1-pentadecanol). In 501 particular, 1-tridecanol and 1-tetradecanol showed no antifungal activity against Saccharomyces 502 cerevisiae (Kubo et al., 2003) and S. sclerotiorum (Giorgio et al., 2015) respectively, in agreement 503 with the absence of Ph. infestans inhibition with VOCs emitted by the PDA-grown Lysobacter type 504 strains. Delta-hexalactone and ethanol were mainly emitted by the PDA-grown Lysobacter type 505 strains and they did not show antagonistic activity against P. infestans. Likewise, ethanol was 506 produced by two Serratia spp. strains grown on the sugar-rich medium (malt extract agar) and not on 507 the protein-rich medium (tryptone soya agar) (Bruce et al., 2004) and did not seem to be implicated 508 in VOC-mediated biocontrol processes (Bruce et al., 2003). Indole emission was detected from the 509 PDA-grown Lysobacter type strains and low dosages of this VOC promoted Arabidopsis thaliana 510 growth (Blom et al., 2011). Likewise, acetoin, responsible for stimulation of A. thaliana growth (Ryu 511 et al., 2003), was mainly emitted by L. antibioticus and L. capsici on PDA, in agreement with higher 512 emission by B. amyloliquefaciens grown on a sugar-rich medium (M9 agar supplemented with 513 glucose) as compared with protein-rich media (tryptic soy agar and Luria-Bertani agar) (Asari et al., 514 2016).

In conclusion, specific VOCs of *Lysobacter* spp. were identified and a tool for recognising four
 Lysobacter type strains *in vitro* was developed according to VOC emission profiles, assessed using

517 SPME/GC-MS or PTR-ToF-MS analysis. The chemical profiles and functional properties of 518 Lysobacter VOCs differed according to the growth medium, suggesting that appropriate nutrient 519 sources should be preferred in dual culture assays in order to maximise biocontrol efficacy against 520 phytopathogens. Bacterial VOC production in soil can differ according to community composition 521 and nutrient availability (Insam and Seewald, 2010), suggesting a possible adaptation to the soil 522 environment and inhabitants. Although our results were obtained from in vitro-grown bacteria, we hypothesized a possible scenario of Lysobacter VOCs that need further validation under soil 523 524 conditions. Particularly, protein sources deriving from the lytic activities of phytopathogenic or saprophytic fungi may stimulate the production of antimicrobial VOCs by Lysobacter type strains 525 526 (volatile pyrazines, pyrrole and decanal) to maximise antagonism to soil microbial inhabitants, such 527 as Ph. infestans. Conversely, an increase in sugar availability due to root exudates in the rhizosphere (Jones et al., 2004) may change the volatilome of Lysobacter type strains, possibly to increase the 528 529 production of plant growth stimulators (acetoin and indole) and non-antimicrobial compounds (1-530 tridecanol, 1-tetradecanol, delta-hexalactone and ethanol) to maximise beneficial interaction with the 531 plant. However, further studies are required to investigate the volatilome shift and properties of 532 Lysobacter spp. in soil conditions.

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Table 1. Number of *Lysobacter* type strain cells developed during volatile organic compound assessment.

| VOC analysis ¹ | Media ² | Lysobacter type strain con | ncentration ³ | | |
|---------------------------|--------------------|---|---|-----------------------------------|----------------------------------|
| | | L. antibioticus | L. capsici | L. enzymogenes | L. gummosus |
| SDME/CC MS | PDA | $7.75 \pm 0.32 \times 10^9$ ab | $4.77 \pm 0.38 \times 10^9 \mathrm{bc}$ | $7.23 \pm 0.49 \times 10^9$ ab | $1.12 \pm 0.03 \times 10^{10}$ a |
| SPINE/OC-INS | NA | $7.72 \pm 0.42 \times 10^8 \text{ c}$ | $5.82 \pm 1.30 \times 10^9$ abc | $5.86 \pm 0.92 \times 10^9$ abc | $4.14 \pm 2.09 \times 10^9$ abc |
| DTD T ₂ E MS | PDA | $6.80 \pm 0.30 \times 10^9$ ab | $4.87 \pm 0.66 \times 10^9 \mathrm{bc}$ | $6.79 \pm 0.18 \times 10^9$ ab | $1.06 \pm 0.08 \times 10^{10}$ a |
| Г I К-10Г-IVIS | NA | $1.40 \pm 0.09 \times 10^9 \mathrm{~c}$ | $5.02 \pm 0.74 \times 10^{9}$ abc | $5.09 \pm 0.24 \times 10^{9}$ abc | $4.01 \pm 1.45 \times 10^9$ abc |

¹ Headspace analysis of the volatile organic compounds (VOCs) emitted by *Lysobacter antibioticus* DSM 2044^T, *L. capsici* DSM 19286^T, *L. enzymogenes* DSM 2043^T and *L. gummosus* DSM 6980^T was carried out using solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) and proton transfer reaction-time of flight-mass spectrometry (PTR-ToF-MS).

² The *Lysobacter* type strains were grown for ten days at 25°C on potato dextrose agar (PDA) or nutrient agar (NA) in headspace vials before VOC analysis.

³ Growth of the *Lysobacter* type strains was measured one day after VOC assessment (11 days after inoculation). Bacterial cells were collected from the headspace vials and the number of *Lysobacter* cells for each vial was calculated by measuring optical density at 600 nm [optical density of 0.1 corresponds to 1×10^8 cells/ml according to Puopolo et al. (2016)]. Mean and standard deviation values of *Lysobacter* cells from four to five replicates are presented for each type strain and growth media. For each headspace VOC analysis, different letters indicate significant differences according to a Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$).

Table 2. Confusion matrix for bacteria strain prediction based on the least absolute shrinkage and selection operator method (LASSO) with a leaveone-out (LOO) procedure based on solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) and proton transfer reactiontime of flight-mass spectrometry (PTR-ToF-MS) data.

| | | | LASSO predicted class | | | | | | | | |
|-----|------|-----------------|-----------------------|------------|----------------|-------------|-----------------|------------|----------------|-------------|--|
| | | | | | PDA | | | | NA | | |
| | | | L. antibioticus | L. capsici | L. enzymogenes | L. gummosus | L. antibioticus | L. capsici | L. enzymogenes | L. gummosus | |
| | SPM | E/GC-MS data | | | | | | | | | |
| | | L. antibioticus | 4 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | |
| | | L. capsici | 0 | 4 | 0 | 0 | 0 | 0 | 0 | 0 | |
| | FDA | L. enzymogenes | 0 | 0 | 5 | 0 | 0 | 0 | 0 | 0 | |
| | | L. gummosus | 0 | 0 | 0 | 4 | 0 | 0 | 0 | 0 | |
| | NA | L. antibioticus | 0 | 0 | 0 | 0 | 4 | 0 | 0 | 0 | |
| | | L. capsici | 0 | 0 | 0 | 0 | 0 | 4 | 0 | 0 | |
| | | L. enzymogenes | 0 | 0 | 0 | 0 | 0 | 0 | 5 | 0 | |
| ass | | L. gummosus | 0 | 0 | 0 | 0 | 0 | 1 | 0 | 3 | |
| alc | PTR- | ToF-MS data | | | | | | | | | |
| Reź | | L. antibioticus | 5 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | |
| | | L. capsici | 0 | 5 | 0 | 0 | 0 | 0 | 0 | 0 | |
| | PDA | L. enzymogenes | 0 | 0 | 4 | 0 | 0 | 0 | 0 | 0 | |
| | | L. gummosus | 0 | 0 | 0 | 5 | 0 | 0 | 0 | 0 | |
| | | L. antibioticus | 0 | 0 | 0 | 0 | 4 | 0 | 0 | 0 | |
| | NT A | L. capsici | 0 | 0 | 0 | 0 | 0 | 4 | 0 | 0 | |
| | INA | L. enzymogenes | 0 | 0 | 0 | 0 | 0 | 2 | 2 | 0 | |
| | | L. gummosus | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 4 | |

SPME/GC-MS and PTR-ToF-MS data were obtained from Lysobacter antibioticus DSM 2044^T, L. capsici DSM 19286^T, L. enzymogenes DSM 2043^T

and *L. gummosus* DSM 6980^T grown on potato dextrose agar (PDA) or nutrient agar (NA).

Columns represent the class predicted with the LASSO method based on SPME/GC-MS or PTR-ToF-MS data and rows represent the real class. Diagonal entries of the matrix correspond to the number of samples correctly classified for each class and off-diagonal entries correspond to prediction errors.

Table 3. Antagonistic activity of pure volatile organic compounds (VOCs) on *Phytophthora infestans* growth.

| Treatment | Inhibition (%) of Phytophthora infestans growth |
|-----------------------------|---|
| control | 0.00 ± 4.86 a |
| 2,5-dimethyl pyrazine | 93.66 ± 1.44 bc |
| 2-methoxy-3-methyl pyrazine | $97.03 \pm 1.08 \text{ c}$ |
| decanal | 94.52 ± 0.82 bc |
| delta-hexalactone | 8.55 ± 3.59 a |
| ethanol | 29.33 ± 6.99 ab |
| pyrrole | $96.58\pm0.96~\mathrm{c}$ |

Each pure VOC was applied to a filter paper disk in split dishes at a concentration of 190 mg/l of air volume (VOC-treated) and distilled water was applied in control dishes (control). One plug of *Ph. infestans* was placed on the other side of each dish containing the pea agar medium (PAM) and colony diameter was measured seven days after incubation at 20°C. The inhibition of *Ph. infestans* growth (percentage) was calculated according to the following formula: (growth in control dishes — growth in VOC-treated dishes) / (growth in control dishes) × 100. Seven replicates (split dishes) were analysed for each treatment and the experiment was carried out twice. Mean and standard error values of 14 replicates pooled from two experiments are presented for each treatment. Different letters indicate significant differences according to the Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$).

Figure legends

Fig. 1. Principal component analysis (PCA) of volatile organic compounds (VOCs) emitted by *Lysobacter* type strains. PCA was based on VOCs measured using solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) for *Lysobacter antibioticus* DSM 2044^T (red), *L. capsici* DSM 19286^T (blue), *L. enzymogenes* DSM 2043^T (black) and *L. gummosus* DSM 6980^T (green) grown for days on nutrient agar (triangles) or potato dextrose agar (circles). The percentage of variance explained by the principal components (PC) is reported in brackets for PC1 and PC6 (A) or PC6 and PC2 (B).

Fig. 2. Profiles of volatile organic compounds (VOCs) emitted by *Lysobacter* type strains. Headspace VOC analysis was carried out using solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) for *Lysobacter antibioticus* DSM 2044^T (La), *L. capsici* DSM 19286^T (Lc), *L. enzymogenes* DSM 2043^T (Le) and *L. gummosus* DSM 6980^T (Lg) grown for ten days on nutrient agar (NA) or potato dextrose agar (PDA). For each compound, the intensity of the colour gradient and letters are based on a Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$) on VOC emission data (Table S1). Compounds were grouped based on their emission profiles into: VOCs with higher emission by all *Lysobacter* type strains on PDA as compared with NA (Cluster 1), VOCs with higher emission by all *Lysobacter* type strains on NA as compared with PDA (Cluster 3), VOCs with higher emission by some *Lysobacter* type strains on NA as compared with PDA (Cluster 4), VOCs with different (Cluster 5) or consistent (Cluster 6) emission by *Lysobacter* type strains on both growth media.

Fig. 3. Profiles of selected volatile organic compounds (VOCs) emitted by *Lysobacter* type strains. Emission of delta-hexalactone (**A**), indole (**B**), 2,5-dimethyl pyrazine (**C**), pyrrole (**D**), decanal (**E**) and 2-methoxy-3-methyl pyrazine (F) was measured using solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) for *L. antibioticus* DSM 2044^T, *L. capsici* DSM 19286^T, *L. enzymogenes* DSM 2043^T and *L. gummosus* DSM 6980^T grown for ten days on nutrient agar (NA) or potato dextrose agar (PDA). For each compound, mean and standard error values of the absolute peak area from four to five replicates are reported for each *Lysobacter* type strain and growth medium. Different letters indicate significant differences according to the Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$). The structural formula is reported for each compound.

Fig. 4. Antagonistic activity of volatile organic compounds (VOCs) produced by *Lysobacter* type strains against *Phytophthora infestans*. *Lysobacter antibioticus* DSM 2044^T, *L. capsici* DSM 19286^T, *L. enzymogenes* DSM 2043^T and *L. gummosus* DSM 6980^T were grown for 72 h at 25°C on potato dextrose agar (PDA) or nutrient agar (NA) in split dishes and uninoculated dishes were used as controls (Uninoculated). One plug of *Ph. infestans* was placed on the other side of each dish containing the pea agar medium (PAM) and colony diameter was measured seven days after incubation at 20°C (**A**). Representative pictures of *Ph. infestans* growth (lower side of each dish) in split dishes with NA and PDA growth medium inoculated (*L. antibioticus*) or not (Uninoculated) with *L. antibioticus* (upper side of each dish) (**B**). Each *Ph. infestans* plug exposed to VOCs emitted by *Lysobacter* type strains grown on NA was then transferred to a fresh PAM dish and colony diameter was measured seven days after incubation at 20°C (**C**). Seven replicates (dishes) were analysed for each treatment and the experiment was carried out twice. Mean and standard error values of mycelium diameters obtained from 14 replicates pooled from two experiments are presented for each bacterial strain and growth media. Different letters indicate significant differences according to the Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$).

Appendix A. Supplementary data

Table S1. Volatile organic compounds (VOCs) emitted by the *Lysobacter* type strains and measured using solid-phase micro extraction gas chromatography-mass spectrometry (SPME/GC-MS) analysis.

Table S2. Volatile organic compounds (VOCs) emitted by *Lysobacter* type strains and measured using proton transfer reaction-time of flight-mass spectrometry (PTR-ToF-MS) analysis.



| | | | P | A | | | N | A | |
|------|---------------------------------------|----|----|----|----|----|----|----|----|
| | | La | Lc | Le | Lg | La | Lc | Le | Lg |
| | 3-methyl-2-buten-1-ol | b | b | b | а | | | | |
| | 1-tridecanol | b | b | а | b | | | | |
| | 1-tetradecanol | b | b | а | b | | | | |
| | 1-pentadecanol | b | b | а | b | | | | |
| | unknown alcohol 1 | b | b | а | b | | | | |
| | unknown alcohol 2 | b | b | а | b | | | | |
| | unknown alcohol 3 | b | b | а | b | | | | |
| 5 | unknown alcohol 4 | b | b | а | b | | | | |
| ute | delta-hexalactone | b | b | а | b | | | | |
| S | dihydro-5-pentyl-2(3H)-furanone | b | а | b | а | | | | |
| 0 | 2-hexadecanone | b | b | а | b | | | | |
| | methyl thiolacetate | b | b | а | b | | | | |
| | indole | b | а | b | b | | | | |
| | unknown compound 1 | b | b | а | b | | | | |
| | methyl 2-methylbutyrate | ab | bc | а | bc | cd | d | d | d |
| | 4-methyl-1-pentanol | b | bc | а | bc | С | | | |
| | (Z)-3-Decen-1-ol | b | b | а | b | | | b | |
| | n-undecanoic acid methyl ester | | b | а | b | | | | |
| | 1-propanol | de | а | b | cd | cd | | С | е |
| | 2-furanmethanol | а | а | а | а | а | | b | b |
| | 3-methyl-1-hexanol | ab | b | а | b | b | | | |
| | 2-undecanol | а | а | а | а | | | | а |
| | methyl isobutyrate | а | b | а | b | b | b | b | b |
| 2 | 2-tridecanone | b | b | а | b | b | b | b | b |
| ter | acetone | b | а | а | b | b | а | b | b |
| Ins | unknown compound 2 | b | b | а | | b | | b | b |
| C | unknown compound 3 | b | b | а | b | b | b | b | b |
| | 2-butanol | b | а | b | b | b | b | b | b |
| | isoamyl alcohol | ab | а | а | а | a | ab | b | ab |
| | phenyl ethyl alcohol | ab | ab | а | ab | ab | ab | b | ab |
| | ethanol | ab | а | ab | ab | ab | b | b | ab |
| | acetoin | а | ab | bc | bc | bc | С | С | С |
| | unknown compound 4 | | | а | b | | | | |
| | 2,5-dimethyl pyrazine | b | b | b | b | a | а | а | а |
| | 2,6-dimethyl pyrazine | b | b | b | b | a | а | а | а |
| | 2,4,6-trimethyl pyridine | b | b | b | b | a | а | а | а |
| ŝ | 1-(2-furanyl)-ethanone | cd | d | bc | d | ab | ab | а | ab |
| ster | unknown compound 5 | | | | | а | а | а | а |
| Ins | 2-butanone | d | d | d | d | С | а | С | b |
| C | pyrrole | d | d | d | d | С | а | b | b |
| - | 2-methoxy-3-(1-methyl-propyl)pyrazine | d | | С | | b | d | а | d |
| | unknown compound 6 | С | | С | | b | С | а | С |
| | unknown compound 7 | | | b | | ab | b | а | ab |

| | | | P | DA | | NA | | | | | |
|------|--------------------------------------|----|----|----|----|----|----|----|----|----|--|
| | | La | Lc | Le | Lg | | La | Lc | Le | Lg | |
| | 1-butanol | b | b | b | b | | а | b | b | b | |
| | 2-nonanone | а | а | а | | | а | а | а | а | |
| | dihydro-3-methyl-2(3H)-furanone | | | | | | | b | а | b | |
| | decanal | | | | | | b | С | а | | |
| _ | unknown compound 8 | | | С | | | b | | а | С | |
| 91 4 | 3-methoxy-2,5-dimethyl pyrazine | а | | а | | | а | а | а | а | |
| ste | 2-methoxy-3-methyl pyrazine | | | b | | | | | а | | |
| 3 | 2-methoxy-6-methyl pyrazine | | | | | | | | а | | |
| 0 | unknown compound 9 | | | | | | b | | а | | |
| | unknown compound 10 | | | | | | а | | b | | |
| | unknown compound 11 | | | С | | | b | | а | | |
| | unknown compound 12 | | | | | | а | | b | | |
| | unknown compound 13 | b | b | b | b | | b | b | а | b | |
| | methyl 2-methyl butanoate | ab | С | bc | С | | а | | С | | |
| | methyl 3-methylbutanoate | а | ab | ab | b | | а | b | ab | b | |
| | 2-methyl-1-propanol | ab | ab | ab | ab | | а | b | а | ab | |
| | 3-octanol | ab | ab | ab | ab | | b | ab | ab | а | |
| | 2-methoxy-3-(1-methylethyl) pyrazine | | | ab | | | b | | а | | |
| ŝ | methyl isobutyl ketone | b | а | ab | b | | b | ab | ab | ab | |
| ter | unknown ketone 1 | b | ab | а | b | | ab | ab | ab | ab | |
| sn | unknown compound 14 | | b | b | ab | | b | b | b | а | |
| Ω | unknown compound 15 | | | ab | b | | | b | а | b | |
| | unknown ketone 2 | | | а | | | | | | а | |
| | unknown compound 16 | | b | а | | | | | | b | |
| | 6-methyl-2-heptanol | а | а | а | а | | а | а | | а | |
| | 2-nonanol | | а | а | а | | а | а | а | а | |
| | 2-decanol | а | а | а | | | а | а | а | а | |
| | 2-pentanone | а | а | а | а | | а | а | а | а | |
| ~ | dimethyl disulfide | а | а | а | а | | а | а | а | а | |
| er 6 | 2-heptanone | а | а | а | а | | а | а | а | а | |
| Ť | 6-methyl-2-heptanone | а | а | а | а | | а | а | а | а | |
| S | 5-methyl-2-heptanol | а | а | а | а | | а | а | а | а | |
| 5 | unknown ketone 3 | а | а | а | а | | а | а | а | а | |
| | unknown alcohol 5 | а | а | а | а | | а | а | а | а | |
| | | | | | | | | | | | |











Supplementary Table S1. Volatile organic compounds (VOCs) emitted by the Lysobacter type strains and measured using solid-phase micro extraction gas chromatography-mass spectrometry (SPME/GC-MS) analysis.

| | Compound | CAS | Retention | Retention | etention NIST Potato dextrose agar | | | | Nutrie | nt agar | | LASSO | | |
|--------|---------------------------------------|------------|-----------|-----------|------------------------------------|--|--|--|---|--|---|---|--|---|
| | | number | time | index | retention | Insohaatar | Insohaatar | Iwahaatar | Insohaatar | Insohaatar | Insohaatar | Insohaatar | Insohaatar | |
| | | | | | muex | antihioticus | Lysobacier cansici | Lysobucier enzymogenes | Lysobucier gummosus | antibioticus | Lysobucier cansici | Lysobucier enzymogenes | Lysobucier gummosus | |
| | 3-methyl-2-buten-1-ol | 556-82-1 | 14.5 | 1351 | 1320 | $3.1 \pm 1.7 \times 10^4 \text{ b}$ | $1.6 \pm 1.5 \times 10^4 \text{ b}$ | $8.1 \pm 2.2 \times 10^4 \text{ b}$ | $2.4 \pm 0.7 \times 10^5 \text{ a }^{*}$ | - | - | - | - | |
| | 1-tridecanol | 112-70-9 | 30.92 | 2056 | 2074 | $1.9 \pm 1.4 \times 10^{5} \mathrm{b}$ | $1.7 \pm 0.7 \times 10^{5} \text{ b}$ | $1.5 \pm 0.5 \times 10^{6}$ a | $4.3 \pm 3.1 \times 10^5 \text{ b}$ | - | - | - | - | |
| | 1-tetradecanol | 112-72-1 | 33.83 | 2161 | 2165 | $5.6 \pm 3.3 \times 10^4 \mathrm{b}$ | $5.6\pm2.3	imes10^4~{ m b}$ | $4.1 \pm 1.6 \times 10^5$ a | $8.5\pm5.0	imes10^4~{ m b}$ | - | - | - | - | |
| | 1-pentadecanol | 629-76-5 | 37.54 | 2297 | 2272 | $7.4 \pm 0.1 \times 10^{3} \mathrm{b}$ | $1.4\pm2.7	imes10^4~{ m b}$ | $1.9 \pm 0.4 \times 10^{5}$ a | $5.0\pm5.7\times10^4b$ | - | - | - | - | |
| | delta-hexalactone | 823-22-3 | 26.12 | 1822 | 1791 | $7.4 \pm 1.7 \times 10^4 \text{ b}$ | $5.3\pm0.8\times10^{4}b$ | $1.4\pm0.2	imes10^{5}$ a | $4.6\pm1.6\times10^4b$ | - | - | - | - | |
| | dihydro-5-pentyl-2(3H)- | 104-61-0 | 31.08 | 2064 | 2024 | $1.6\pm0.5\times10^{5}b$ | $3.0\pm0.3	imes10^5~\mathrm{a}$ | $1.4\pm0.4\times10^{5}~b$ | $2.9\pm0.7	imes10^{5}$ a | - | - | - | - | |
| | furanone | | | | | | | | | | | | | |
| - | 2-hexadecanone | 18787-63-8 | 31.97 | 2106 | 2121 | $5.7\pm3.7	imes10^4b$ | $4.6\pm1.5\times10^4b$ | $2.6\pm0.6	imes10^5$ a | $2.9\pm3.4\times10^{4}b$ | - | - | - | - | |
| ter | methyl thiolacetate | 1534-08-3 | 6.39 | 1078 | 1052 | $9.1\pm0.7\times10^{4}b$ | $2.7\pm0.7\times10^5b$ | $6.6 \pm 2.0 \times 10^6 \text{ a }^*$ | $1.2\pm0.7	imes10^4~{b}$ | - | - | - | - | |
| lsu | indole | 120-72-9 | 38.65 | 2342 | 2345 | $6.0 \pm 2.4 \times 10^5 \mathrm{b}$ | $1.0 \pm 0.2 \times 10^{7}$ a | $5.3 \pm 4.4 \times 10^{5} \text{ b}$ | $7.4 \pm 8.1 \times 10^{5} \mathrm{b}$ | - | - | - | - | |
| 0 | unknown compound 1 | | 32.91 | 2134 | | $6.6 \pm 4.1 \times 10^4 \mathrm{b}$ | $2.1 \pm 1.4 \times 10^4 \mathrm{b}$ | $2.0 \pm 0.6 \times 10^5$ a | $6.9 \pm 4.1 \times 10^4 \mathrm{b}$ | - | - | - | - | |
| | unknown compound 2 | | 34.43 | 2179 | | $9.1 \pm 0.1 \times 10^{3} \mathrm{b}$ | $5.2 \pm 9.7 \times 10^{3} \text{ b}$ | $1.3 \pm 0.7 \times 10^5$ a | $2.1 \pm 2.4 \times 10^4 \text{ b}$ | - | - | - | - | |
| | unknown compound 3 | | 34.82 | 2190 | | $2.9 \pm 2.5 \times 10^{5} \mathrm{b}$ | $2.4 \pm 1.3 \times 10^{5} \text{ b}$ | $3.3 \pm 0.8 \times 10^6$ a | $7.3 \pm 5.8 \times 10^{5} \text{ b}$ | - | - | - | - | |
| | unknown compound 4 | | 35.11 | 2199 | | $2.3 \pm 3.4 \times 10^4 \mathrm{b}$ | $3.8 \pm 1.8 \times 10^4 \mathrm{b}$ | $2.5 \pm 0.4 \times 10^{5}$ a | $1.0 \pm 1.1 \times 10^{5} \mathrm{b}$ | - | - | - | - | |
| | unknown compound 5 | | 29.55 | 1987 | 1000 | $6.3 \pm 4.3 \times 10^3$ b | $1.7 \pm 0.3 \times 10^4 \text{ b}$ | $5.3 \pm 0.9 \times 10^4$ a | $1.4 \pm 0.2 \times 10^4 \text{ b}$ | - | - | - | - | |
| | methyl 2-methyl butanoate | 868-57-5 | 5.27 | 1032 | 1009 | $2.0 \pm 0.1 \times 10^7$ ab | $1.2 \pm 0.1 \times 10^{7} \text{ bc}$ | $2.7 \pm 0.3 \times 10^{7}$ a | $1.3 \pm 0.6 \times 10^{7} \text{ bc}$ | $4.7 \pm 3.2 \times 10^{6} \text{ cd}$ | $1.7 \pm 1.4 \times 10^{6} \mathrm{d}$ | $3.8 \pm 3.5 \times 10^{3} \mathrm{d}$ | $4.3 \pm 3.2 \times 10^{3} \mathrm{d}$ | |
| | 4-methyl-1-pentanol | 626-89-1 | 14.28 | 1343 | 1314 | $4.9 \pm 1.8 \times 10^4$ b | $3.7 \pm 0.9 \times 10^{4} \text{ bc}$ | $9.2 \pm 1.6 \times 10^4$ a | $4.2 \pm 0.9 \times 10^4$ bc | $1.1 \pm 0.9 \times 10^{4} \text{ c}$ | - | - | - | |
| · | (Z)-3-decen-1-ol | 10340-22-4 | 25.97 | 1815 | 1789 | $3.0 \pm 5.6 \times 10^{3}$ b | $9.5 \pm 6.3 \times 10^{3} \text{ b}$ | $1.0 \pm 0.2 \times 10^{5}$ a | $6.9 \pm 8.3 \times 10^{3} \text{ b}$ | - | - | $1.7 \pm 3.7 \times 10^{-1}$ b | - | |
| | n-undecanoic acid methyl | 1/31-86-8 | 22.88 | 16/8 | 1704 | - | $2.7 \pm 5.1 \times 10^{5}$ b | $2.1 \pm 0.6 \times 10^{5}$ a | $1.2 \pm 1.3 \times 10^{4} \text{ b}$ | - | - | - | - | |
| | ester | 71 22 9 | 6 11 | 1066 | 1026 | $1.1 \pm 0.1 \times 105$ do | $65 \pm 0.7 \times 105$ | $4.4 \pm 0.2 \times 105$ h | $24 \pm 0.2 \times 105$ at | $1.7 \pm 0.1 \times 105$ ad | | $25 \pm 0.5 \times 105$ | $21 + 41 \times 104$ | |
| | 1-propanol | /1-23-8 | 0.11 | 1000 | 1030 | $1.1 \pm 0.1 \times 10^{5} \text{ de}$ | $0.5 \pm 0.7 \times 10^{5} a$ | $4.4 \pm 0.3 \times 10^{5}$ D 7.2 + 1.7 × 10 ⁵ c | $2.4 \pm 0.2 \times 10^{5}$ cd | $1.7 \pm 0.1 \times 10^{5}$ cd $1.6 \pm 0.2 \times 10^{5}$ cd | - | $2.5 \pm 0.5 \times 10^{\circ} \text{ C}$ | $2.1 \pm 4.1 \times 10^{4} \text{ e}$ | |
| | 2-Iuranneunanoi 3 mothyl 1 hoyanol | 98-00-0 | 25.17 | 1690 | 1000 | $5.8 \pm 2.1 \times 10^{6} a$ 1.0 ± 0.5 × 105 ab | $4.0 \pm 0.8 \times 10^{\circ} a$ 7 3 ± 1 0 × 10 ⁴ b | $7.2 \pm 1.7 \times 10^{\circ} a$ 1.7 ± 0.2 × 10 ⁵ a | $0.4 \pm 1.3 \times 10^{4} \text{ a}$ 6.1 ± 2.2 × 10 ⁴ h | $4.0 \pm 0.3 \times 10^{6} a^{*}$ | - | $1.0 \pm 3.0 \times 10^{-0}$ | $2.0 \pm 5.9 \times 10^{-10}$ | |
| | 2 undecanol | 1653 30 1 | 24.38 | 1440 | 1413 | $1.0 \pm 0.3 \times 103 \text{ ab}$ $1.0 \pm 1.2 \times 10^4 \text{ a}$ | $7.3 \pm 1.9 \times 10^{-0}$ 8 4 ± 1 9 × 10 ⁴ a | $1.7 \pm 0.2 \times 10^{5} a$ 1.0 + 0.4 × 10 ⁵ a | $0.1 \pm 2.2 \times 10^{-0.1}$ | $4.0 \pm 4.3 \times 10^{-0}$ | - | - | $\frac{-}{13+20} \times 10^{5}$ a | |
| | methyl isobutyrate | 547-63-7 | 24.50 | 033 | 922 | $1.0 \pm 1.2 \times 10^{5} a$ 3.8 ± 0.7 × 10 ⁵ a | $8.4 \pm 1.9 \times 10^{4} \text{ h}$ $8.8 \pm 2.2 \times 10^{4} \text{ h}$ | $1.9 \pm 0.4 \times 10^{5} a$ 3 3 + 0 5 × 10 ⁵ a | $1.1 \pm 1.3 \times 10^{4} \text{ a}$ 5 3 + 2 1 × 10 ⁴ h | $58 + 61 \times 10^4$ h | $-1.6 \pm 2.6 \times 10^{3}$ h | $-1.0 + 3.2 \times 10^4$ h | $1.3 \pm 2.0 \times 10^{4} \text{ a}$ 5 8 + 9 1 × 10 ⁴ h | |
| ster 2 | 2-tridecanone | 593-08-8 | 26.41 | 1836 | 1809 | $3.0 \pm 0.7 \times 10^{4} \text{ h}$ | $1.0 \pm 0.2 \times 10^{5}$ h | $5.5 \pm 0.5 \times 10^{5} a$ 5 $4 \pm 1.6 \times 10^{5} a$ | $3.3 \pm 2.1 \times 10^{-6} \text{ b}$ $3.7 \pm 3.2 \times 10^{4} \text{ b}$ | $3.8 \pm 0.1 \times 10^{4} \text{ b}$ | $1.0 \pm 2.0 \times 10^{-0}$ b 2.6 ± 4.7 × 10 ³ b | $4.0 \pm 3.2 \times 10^{-6} \text{ b}$ $1.1 \pm 1.0 \times 10^{4} \text{ b}$ | $3.0 \pm 9.1 \times 10^{5} \text{ b}$ | |
| | acetone | 67-64-1 | 1 96 | 822 | 819 | $1.4 \pm 0.2 \times 10^{6} \text{ h}$ | $1.0 \pm 0.2 \times 10^{6} \text{ a}$ 2 9 + 0 2 × 10 ⁶ a | $3.4 \pm 1.0 \times 10^{6} a$ $3.2 \pm 0.7 \times 10^{6} a$ | $3.7 \pm 3.2 \times 10^{6} \text{ b}$ 1 4 + 0 3 × 10 ⁶ b | $9.3 \pm 0.3 \times 10^{5} \text{ h}$ | $3.6 \pm 0.3 \times 10^{6} \text{ a}^{*}$ | $1.1 \pm 1.0 \times 10^{6} \text{ b}$ $1.2 \pm 0.1 \times 10^{6} \text{ b}$ | $1.0 \pm 1.1 \times 10^{6} \text{ b}$ $1.4 \pm 0.7 \times 10^{6} \text{ b}$ | |
| lus | unknown compound 6 | 07 04 1 | 11.65 | 1255 | 017 | $1.4 \pm 0.2 \times 10^{3} \text{ b}$ 1 1 + 1 5 × 10 ³ b | $3.8 \pm 4.4 \times 10^3$ h | $3.2 \pm 0.7 \times 10^{5} a$ $3.6 \pm 1.8 \times 10^{5} a$ | - | $7.0 \pm 7.0 \times 10^{3} \text{ h}$ | - - 0.5 × 10 a | $3.4 + 5.5 \times 10^3$ h | $3.3 \pm 3.8 \times 10^4$ h | |
| 0 | unknown compound 7 | | 27.6 | 1892 | | $1.1 \pm 1.5 \times 10^{5} \text{ b}$ $1.1 \pm 0.6 \times 10^{5} \text{ b}$ | $1.6 \pm 0.7 \times 10^5$ b | $1.3 \pm 0.4 \times 10^6$ a | $1.2 \pm 0.5 \times 10^5$ b | $1.2 \pm 0.2 \times 10^5 \text{ b}$ | $8.1 \pm 0.4 \times 10^4$ b | $1.0 \pm 0.3 \times 10^{5} \text{ b}$ | $2.0 \pm 0.9 \times 10^5$ b | |
| | 2-butanol | 78-92-2 | 3.31 | 937 | 927 | $1.4 \pm 0.7 \times 10^4 \text{ b}$ | $1.6 \pm 0.2 \times 10^6$ a * | $1.6 \pm 0.2 \times 10^4 \text{ b}$ | $6.1 \pm 2.1 \times 10^4 \text{ b}$ | $3.1 \pm 4.7 \times 10^3$ b | $1.1 \pm 0.8 \times 10^4 \text{ b}$ | $2.9 \pm 3.6 \times 10^{3}$ b | $2.2 \pm 1.6 \times 10^4$ b | |
| | isoamvl alcohol | 123-51-3 | 11.35 | 1246 | 1209 | $1.2 \pm 0.5 \times 10^7$ ab | $2.0 \pm 0.6 \times 10^7$ a | $2.0 \pm 0.3 \times 10^7$ a | $2.2 \pm 0.4 \times 10^7$ a | $1.9 \pm 0.1 \times 10^7$ a | $4.6 \pm 0.4 \times 10^{6}$ ab | $2.4 \pm 1.9 \times 10^{6} \mathrm{b}$ | $1.5 \pm 1.0 \times 10^7$ ab | |
| | phenylethyl alcohol | 60-12-8 | 28.68 | 1944 | 1906 | $1.8 \pm 0.6 \times 10^{6} \text{ ab}$ | $2.2 \pm 0.3 \times 10^{6}$ ab | $2.4 \pm 0.5 \times 10^{6}$ a | $2.1 \pm 0.6 \times 10^{6}$ ab | $7.4 \pm 2.9 \times 10^5$ ab | $7.0 \pm 4.9 \times 10^5$ ab | $5.5 \pm 3.7 \times 10^{5} \mathrm{b}$ | $1.0 \pm 0.8 \times 10^{6} \text{ ab}$ | |
| | ethanol | 64-17-5 | 3.45 | 944 | 932 | $1.3 \pm 0.1 \times 10^{6} \text{ ab}$ | $2.2\pm0.3	imes10^{6}$ a | $1.4\pm0.1	imes10^{6}~\mathrm{ab}$ | $1.5\pm0.2	imes10^{6}~\mathrm{ab}$ | $3.9 \pm 0.3 \times 10^{5}$ ab | $5.2\pm3.6\times10^3b$ | $3.0\pm0.2\times10^5b$ | $7.3 \pm 1.4 \times 10^{5}$ ab | |
| | acetoin | 513-86-0 | 13.43 | 1313 | 1284 | $1.5 \pm 0.4 \times 10^{6}$ a | $1.4\pm0.5	imes10^{6}~\mathrm{ab}$ | $8.0 \pm 5.3 \times 10^{5}$ abc | $5.3 \pm 5.1 \times 10^{5}$ abc | $1.8\pm0.2\times10^{5}\mathrm{bc}$ | $1.8\pm2.2\times10^4~c$ | $1.1\pm0.2	imes10^5~{ m c}$ | $9.5\pm6.0\times10^4c$ | |
| | unknown compound 8 | | 27.12 | 1869 | | - | - | $8.8\pm2.9	imes10^4~{ m a}$ | $1.3\pm2.6\times10^{4}b$ | - | - | - | - | |
| | 2,5-dimethyl pyrazine | 123-32-0 | 14.33 | 1345 | 1320 | $4.5\pm1.9\times10^4b$ | $6.9\pm1.4	imes10^4~{b}$ | $1.3\pm0.3	imes10^{5}$ b | $6.1 \pm 1.6 	imes 10^4 \mathrm{b}$ | $2.7 \pm 1.0 \times 10^{6}$ a | $3.4 \pm 0.4 \times 10^{6}$ a | $2.6\pm0.4	imes10^{6}$ a | $2.4\pm0.7	imes10^{6}$ a | |
| | 2,6-dimethyl pyrazine | 108-50-9 | 14.53 | 1352 | 1328 | $4.0\pm2.2	imes10^4~{b}$ | $6.3\pm1.6\times10^4b$ | $1.3\pm0.4	imes10^{5}$ b | $5.9\pm1.2	imes10^4~b$ | $3.7 \pm 1.4 \times 10^{6} \text{ a}$ | $4.2\pm0.6	imes10^{6}$ a | $3.4\pm1.1	imes10^{6}$ a | $3.0\pm1.4	imes10^{6}$ a | |
| | 2,4,6-trimethyl pyridine | 108-75-8 | 15.59 | 1389 | 1371 | $6.2\pm4.5\times10^3b$ | $9.5\pm4.7\times10^3b$ | $1.7\pm0.9\times10^{4}~b$ | $1.1\pm0.5	imes10^4b$ | $5.2 \pm 0.8 \times 10^5 \text{ a}$ | $5.6\pm0.7	imes10^5$ a | $5.8\pm0.6	imes10^5~a$ | $5.7\pm0.6	imes10^5$ a | * |
| ~ | 1-(2-furanyl)-ethanone | 1192-62-7 | 19.28 | 1530 | 1499 | $1.4 \pm 0.6 \times 10^5 \text{ cd}$ | $9.5\pm1.8\times10^4d$ | $2.9\pm0.8	imes10^5$ bc | $6.2 \pm 1.7 \times 10^4 \text{ d}$ | $4.3 \pm 0.2 \times 10^5$ ab | $4.2 \pm 0.3 \times 10^{5}$ ab | $4.5 \pm 0.2 \times 10^5$ a | $4.0 \pm 0.6 \times 10^5$ ab | |
| er () | unknown compound 9 | | 16.87 | 1437 | | - | - | - | - | $1.6 \pm 0.2 \times 10^5$ a | $1.8 \pm 0.1 \times 10^5$ a | $1.6 \pm 0.1 \times 10^5$ a | $1.7 \pm 0.1 \times 10^5$ a | |
| uste | 2-butanone | 78-93-3 | 2.86 | 913 | 907 | $2.9 \pm 0.2 \times 10^5 \mathrm{d}$ | $3.2 \pm 0.2 \times 10^5 \mathrm{d}$ | $3.9 \pm 2.2 \times 10^5 \mathrm{d}$ | $3.9 \pm 0.5 \times 10^5 \mathrm{d}$ | $1.2 \pm 0.4 \times 10^{6} \mathrm{c}$ | $2.1 \pm 0.1 \times 10^{6}$ a | $1.4 \pm 0.1 \times 10^{6} \mathrm{c}$ | $1.8 \pm 0.2 \times 10^{6} \mathrm{b}$ | |
| Ū | pyrrole | 109-97-7 | 19.52 | 1539 | 1514 | $4.8 \pm 6.3 \times 10^{3} \mathrm{d}$ | $1.3 \pm 0.1 \times 10^4 d$ | $1.5 \pm 0.9 \times 10^4 \mathrm{d}$ | $1.9 \pm 0.4 \times 10^4 \mathrm{d}$ | $1.1 \pm 0.1 \times 10^{6} \mathrm{c}$ | $3.1 \pm 0.3 \times 10^6$ a | $1.7 \pm 0.1 \times 10^{6} \mathrm{b}$ | $2.0 \pm 0.2 \times 10^{6} \mathrm{b}$ | |
| | 2-methoxy-3-(1-methyl- | 24168-70-5 | 19.23 | 1528 | 1500 | $1.1 \pm 0.8 \times 10^4 \mathrm{d}$ | - | $1.6 \pm 0.4 \times 10^{5} \text{ c}$ | - | $3.5 \pm 0.2 \times 10^{5} \mathrm{b}$ | $3.6 \pm 2.3 \times 10^4 \mathrm{d}$ | $7.4 \pm 0.3 \times 10^{5}$ a | $3.0 \pm 5.9 \times 10^4 \mathrm{d}$ | |
| | propyl) pyrazine | | 00.44 | 1 | | 15.05.102 | | | | 24.07.1051 | 20.15.104 | 0.0 | | |
| | unknown compound 10 | | 20.41 | 1575 | | $1.5 \pm 2.5 \times 10^{3} \text{ c}$ | - | $7.5 \pm 1.9 \times 10^4 \text{ c}$ | - | $3.4 \pm 0.7 \times 10^{5} \text{ b}$ | $2.9 \pm 1.5 \times 10^4 \text{ c}$ | $8.8 \pm 0.4 \times 10^{5}$ a | $2.3 \pm 2.0 \times 10^4 \text{ c}$ | |
| | unknown compound 11 | 71.04.0 | 18.05 | 1482 | 11.40 | - | - | $8.0 \pm 1.9 \times 10^4 \text{ b}$ | - | $2.9 \pm 2.4 \times 10^{5}$ ab | $9.5 \pm 0.2 \times 10^4 \text{ b}$ | $8.7 \pm 2.8 \times 10^{5} a$ | $2.5 \pm 0.4 \times 10^{5}$ ab | |
| | 1-butanol | /1-36-3 | 9.19 | 1175 | 1142 | $0.5 \pm 2.8 \times 10^{3} \text{ b}$ | $6.1 \pm 1.8 \times 10^{3} \text{ b}$ | $7.8 \pm 2.2 \times 10^{3} \text{ b}$ | $9.1 \pm 1.7 \times 10^{3} \text{ b}$ | $5.9 \pm 0.6 \times 10^{\circ} a$ | $1.5 \pm 0.6 \times 10^4$ b | $1.7 \pm 1.3 \times 10^{\circ} \text{ b}$ | $2.2 \pm 0.5 \times 10^{3} \text{ b}$ | |
| | 2-nonanone | 821-55-6 | 16.24 | 1413 | 1390 | $5.7 \pm 1.8 \times 10^4$ a | $1.1 \pm 0.9 \times 10^{4} \text{ a}$ | $2.0 \pm 0.5 \times 10^{3}$ a | - | $9.6 \pm 3.4 \times 10^4$ a | $1.5 \pm 0.2 \times 10^{3} a$ | $1.8 \pm 0.4 \times 10^{5}$ a | $4.9 \pm 5.1 \times 10^{3} a$ | |
| | ainydro-3-methyl-2(3H)- | 16/9-4/-6 | 21.34 | 1612 | 1585 | - | - | - | - | - | $1.9 \pm 2.2 \times 10^{4} \text{ b}$ | $1.3 \pm 0.1 \times 10^{5}$ a | $1.6 \pm 1.9 \times 10^{4} \text{ b}$ | |
| r 4 | luranone | 110 21 0 | 10.14 | 1504 | 1409 | | | | | $21 + 0.7 \times 1041$ | $21 \pm 20 \times 103$ - | $9.2 \pm 1.1 \pm 104$ | | |
| ıste | unknown compound 10 | 112-31-2 | 19.14 | 1524 | 1498 | - | - | $-7.2 \pm 1.2 \times 10^{4}$ | - | $3.1 \pm 0.7 \times 10^{\circ} \text{ D}$ $2.0 \pm 0.2 \times 10^{\circ} \text{ b}$ | $2.1 \pm 3.9 \times 10^{5} \text{ C}$ | $0.5 \pm 1.1 \times 10^{-6} a$ | $-27 \pm 5.4 \times 10^{4}$ | |
| Clu | 3-methoxy 2.5 dimothyl | 198/6 22 1 | 17.33 | 1340 | 1/10 | - 93 + 10×10^{3} o | - | $1.2 \pm 1.3 \times 10^{-10}$ 1.2 + 0.4 × 10 ⁵ a | - | $2.0 \pm 0.2 \times 10^{\circ} \text{ D}$ 1 2 + 0 1 × 10 ⁶ o | $-1.5 + 6.6 \times 10^{5}$ a | $1.0 \pm 0.1 \times 10^{\circ} a$ 1.0 + 1.2 × 106 a | $2.7 \pm 3.4 \times 10^{-6} \text{ C}$ 9.2 + 1.4 × 10 ⁵ c | |
| - | pyrazine | 17040-22-1 | 17.50 | 1450 | 1417 | $\int J J = 4.0 \times 10^{\circ} a$ | - | $1.2 \pm 0.4 \times 10^{\circ}$ a | - | $1.2 \pm 0.1 \times 10^{-} a$ | $+.5 \pm 0.0 \times 10^{\circ}$ a | $1.7 \pm 1.2 \times 10^{-3} d$ | $7.2 \pm 1.4 \times 10^{-}$ a | |
| | 2-methoxy-3-methyl pyrazine | 2847-30-5 | 15.81 | 1396 | 1339 | - | - | $7.7\pm2.4\times10^3b$ | - | - | - | $2.1 \pm 0.2 \times 10^5$ a * | - | |

| 2-methoxy-6-methyl | 2882-21-5 | 16.49 | 1422 | 1371 | - | - | - | - | - | - | $3.1\pm0.3\times10^5$ a * | - |
|---------------------------|------------|-------|------|------|-------------------------------------|--------------------------------------|--|--|--|--------------------------------------|--------------------------------------|--|
| pyrazine | | | | | | | | | | | | |
| unknown compound 13 | | 19.13 | 1524 | | - | - | - | - | $2.0 \pm 0.8 \times 10^{5} \text{ b}$ | - | $4.2 \pm 0.3 \times 10^{3}$ a | - |
| unknown compound 14 | | 20.25 | 1568 | | - | - | - | - | $1.8 \pm 0.1 \times 10^5$ a * | - | $2.3 \pm 0.4 \times 10^4 b$ | - |
| unknown compound 15 | | 19.6 | 1543 | | - | - | $1.1 \pm 0.6 \times 10^4 \text{ c}$ | - | $1.3 \pm 0.3 \times 10^{5} \mathrm{b}$ | - | $1.8 \pm 0.2 \times 10^{5}$ a | - |
| unknown compound 16 | | 18.87 | 1514 | | - | - | - | - | $6.3 \pm 3.4 \times 10^5$ a | - | $7.3 \pm 1.7 \times 10^4 \mathrm{b}$ | - |
| unknown compound 17 | | 4.4 | 995 | | $2.6 \pm 1.2 \times 10^4 \text{ b}$ | $1.9 \pm 1.5 	imes 10^4 	ext{ b}$ | $2.2 \pm 0.2 \times 10^{5} \text{ b}$ | $3.7 \pm 2.0 \times 10^4 \text{ b}$ | $7.9 \pm 0.3 \times 10^3 \mathrm{b}$ | $2.7 \pm 2.3 \times 10^4 \mathrm{b}$ | $2.4 \pm 0.4 \times 10^{6}$ a | $2.7 \pm 1.8 \times 10^5 \text{ b}$ |
| methyl 2-methyl butanoate | 868-57-5 | 10.65 | 1223 | 1229 | $5.0 \pm 2.0 \times 10^5$ ab * | $3.2 \pm 1.4 \times 10^4 \mathrm{c}$ | $1.8 \pm 0.4 \times 10^{5} \text{ bc}$ | $5.9 \pm 4.7 \times 10^4 \text{ c}$ | $7.1 \pm 2.7 \times 10^5$ a | - | $6.8 \pm 6.3 \times 10^4 \text{ c}$ | - |
| methyl 3-methyl butanoate | 556-24-1 | 5.54 | 1043 | 1018 | $1.2 \pm 0.1 \times 10^{6} a^{*}$ | $8.5 \pm 1.0 \times 10^{5}$ ab | $8.7 \pm 2.5 \times 10^5$ ab | $1.0\pm0.3\times10^5b$ | $1.2 \pm 0.6 \times 10^{6} \text{ a}$ | $6.3\pm3.0	imes10^4~{b}$ | $5.0 \pm 3.2 \times 10^{5}$ ab | $1.9 \pm 1.3 	imes 10^5 	ext{ b}$ |
| 2-methyl-1-propanol | 78-83-1 | 7.69 | 1125 | 1092 | $1.4\pm0.1	imes10^{6}~\mathrm{ab}$ | $1.5\pm0.1	imes10^{6}~\mathrm{ab}$ | $1.9\pm0.1	imes10^{6}~\mathrm{ab}$ | $1.6\pm0.2	imes10^{6}$ ab | $2.2 \pm 0.1 \times 10^{6} a$ | $7.8\pm0.7	imes10^{5}\mathrm{b}$ | $2.2\pm0.4	imes10^{6}$ a | $1.4\pm0.9	imes10^{6}~\mathrm{ab}$ |
| 3-octanol | 589-98-0 | 16.33 | 1416 | 1393 | $1.6 \pm 0.9 \times 10^4$ ab | $4.0\pm0.7	imes10^4~ab$ | $1.6 \pm 2.1 \times 10^4 \text{ ab}$ | $4.7 \pm 1.6 \times 10^4$ ab | $6.0\pm4.3\times10^3b$ | $2.4 \pm 1.4 \times 10^4$ ab | $1.1 \pm 0.5 \times 10^4$ ab | 5.3 ± 1.6 × 10 ⁴ a * |
| 2-methoxy-3-(1- | 25773-40-4 | 17.51 | 1461 | 1427 | - | - | $2.9\pm0.9	imes10^4~ab$ | - | $7.6\pm5.2\times10^3b$ | - | $7.9\pm4.4	imes10^4~{ m a}$ | - |
| methylethyl) pyrazine | | | | | | | | | | | | |
| methyl isobutyl ketone | 108-10-1 | 5.16 | 1027 | 1010 | $1.4\pm0.3\times10^{4}b$ | $1.5\pm0.2	imes10^{5}$ a | $6.7\pm0.9	imes10^4~\mathrm{ab}$ | $9.4 \pm 1.3 	imes 10^3 b$ | $3.8\pm2.0\times10^4b$ | $5.5 \pm 1.6 \times 10^4$ ab | $6.0 \pm 0.9 \times 10^4$ ab | $6.7 \pm 7.9 	imes 10^4 	ext{ ab}$ |
| unknown compound 18 | | 22.93 | 1680 | | $8.2\pm6.3	imes10^4\mathrm{b}$ | $2.4\pm0.9	imes10^{5}$ ab | $1.6 \pm 0.6 \times 10^{6}$ a | $9.4\pm5.2\times10^{4}b$ | $5.4 \pm 0.7 \times 10^{5}$ ab | $3.0\pm0.7	imes10^{5}~ab$ | $5.4 \pm 1.1 \times 10^{5}$ ab | $1.4 \pm 1.0 \times 10^{6} \text{ ab}$ |
| unknown compound 19 | | 12.25 | 1274 | | - | $6.6\pm0.1	imes10^3b$ | $9.5\pm0.1	imes10^3~b$ | $5.9 \pm 4.2 \times 10^{4} \text{ ab}$ | $2.5\pm0.2\times10^4b$ | $2.8\pm0.6\times10^4b$ | $8.2\pm6.3	imes10^3\mathrm{b}$ | $1.6 \pm 0.9 \times 10^5 \mathrm{a}$ |
| unknown compound 20 | | 15.14 | 1373 | | - | - | $2.9 \pm 0.4 \times 10^5$ ab | $7.8\pm1.4\times10^3b$ | - | $7.9\pm6.2\times10^3b$ | $4.3 \pm 2.7 \times 10^5$ a | $1.9 \pm 1.6 	imes 10^4 	ext{ b}$ |
| unknown compound 21 | | 25.32 | 1786 | | - | - | $1.8 \pm 0.5 \times 10^5$ a | - | - | - | - | $1.2 \pm 2.1 \times 10^5$ a |
| unknown compound 22 | | 26.88 | 1858 | | - | $9.5\pm1.9\times10^3b$ | $2.0 \pm 0.7 \times 10^5$ a | - | - | - | - | $2.0\pm2.1	imes10^4~\mathrm{b}$ |
| 6-methyl-2-heptanol | 4730-22-7 | 15.81 | 1396 | 1372 | $2.1\pm0.9	imes10^4~{ m a}$ | $1.0 \pm 0.2 \times 10^{5}$ a | $6.7 \pm 1.7 \times 10^4 \text{ a}$ | $2.0 \pm 1.4 \times 10^4$ a | $9.0 \pm 3.1 \times 10^4$ a | $3.1 \pm 3.6 \times 10^4 \text{ a}$ | - | $1.6 \pm 2.0 \times 10^5 \text{ a}$ |
| 2-nonanol | 628-99-9 | 19.6 | 1543 | 1521 | - | $7.5 \pm 1.2 \times 10^4 \text{ a}$ | $1.4 \pm 0.3 \times 10^5$ a | $9.2 \pm 1.8 \times 10^4 \text{ a}$ | $3.8 \pm 1.8 \times 10^4 \text{ a}$ | $1.1 \pm 1.4 	imes 10^4$ a | $6.3 \pm 5.3 \times 10^4 \text{ a}$ | $5.4 \pm 7.1 \times 10^5$ a |
| 2-decanol | 1120-06-5 | 20.96 | 1596 | 1601 | $1.9 \pm 1.7 \times 10^4 \text{ a}$ | $1.6\pm0.2	imes10^{5}$ a | $3.3 \pm 0.8 \times 10^5 \text{ a}$ | - | $6.1 \pm 1.6 \times 10^4 \text{ a}$ | $3.4\pm2.8	imes10^4~\mathrm{a}$ | $1.1 \pm 0.9 \times 10^5$ a | $1.8 \pm 2.3 \times 10^{6} a$ |
| 2-pentanone | 107-87-9 | 4.4 | 995 | 981 | $2.6 \pm 0.6 \times 10^4$ a | $4.9 \pm 0.4 \times 10^4 \text{ a}$ | $7.7 \pm 0.7 \times 10^4 \text{ a}$ | $7.0 \pm 8.2 \times 10^3$ a | $1.2 \pm 0.4 \times 10^5$ a | $1.1 \pm 0.4 \times 10^{5}$ a | $4.6 \pm 1.3 \times 10^4 \text{ a}$ | $5.9 \pm 1.1 \times 10^5$ a |
| dimethyl disulfide | 624-92-0 | 7.01 | 1103 | 1077 | $1.0 \pm 0.5 \times 10^5$ a | $4.8 \pm 7.3 \times 10^5$ a | $2.2 \pm 0.6 \times 10^{6}$ a | $2.0 \pm 0.6 \times 10^{6}$ a | $2.7 \pm 0.7 \times 10^5$ a | $4.3 \pm 0.8 \times 10^5$ a | $1.3 \pm 1.9 \times 10^{6}$ a | $6.5 \pm 4.6 \times 10^5$ a |
| 2-heptanone | 110-43-0 | 10.28 | 1211 | 1182 | $1.6 \pm 0.6 \times 10^5$ a | $2.4 \pm 0.3 \times 10^5$ a | $7.0 \pm 1.9 \times 10^5$ a | $5.3 \pm 2.1 \times 10^4$ a | $5.6 \pm 2.6 \times 10^5$ a | $8.1 \pm 1.2 \times 10^5$ a | $7.2 \pm 0.8 \times 10^5$ a | $1.5 \pm 2.4 \times 10^{6}$ a |
| 6-methyl-2-heptanone | 928-68-7 | 11.99 | 1265 | 1237 | $2.3 \pm 1.2 \times 10^5$ a | $2.7 \pm 0.4 \times 10^5$ a | $8.0 \pm 1.7 \times 10^{5}$ a | $1.7 \pm 0.5 \times 10^4$ a | $9.5 \pm 3.7 \times 10^5$ a | $8.7 \pm 1.7 	imes 10^5$ a | $1.2 \pm 0.2 \times 10^{6}$ a | $9.3 \pm 1.3 \times 10^5$ a |
| 5-methyl-2-heptanol | 54630-50-1 | 15.93 | 1401 | 1395 | $2.2 \pm 4.1 \times 10^3$ a | $3.6 \pm 0.5 \times 10^4$ a | $1.7 \pm 0.5 \times 10^4$ a | $5.3 \pm 6.6 \times 10^3$ a | $7.0 \pm 8.0 \times 10^3$ a | $4.2 \pm 7.9 \times 10^3$ a | $6.5 \pm 1.4 \times 10^3$ a | $4.5 \pm 5.7 \times 10^4$ a |
| unknown compound 23 | | 17.79 | 1472 | | $1.3 \pm 0.7 \times 10^5$ a | $1.3 \pm 0.3 \times 10^5$ a | $4.0 \pm 0.9 \times 10^5$ a | $6.8 \pm 8.5 \times 10^3$ a | $3.5 \pm 0.6 \times 10^5$ a | $2.1 \pm 0.5 \times 10^5$ a | $3.7 \pm 0.7 \times 10^5$ a | $1.6 \pm 1.6 \times 10^{6} a$ |
| unknown compound 24 | | 25.58 | 1797 | | $1.0 \pm 1.3 \times 10^4$ a | $7.3 \pm 1.9 \times 10^4$ a | $2.2 \pm 0.4 \times 10^5$ a | $1.8 \pm 1.4 \times 10^4 \text{ a}$ | $9.4 \pm 4.1 \times 10^4$ a | $1.0 \pm 0.1 \times 10^5$ a | $1.2 \pm 0.2 \times 10^5$ a | $3.4 \pm 4.4 \times 10^5 \text{ a}$ |
| * | | | | | 4 | | | | • | | | |

Headspace volatile organic compound (VOC) analysis was carried out by solid-phase microextraction gas chromatography-mass spectrometry (SPME/GC-MS) for Lysobacter antibioticus DSM 2044^T, L. capsici DSM 19286^T, L. enzymogenes DSM 2043^T and L. gummosus DSM 6980^T grown for days on nutrient agar (NA) or potato dextrose agar (PDA).

CAS registry numbers of compound annotation are reported (<u>http://webbook.nist.gov/chemistry/</u>). Mean ± standard deviation values of the absolute peak area from four to five replicates are reported for each compound, *Lysobacter* type strain and growth media. For each compound, different letters indicate significant differences according to Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$). Only compounds with a background-corrected headspace concentration significantly emitted (Kruskal-Wallis with Bonferroni correction, $p \le 0.05$) in *Lysobacter*-inoculated headspace vials as compared with uninoculated vials for at least one strain and growth medium are reported. VOCs were grouped based on their emission profiles in: VOCs with higher emission by all Lysobacter type strains on PDA as compared with NA (Cluster 1), VOCs with higher emission by some Lysobacter type strains on PDA as compared with NA (Cluster 2), VOCs with higher emission by all Lysobacter type strains on NA as compared with PDA (Cluster 3), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as compared with PDA (Cluster 4), VOCs with higher emission by some Lysobacter type strains on NA as different (Cluster 5) or consistent (Cluster 6) emission by Lysobacter type strains on both growth media.

Asterisks in bold (*) indicate VOC emission values associated to non-zero coefficients of the least absolute shrinkage and selection operator (LASSO) model calculated for the growth media prediction (column LASSO) and Lysobacter type strain prediction on PDA (columns Potato dextrose agar) and NA (columns Nutrient agar). Non-zero coefficients specify VOCs that can be used to predict the growth media and the Lysobacter type strain, respectively.

| | <i>m/z</i> value | | Potato | dextrose agar | | | Nut | rient agar | | LASSO |
|-----|------------------|--|--|---|--|--------------------|----------------------|----------------------|----------------------|-------|
| | | Lysobacter | Lysobacter | Lysobacter | Lysobacter | Lysobacter | Lysobacter | Lysobacter | Lysobacter | |
| | | antibioticus | capsici | enzymogenes | gummosus | antibioticus | capsici | enzymogenes | gummosus | |
| | 49.023 | $0.213 \pm 0.069 \text{ b}$ | 0.158 ± 0.019 b | 4.637 ± 2.041 a | $0.123 \pm 0.009 \text{ b}$ | - | - | - | - | |
| | 49.054 | $0.193 \pm 0.049 \text{ b}$ | $0.254 \pm 0.02 \text{ b}$ | 0.542 ± 0.147 a | $0.161 \pm 0.025 \text{ b}$ | - | - | - | - | |
| | 61.054 | $0.763 \pm 0.168 \text{ c}$ | 1.998 ± 0.4 a | 1.66 ± 0.133 ab | 0.994 ± 0.136 bc | - | - | - | - | |
| | 62.057 | $0.024 \pm 0.007 \text{ b}$ | 0.056 ± 0.012 a | 0.05 ± 0.002 ab | 0.033 ± 0.004 ab | - | - | - | - | |
| | 63.984 | $0.034 \pm 0.008 \text{ b}$ | $0.03 \pm 0.008 \text{ b}$ | 0.097 ± 0.01 a | 0.081 ± 0.007 a | - | - | - | - | |
| | 64.007 | 0.063 ± 0.007 a | 0.042 ± 0.008 a | 0.064 ± 0.004 a | 0.042 ± 0.011 a | - | - | - | - | |
| | 68.050 | 2.712 ± 0.226 a | 3.235 ± 0.874 a | 2.293 ± 0.229 a | 2.672 ± 0.49 a | - | - | - | - | * |
| | 79.074 | 0.023 ± 0.006 b | 0.07 ± 0.006 a | $0.037 \pm 0.006 \text{ b}$ | 0.028 ± 0.004 b | - | - | - | - | |
| | 84.004 | 0.064 ± 0.015 ab | 0.094 ± 0.009 a | $0.053 \pm 0.004 \text{ b}$ | 0.088 ± 0.009 ab | - | - | - | - | |
| | 85.016 | 1.331 ± 0.284 b | 2.075 ± 0.092 a | 1.283 ± 0.098 b | 1.786 ± 0.131 ab | - | - | - | - | |
| | 89.045 | 0.129 ± 0.021 b | 0.231 ± 0.029 a | 0.136 ± 0.024 b | 0.106 ± 0.017 b | - | - | - | - | |
| | 92.989 | 0.012 ± 0.005 b | 0.016 ± 0.009 b | 0.125 ± 0.026 a | 0.015 ± 0.004 b | - | - | - | - | |
| | 93.022 | 0.107 ± 0.035 b | $0.097 \pm 0.009 \text{ b}$ | 0.302 ± 0.03 a | 0.094 ± 0.032 b | - | - | - | - | |
| | 110.949 | 0.01 + 0.004 b | 0.009 + 0.002 b | 0.117 + 0.017 a | 0.016 ± 0.007 b | - | - | - | - | |
| | 114.065 | 0.033 ± 0.003 a | 0.021 + 0.002 b | 0.022 + 0.003 b | 0.022 ± 0.003 b | - | - | - | - | |
| | 116.913 | 0.05 ± 0.009 a | 0.05 ± 0.004 a | 0.05 ± 0.009 a | 0.032 ± 0.004 a | - | - | - | - | |
| | 116.964 | 0.082 ± 0.026 a | 0.071 ± 0.013 a | 0.091 ± 0.019 a | 0.03 ± 0.007 a | - | - | - | - | |
| | 117.076 | 3.083 ± 0.694 a | 2.909 ± 0.474 ab | 3.467 ± 0.605 a | 1.085 ± 0.362 b | - | - | - | - | |
| | 118.053 | 0.002 ± 0.002 h | 0.152 ± 0.066 a | 0.023 ± 0.024 ab | 0.005 ± 0.004 b | - | - | - | - | |
| | 119.059 | 0.035 ± 0.008 b | 0.073 ± 0.019 b | 0.36 ± 0.024 a | 0.025 ± 0.006 b | - | - | - | - | |
| | 129.046 | $0.066 \pm 0.007 \text{ b}$ | 0.064 ± 0.006 b | 0.452 ± 0.038 a | $0.082 \pm 0.006 \text{ b}$ | - | - | - | - | |
| _ | 129.091 | 0.116 ± 0.011 a | 0.085 ± 0.006 b | 0.09 ± 0.002 ab | 0.07 ± 0.01 h | _ | - | - | - | * |
| er | 129.132 | 0.086 ± 0.008 b | 0.065 ± 0.008 b | 0.125 ± 0.007 a | 0.026 ± 0.004 c | - | - | - | - | |
| ust | 135 049 | $0.020 \pm 0.000 \text{ b}$ $0.024 \pm 0.002 \text{ b}$ | $0.005 \pm 0.000 \text{ b}$ $0.025 \pm 0.002 \text{ b}$ | 0.046 ± 0.004 a | $0.020 \pm 0.001 \text{ c}$ $0.025 \pm 0.005 \text{ b}$ | _ | - | - | - | |
| Ū | 135.086 | 0.021 ± 0.002 b 0.043 ± 0.005 b | 0.029 ± 0.002 b 0.038 ± 0.004 b | 0.061 ± 0.002 a | 0.025 ± 0.005 b | _ | - | - | - | |
| | 139.086 | 0.103 ± 0.009 b | 0.078 ± 0.01 hc | 0.149 ± 0.011 a | 0.058 ± 0.007 c | _ | - | - | - | |
| | 141.927 | 0.018 ± 0.01 b | 0.002 ± 0.002 b | 0.049 ± 0.007 a | 0.066 ± 0.006 a | - | - | - | - | |
| | 143 112 | 0.062 ± 0.007 ab | 0.002 ± 0.002 b 0.041 + 0.009 b | 0.019 ± 0.007 a 0.106 + 0.025 a | $0.000 \pm 0.000 \text{ a}$ $0.055 \pm 0.008 \text{ b}$ | _ | - | - | - | |
| | 144 151 | 0.002 ± 0.007 do | 0.001 ± 0.002 h | 0.018 ± 0.004 a | $0.002 \pm 0.002 \text{ h}$ | _ | _ | _ | - | |
| | 153 101 | $0.069 \pm 0.006 ab$ | 0.000 ± 0.002 b 0.078 + 0.013 ab | $0.010 \pm 0.004 \text{a}$ | 0.002 ± 0.002 b 0.046 + 0.005 b | _ | - | - | - | |
| | 159.096 | $0.02 \pm 0.005 \text{ h}$ | 0.016 ± 0.003 h | 0.049 ± 0.004 a | 0.010 ± 0.000 b 0.011 ± 0.004 b | _ | _ | _ | - | |
| | 163 150 | 0.02 ± 0.003 b 0.02 ± 0.004 b | 0.010 ± 0.000 b 0.017 ± 0.002 b | $0.049 \pm 0.004 \text{a}$ | $0.011 \pm 0.004 0$ $0.063 \pm 0.008 a$ | _ | _ | _ | _ | |
| | 167 158 | $0.02 \pm 0.004 0$ $0.021 \pm 0.003 a$ | 0.017 ± 0.002 b 0.009 ± 0.002 b | 0.010 ± 0.003 a | $0.005 \pm 0.000 \text{a}$ | _ | _ | _ | - | |
| | 171 182 | $0.021 \pm 0.005 \text{ a}$ $0.04 \pm 0.012 \text{ b}$ | 0.009 ± 0.002 b 0.048 + 0.005 b | $0.02 \pm 0.003 \mathrm{a}$ 0.177 + 0.015 a | 0.000 ± 0.001 b 0.015 ± 0.003 b | _ | _ | _ | - | |
| | 172 170 | 0.04 ± 0.012 b 0.009 ± 0.002 b | 0.040 ± 0.003 b | 0.025 ± 0.005 a | $0.013 \pm 0.003 \text{ b}$ $0.003 \pm 0.001 \text{ b}$ | _ | _ | _ | - | |
| | 181 144 | 0.007 ± 0.002 b 0.027 ± 0.003 b | $0.009 \pm 0.009 b$ 0.019 ± 0.002 b | $0.023 \pm 0.003 \mathrm{a}$ | 0.005 ± 0.001 b 0.015 ± 0.002 b | _ | _ | _ | _ | |
| | 185 157 | 0.027 ± 0.003 0 0.026 ± 0.004 ab | 0.017 ± 0.002 b 0.022 ± 0.005 b | $0.007 \pm 0.009 a$ $0.046 \pm 0.008 a$ | 0.013 ± 0.002 b 0.018 ± 0.003 b | | _ | _ | _ | |
| | 185 199 | 0.020 ± 0.004 ab 0.02 + 0.008 h | 0.022 ± 0.005 b 0.039 ± 0.008 b | $0.040 \pm 0.000 a$ | 0.010 ± 0.000 b | | _ | _ | _ | |
| | 197 211 | 0.02 ± 0.0000 | 0.009 ± 0.0000 | 0.03 ± 0.004 a | 0.02 ± 0.0070 | _ | _ | - | - | |
| | 199.106 | 0.007 ± 0.002 0 0.032 ± 0.003 a | 0.007 ± 0.001 b 0.021 ± 0.002 h | $0.03 \pm 0.004 a$ 0.017 + 0.002 h | 0.011 ± 0.002 b 0.015 + 0.003 h | | _ | - | _ | |
| | 199.100 | $0.052 \pm 0.005 a$ 0.236 ± 0.031 a | 0.021 ± 0.0020 0.014 ± 0.006 b | 0.017 ± 0.0020 0.003 ± 0.004 h | 0.013 ± 0.003 b 0.011 + 0.004 b | | - | - | - | |
| | 200 186 | $0.230 \pm 0.031 a$ $0.047 \pm 0.006 a$ | 0.014 ± 0.0000 | 0.003 ± 0.004 b 0.01 + 0.002 h | 0.011 ± 0.004 D 0.007 + 0.002 h | | _ | - | _ | |
| | 210.100 | $0.047 \pm 0.000 a$ $0.014 \pm 0.002 h$ | 0.003 ± 0.002 b 0.012 ± 0.002 b | 0.01 ± 0.002 b 0.01 + 0.002 h | 0.007 ± 0.0020 | | - | - | - | |
| | 219.217 | 0.014 ± 0.002 b 0.005 + 0.002 b | 0.012 ± 0.002 b 0.004 ± 0.002 h | 0.01 ± 0.002 0 0.004 + 0.001 h | 0.08 + 0.00 a * | | _ | - | _ | |
| | 45 992 | 3502 ± 0.0020 | $3.004 \pm 0.002.0$ | 3.561 ± 0.0010 | 3.851 ± 0.01 a 3.851 ± 0.105 a | 1.82 ± 0.056 b | - 1 879 + 0 077 b | - 1 887 + 0 165 b | - 2 181 + 0 548 b | |
| | 149 137 | 0.041 ± 0.006 h | 0.024 ± 0.008 h | 0.022 ± 0.002 h | 0.063 ± 0.004 a | 1.02 - 0.000 0 | - | - | 2.101 ± 0.340 0 | |
| | 177.137 | 0.071 ± 0.0000 | $0.02 \pm 0.000 0$ | 0.022 ± 0.0020 | 0.005 ± 0.004 a | | - | - | - | 1 |

Supplementary Table S2. Volatile organic compounds (VOCs) emitted by Lysobacter type strains and measured using proton transfer reaction-time of flight-mass spectrometry (PTR-ToF-MS) analysis.

| H | 34.995 | $0.011 \pm 0.008 \text{ b}$ | $0.018 \pm 0.008 \text{ b}$ | 11.398 ± 3.371 a | $0.023 \pm 0.007 \text{ b}$ | $0.025 \pm 0.026 \text{ b}$ | 1.716 ± 3.631 b | 1.758 ± 2.588 b | $0.028 \pm 0.038 \text{ b}$ |
|------------------|----------|------------------------------|------------------------------|------------------------------|------------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|
| 2 ste | 48.053 | 1.777 ± 0.427 ab | 2.386 ± 0.253 a | 1.925 ± 0.227 a | 1.446 ± 0.221 abc | $0.495 \pm 0.042 \text{ c}$ | 0.448 ± 0.478 c | 0.713 ± 0.347 bc | $0.515 \pm 0.311 \text{ c}$ |
| Gu | 81.036 | 0.932 ± 0.063 ab | 0.851 ± 0.043 abc | 0.999 ± 0.048 a | $0.84 \pm 0.06 \text{ abc}$ | $0.75 \pm 0.051 \text{ bc}$ | $0.697 \pm 0.085 c$ | $0.696 \pm 0.109 \text{ c}$ | $0.64 \pm 0.056 c$ |
| 0 | 126.973 | $0.026 \pm 0.008 \text{ b}$ | $0.022 \pm 0.005 \text{ b}$ | 0.125 ± 0.015 a | $0.029 \pm 0.004 \text{ b}$ | $0.033 \pm 0.005 \text{ b}$ | $0.042 \pm 0.018 \ b$ | $0.061 \pm 0.015 \text{ b}$ | $0.053 \pm 0.016 \text{ b}$ |
| | 51.044 | $0.754 \pm 0.032 \text{ b}$ | $0.681 \pm 0.042 \text{ b}$ | $0.748 \pm 0.077 \text{ b}$ | $0.608 \pm 0.038 \text{ b}$ | 1.545 ± 0.034 a | 1.831 ± 0.456 a | 1.839 ± 0.266 a | 1.753 ± 0.185 a |
| | 38.034 | 16.683 ± 0.349 b | 16.546 ± 0.184 b | 17.102 ± 0.892 b | 16.555 ± 0.294 b | 30.777 ± 0.35 a | 29.219 ± 3.998 a | 31.01 ± 0.384 a | 31.713 ± 0.886 a |
| | 39.033 | $48.288 \pm 1.058 \text{ b}$ | 47.781 ± 0.359 b | 48.731 ± 2.21 b | 47.63 ± 1.055 b | 101.93 ± 1.087 a | 94.976 ± 17.224 a | 104.084 ± 2.246 a | 104.764 ± 2.061 a |
| | 102.097 | $0.026 \pm 0.005 \text{ d}$ | 0.051 ± 0.003 bcd | $0.035 \pm 0.002 \text{ cd}$ | $0.021 \pm 0.003 \text{ d}$ | 0.091 ± 0.013 ab | 0.112 ± 0.017 a | 0.118 ± 0.031 a | 0.079 ± 0.023 abc |
| $\tilde{\omega}$ | 115.114 | $0.161 \pm 0.017 \text{ b}$ | $0.093 \pm 0.009 \text{ b}$ | 0.335 ± 0.016 ab | $0.031 \pm 0.005 \text{ b}$ | 0.812 ± 0.047 a | 0.677 ± 0.248 a | 0.768 ± 0.328 a | $0.134 \pm 0.028 \text{ b}$ |
| ter | 116.116 | $0.014 \pm 0.003 \text{ c}$ | $0.011 \pm 0.003 \text{ c}$ | 0.027 ± 0.003 bc | $0.003 \pm 0.002 \text{ c}$ | 0.063 ± 0.016 a | 0.052 ± 0.003 ab | 0.065 ± 0.022 a | 0.011 ± 0.007 c * |
| Inst | 123.948 | $0.572 \pm 0.019 \text{ b}$ | $0.525 \pm 0.012 \text{ b}$ | $0.517 \pm 0.019 \text{ b}$ | $0.491 \pm 0.017 \text{ b}$ | 0.79 ± 0.073 a | 0.81 ± 0.046 a | 0.792 ± 0.032 a | 0.842 ± 0.05 a |
| Ū | 147.085 | $0.018 \pm 0.001 \text{ cd}$ | $0.017 \pm 0.003 \text{ cd}$ | 0.029 ± 0.002 bcd | $0.012 \pm 0.001 \text{ d}$ | 0.083 ± 0.009 ab | 0.093 ± 0.028 a | 0.105 ± 0.038 a | 0.073 ± 0.016 abc |
| | 157.165 | $0.074 \pm 0.018 \text{ bc}$ | $0.074 \pm 0.008 \text{ bc}$ | 0.241 ± 0.012 bc | $0.029 \pm 0.009 \text{ c}$ | 0.615 ± 0.047 a * | 0.333 ± 0.119 ab | 0.365 ± 0.206 ab | 0.079 ± 0.004 bc |
| | 158.160 | 0.013 ± 0.002 bc | 0.014 ± 0.003 bc | 0.037 ± 0.004 abc | 0.006 ± 0.001 c | 0.073 ± 0.006 a | 0.047 ± 0.015 ab | 0.049 ± 0.03 ab | 0.016 ± 0.004 bc |
| | 161.100 | 0.039 ± 0.004 b | $0.037 \pm 0.005 \text{ b}$ | 0.076 ± 0.004 ab | $0.029 \pm 0.003 \text{ b}$ | 0.087 ± 0.01 ab | 0.121 ± 0.045 a | 0.133 ± 0.043 a | 0.07 ± 0.013 ab |
| | 252.1681 | $0.005 \pm 0.002 \text{ c}$ | $0.011 \pm 0.003 \text{ c}$ | $0.013 \pm 0.001 \text{ c}$ | $0.014 \pm 0.001 \text{ bc}$ | 0.03 ± 0.005 a | 0.035 ± 0.005 a | 0.028 ± 0.004 ab | 0.03 ± 0.008 a |
| | 33.034 | 25.699 ± 2.406 abc | 23.228 ± 0.934 bc | 23.886 ± 1.368 bc | 21.03 ± 1.429 c | 26.261 ± 1.46 abc | 31.213 ± 1.31 a | 31.291 ± 2.834 a | 27.634 ± 2.437 ab |
| r 4 | 44.010 | 0.189 ± 0.022 b | 0.192 ± 0.015 b | $0.202 \pm 0.022 \text{ b}$ | $0.197 \pm 0.012 \text{ b}$ | 0.26 ± 0.006 ab | 0.336 ± 0.071 a | 0.245 ± 0.017 ab | 0.271 ± 0.059 ab |
| ste | 73.065 | 14.243 ± 1.079 c | 12.753 ± 1.788 c | 17.131 ± 1.065 c | 15.091 ± 1.186 c | 40.759 ± 3.528 bc | 48.153 ± 11.972 b | 42.231 ± 5.577 bc | 79.89 ± 21.757 a |
| Clu | 74.068 | 0.621 ± 0.043 de | 0.554 ± 0.066 e | 0.733 ± 0.033 cde | 0.668 ± 0.046 cde | 1.973 ± 0.133 bcd | 2.316 ± 0.598 b | 2.009 ± 0.274 bc | 3.731 ± 0.935 a |
| 0 | 143.148 | 0.052 ± 0.006 c | $0.05 \pm 0.008 \text{ c}$ | 0.154 ± 0.015 bc | 0.022 ± 0.001 c | 0.499 ± 0.051 a | 0.4 ± 0.085 ab | 0.406 ± 0.182 ab | 0.143 ± 0.07 bc |
| | 214.096 | $0.014 \pm 0.003 \text{ b}$ | 0.016 ± 0.003 ab | 0.021 ± 0.004 ab | 0.024 ± 0.001 a | $0.014 \pm 0.003 \text{ b}$ | 0.016 ± 0.003 ab | 0.021 ± 0.004 ab | 0.024 ± 0.001 a |
| | 68.060 | $0.392 \pm 0.02 \text{ b}$ | $0.369 \pm 0.03 \text{ b}$ | 0.497 ± 0.019 a | 0.561 ± 0.018 a | $0.392 \pm 0.02 \text{ b}$ | $0.369 \pm 0.03 \text{ b}$ | 0.497 ± 0.019 a | 0.561 ± 0.018 a |
| | 27.025 | 0.047 ± 0.005 b | 0.155 ± 0.008 ab | 0.07 ± 0.008 ab | 0.058 ± 0.011 b | 0.121 ± 0.008 ab | 0.198 ± 0.154 ab | 0.309 ± 0.072 a | 0.198 ± 0.091 ab |
| | 34.037 | 0.373 ± 0.031 ab | 0.337 ± 0.01 ab | 0.359 ± 0.022 ab | $0.31 \pm 0.026 \text{ b}$ | 0.36 ± 0.036 ab | 0.454 ± 0.034 a | 0.412 ± 0.068 ab | 0.394 ± 0.045 ab |
| | 44.058 | 0.416 ± 0.035 b | 1.808 ± 0.111 a | 0.759 ± 0.036 ab | 0.537 ± 0.066 ab | 0.795 ± 0.054 ab | 1.068 ± 0.883 ab | 1.72 ± 0.282 ab | 1.273 ± 0.548 ab |
| | 57.070 | 17.593 ± 0.508 b | 22.8 ± 0.816 ab | 21.695 ± 1.28 ab | 14.996 ± 1.781 b | 53.129 ± 3.529 ab | 33.799 ± 33.535 ab | 75.832 ± 21.326 a | 46.001 ± 21.77 ab |
| | 58.073 | $0.8 \pm 0.027 \text{ b}$ | 1.042 ± 0.061 ab | 1.001 ± 0.065 ab | $0.672 \pm 0.078 \text{ b}$ | 2.426 ± 0.116 ab | 1.583 ± 1.552 ab | 3.421 ± 0.974 a | 2.128 ± 0.972 ab |
| | 65.023 | 0.166 ± 0.012 b | 0.214 ± 0.013 ab | $0.188 \pm 0.005 \text{ ab}$ | $0.174 \pm 0.01 \text{ b}$ | 0.407 ± 0.068 ab | 0.466 ± 0.285 ab | 0.94 ± 0.445 a | 0.677 ± 0.342 ab |
| | 72.090 | 0.264 ± 0.04 b | 0.478 ± 0.029 ab | 0.395 ± 0.043 ab | 0.308 ± 0.047 ab | 0.436 ± 0.033 ab | 0.526 ± 0.447 ab | 1.035 ± 0.329 a * | 0.614 ± 0.193 ab |
| r 5 | 77.060 | 0.19 ± 0.047 c | 0.475 ± 0.112 abc | 0.457 ± 0.057 abc | 0.257 ± 0.037 bc | 0.525 ± 0.057 abc | 0.925 ± 0.247 a | 0.755 ± 0.143 ab | 0.575 ± 0.254 abc |
| ste | 101.097 | 0.126 ± 0.017 bc | 0.467 ± 0.041 abc | 0.287 ± 0.007 bc | $0.067 \pm 0.005 \text{ c}$ | 0.939 ± 0.096 ab | 0.838 ± 0.453 abc | 1.234 ± 0.323 a | 0.622 ± 0.363 abc |
| Clu | 118.073 | 0.188 ± 0.031 ab | 0.609 ± 0.242 a | 0.266 ± 0.058 ab | 0.085 ± 0.022 b | 0.188 ± 0.031 ab | 0.609 ± 0.242 a | 0.266 ± 0.058 ab | $0.085 \pm 0.022 \text{ b}$ |
| 0 | 127.968 | $0.004 \pm 0.001 \text{ b}$ | 0.006 ± 0.001 ab | 0.012 ± 0.002 a | 0.004 ± 0.003 ab | $0.004 \pm 0.001 \text{ b}$ | 0.006 ± 0.001 ab | 0.012 ± 0.002 a | 0.004 ± 0.003 ab |
| | 130.130 | 0.011 ± 0.003 ab | 0.008 ± 0.001 ab | 0.015 ± 0.002 a | 0.007 ± 0.001 b | 0.011 ± 0.003 ab | 0.008 ± 0.001 ab | 0.015 ± 0.002 a | $0.007 \pm 0.001 \text{ b}$ |
| | 131.054 | 0.012 ± 0.002 b | $0.01 \pm 0.006 \text{ b}$ | 0.036 ± 0.004 a | 0.008 ± 0.003 b | 0.012 ± 0.002 b | 0.01 ± 0.006 b | 0.036 ± 0.004 a | 0.008 ± 0.003 b |
| | 151.116 | 0.116 ± 0.01 a | 0.094 ± 0.003 ab | $0.044 \pm 0.003 \text{ b}$ | 0.069 ± 0.007 ab | 0.112 ± 0.007 a | 0.11 ± 0.038 a | 0.093 ± 0.009 ab | 0.108 ± 0.006 a |
| | 186.199 | 0.004 ± 0.002 bc | 0.007 ± 0.002 bc | 0.06 ± 0.01 a | 0.004 ± 0.002 c | 0.019 ± 0.002 bc | $0.025 \pm 0.007 \text{ b}$ | 0.021 ± 0.013 bc | 0.017 ± 0.005 bc |
| | 199.200 | 0.078 ± 0.018 a * | 0.02 ± 0.005 b | 0.072 ± 0.011 a | 0.023 ± 0.01 b | 0.043 ± 0.014 ab | 0.054 ± 0.017 ab | 0.038 ± 0.007 ab | 0.039 ± 0.011 ab |
| | 225.053 | 0.217 ± 0.015 b | 0.223 ± 0.023 b | 0.23 ± 0.056 b | 0.473 ± 0.121 a | 0.35 ± 0.019 ab | 0.345 ± 0.078 ab | 0.431 ± 0.038 ab | 0.34 ± 0.042 ab |
| | 226.053 | 0.045 ± 0.005 b | 0.043 ± 0.006 b | 0.043 ± 0.013 b | 0.096 ± 0.024 a | 0.076 ± 0.008 ab | 0.078 ± 0.022 ab | 0.09 ± 0.006 ab | 0.068 ± 0.017 ab |
| | 227.0403 | 0.051 ± 0.004 b | 0.05 ± 0.002 b | 0.052 ± 0.008 b | 0.105 ± 0.025 a | 0.051 ± 0.004 b | 0.05 ± 0.002 b | 0.052 ± 0.008 b | 0.105 ± 0.025 a |
| | 50.013 | 0.148 ± 0.089 a | 0.056 ± 0.026 a | 7.555 ± 0.585 a | 0.076 ± 0.012 a | 2.197 ± 0.285 a | 12.166 ± 26.192 a | 54.394 ± 44.594 a | 7.213 ± 13.255 a |
| | 51.007 | 0.472 ± 0.187 a | 0.221 ± 0.058 a | 20.17 ± 1.49 a | 0.27 ± 0.015 a | 4.726 ± 0.776 a | 28.648 ± 62.18 a | 95.337 ± 98.17 a | 16.76 ± 31.107 a |
| | 52.007 | 0.042 ± 0.006 a | 0.043 ± 0.004 a | 0.252 ± 0.022 a | 0.036 ± 0.006 a | 0.072 ± 0.035 a | 0.37 ± 0.774 a | 1.454 + 1.198 a | 0.211 + 0.373 a |
| r 6 | 53.002 | 0.081 ± 0.011 a | 0.124 ± 0.019 a | 0.185 ± 0.007 a | 0.088 ± 0.01 a | 0.085 ± 0.013 a | 0.167 ± 0.194 a | 0.556 ± 0.359 a * | 0.135 ± 0.123 a |
| ste | 59.049 | 247.689 ± 60.016 a | 649.633 ± 160.071 a | 587.059 ± 52.154 a | 341.874 ± 56.624 a | 342.242 ± 27.044 a | 720.223 ± 360.911 a * | 482.783 ± 113.029 a | 344.305 ± 144.231 a |
| Clu | 60.053 | 8.366 ± 1.986 a | 21.807 ± 5.327 a | 19.649 ± 1.712 a | 11.536 ± 1.791 a | 12.153 ± 1.04 a | 24.689 ± 14.167 a | 17.055 ± 3.959 a | 12.132 ± 4.928 a |
| G | 63.008 | 3.743 ± 0.347 a | 3.223 ± 0.235 a | 3 ± 0.066 a | 3.057 ± 0.132 a | 4.017 ± 0.162 a | 3.028 ± 2.575 a | 3.353 ± 0.459 a | 5.368 ± 1.109 a * |
| | 63.026 | 1.086 ± 0.083 a | 2.068 ± 0.08 a | 2.152 ± 0.093 a | 1.695 ± 0.038 a | 2.169 ± 0.041 a | 4.626 ± 4.577 a | 15.604 ± 12.076 a | 8.598 ± 9.076 a |
| | 64.023 | 0.018 ± 0.011 a | 0.052 ± 0.011 a | 0.098 ± 0.007 a | 0.046 ± 0.009 a | 0.077 ± 0.021 a | 0.159 ± 0.16 a | 0.505 ± 0.422 a | 0.286 ± 0.296 a |

| 65.060 | 2.232 ± 0.515 a | 2.922 ± 0.342 a | 2.413 ± 0.38 a | 1.808 ± 0.285 a | 1.144 ± 0.14 a | 0.987 ± 1.109 a | 1.463 ± 0.747 a | 1.149 ± 0.745 a |
|---------|---------------------|----------------------|-------------------------|----------------------|------------------------|-------------------------|-------------------------|----------------------------|
| 67.022 | 0.013 ± 0.006 a | 0.009 ± 0.005 a | 0.108 ± 0.012 a | 0.013 ± 0.007 a | 0.034 ± 0.004 a | 0.193 ± 0.392 a | 0.821 ± 0.665 a | 0.112 ± 0.191 a |
| 71.086 | 4.792 ± 0.747 a | 8.483 ± 0.524 a | 6.946 ± 0.773 a | 5.426 ± 0.709 a | 7.7 ± 0.561 a | 9.334 ± 8.077 a | 17.805 ± 6.239 a | 11.087 ± 3.709 a |
| 73.250 | 0.069 ± 0.005 a | 0.077 ± 0.032 a | 0.077 ± 0.012 a | 0.147 ± 0.024 a | 0.069 ± 0.005 a | 0.077 ± 0.032 a | 0.077 ± 0.012 a | 0.147 ± 0.024 a |
| 77.998 | 0.022 ± 0.008 a | 0.022 ± 0.006 a | 0.084 ± 0.04 a | 0.017 ± 0.004 a | 0.022 ± 0.008 a | 0.022 ± 0.006 a | 0.084 ± 0.04 a | 0.017 ± 0.004 a |
| 80.986 | 0.04 ± 0.022 a | 0.156 ± 0.073 a | 0.185 ± 0.045 a | 0.075 ± 0.041 a | 0.04 ± 0.022 a | 0.156 ± 0.073 a | 0.185 ± 0.045 a | 0.075 ± 0.041 a |
| 88.053 | 0.624 ± 0.232 a | 0.516 ± 0.11 a | 0.704 ± 0.133 a | 0.232 ± 0.068 a | 0.196 ± 0.029 a | 0.257 ± 0.22 a | 0.445 ± 0.137 a | 0.246 ± 0.094 a |
| 89.061 | 1.194 ± 0.14 a | 2.271 ± 0.193 a | 1.073 ± 0.307 a | 0.785 ± 0.051 a | 0.913 ± 0.512 a | 1.259 ± 1.154 a | 0.805 ± 0.382 a | 0.687 ± 0.499 a |
| 90.064 | 0.068 ± 0.007 a | 0.114 ± 0.009 a | 0.06 ± 0.018 a | 0.042 ± 0.005 a | 0.054 ± 0.022 a | 0.088 ± 0.063 a | 0.06 ± 0.015 a | 0.034 ± 0.024 a |
| 91.024 | 0.093 ± 0.036 a | 0.185 ± 0.044 a | 4.541 ± 0.58 a | 0.053 ± 0.005 a | 0.938 ± 0.071 a | 3.365 ± 7.038 a | 9.482 ± 7.199 a | 0.18 ± 0.276 a |
| 92.028 | 0.007 ± 0.005 a | 0.009 ± 0.004 a | 0.2 ± 0.023 a | 0.005 ± 0.002 a | 0.043 ± 0.009 a | 0.157 ± 0.303 a | 0.423 ± 0.318 a | $0.01 \pm 0.017 \text{ a}$ |
| 101.027 | 0.23 ± 0.014 a | 0.242 ± 0.001 a | 0.242 ± 0.003 a | 0.215 ± 0.009 a | 0.233 ± 0.029 a | 0.268 ± 0.077 a | 0.348 ± 0.086 a | 0.222 ± 0.031 a |
| 103.078 | 1.86 ± 0.318 a | 0.828 ± 0.123 a | 0.986 ± 0.171 a | 0.439 ± 0.187 a | 2.651 ± 0.286 a | 1.467 ± 1.535 a | 2.661 ± 1.752 a | 10.323 ± 19.343 a |
| 104.074 | 0.141 ± 0.02 a | 0.084 ± 0.01 a | 0.088 ± 0.013 a | 0.065 ± 0.023 a | 0.205 ± 0.026 a | 0.149 ± 0.095 a | 0.218 ± 0.083 a | 0.689 ± 1.173 a |
| 115.042 | 0.096 ± 0.006 a | 0.087 ± 0.007 a | 0.08 ± 0.005 a | 0.073 ± 0.009 a | 0.081 ± 0.018 a | 0.082 ± 0.009 a | 0.107 ± 0.013 a | 0.08 ± 0.021 a |
| 115.078 | 0.364 ± 0.089 a | 0.141 ± 0.02 a | 0.249 ± 0.038 a | 0.146 ± 0.021 a | 0.241 ± 0.032 a | 0.218 ± 0.167 a | 0.376 ± 0.181 a | 0.299 ± 0.212 a |
| 117.009 | 0.103 ± 0.02 a | 0.114 ± 0.049 a | 0.111 ± 0.022 a | 0.039 ± 0.011 a | 0.103 ± 0.02 a | 0.114 ± 0.049 a | 0.111 ± 0.022 a | 0.039 ± 0.011 a |
| 117.094 | 30.884 ± 10.018 a | 26.295 ± 4.444 a | 33.94 ± 5.697 a | 10.929 ± 3.005 a | 21.92 ± 1.579 a | 26.245 ± 36.166 a | 57.939 ± 26.697 a | 25.151 ± 12.459 a |
| 117.327 | 0.068 ± 0.027 a | 0.059 ± 0.013 a | 0.071 ± 0.013 a | 0.025 ± 0.007 a | 0.068 ± 0.027 a | 0.059 ± 0.013 a | 0.071 ± 0.013 a | 0.025 ± 0.007 a |
| 118.098 | 2.093 ± 0.693 a | 1.8 ± 0.324 a | 2.297 ± 0.377 a | 0.754 ± 0.21 a | 1.485 ± 0.126 a | 1.825 ± 2.527 a | 3.863 ± 1.768 a | 1.711 ± 0.865 a |
| 119.095 | 0.281 ± 0.065 a | 0.275 ± 0.032 a | 0.286 ± 0.04 a | 0.156 ± 0.026 a | 0.267 ± 0.032 a | 0.293 ± 0.215 a | 0.488 ± 0.146 a | 0.42 ± 0.342 a |
| 120.052 | 0.018 ± 0.004 a | 0.013 ± 0.001 a | 0.03 ± 0.005 a | 0.011 ± 0.002 a | 0.011 ± 0.007 a | 0.02 ± 0.005 a | 0.046 ± 0.032 a | 0.308 ± 0.599 a |
| 125.635 | 0 ± 0 a | 0.009 ± 0.004 a | 0.008 ± 0.001 a | 0.002 ± 0.003 a | 0 ± 0 a | 0.009 ± 0.004 a | 0.008 ± 0.001 a | 0.002 ± 0.003 a |
| 130.040 | 0.009 ± 0.001 a | 0.008 ± 0.002 a | 0.042 ± 0.004 a | 0.011 ± 0.002 a | 0.02 ± 0.003 a | 0.091 ± 0.125 a | 0.237 ± 0.149 a | 0.035 ± 0.004 a |
| 131.034 | 0.005 ± 0.003 a | 0.006 ± 0.001 a | 0.019 ± 0.004 a | 0.006 ± 0.002 a | 0.014 ± 0.006 a | 0.074 ± 0.108 a | 0.159 ± 0.101 a | 0.041 ± 0.033 a |
| 133.072 | 0.038 ± 0.005 a | 0.051 ± 0.006 a | 0.842 ± 0.065 a | 0.036 ± 0.004 a | 0.736 ± 0.093 a | 1.016 ± 2.049 a | 5.284 ± 3.964 a | 2.865 ± 5.569 a |
| 134.073 | 0.01 ± 0.003 a | 0.01 ± 0.002 a | 0.075 ± 0.012 a | 0.006 ± 0.002 a | 0.073 ± 0.008 a | 0.103 ± 0.161 a | 0.435 ± 0.328 a | 0.241 ± 0.454 a |
| 149.055 | 0.045 ± 0.005 a | 0.07 ± 0.011 a | 0.074 ± 0.021 a | 0.066 ± 0.015 a | 0.045 ± 0.005 a | 0.07 ± 0.011 a | 0.074 ± 0.021 a | 0.066 ± 0.015 a |
| 167.118 | 0.101 ± 0.009 a | 0.07 ± 0.005 a | 0.16 ± 0.014 a | 0.039 ± 0.007 a | 0.07 ± 0.015 a | 0.109 ± 0.072 a | 0.166 ± 0.117 a | 0.029 ± 0.007 a |
| 169.043 | 0.08 ± 0.014 a | 0.093 ± 0.008 a | 0.106 ± 0.012 a | 0.099 ± 0.009 a | 0.12 ± 0.017 a | 0.139 ± 0.049 a | 0.137 ± 0.025 a | 0.111 ± 0.015 a |
| 186.917 | 0.037 ± 0.008 a | 0.031 ± 0.006 a | 0.036 ± 0.009 a | 0.047 ± 0.004 a | 0.037 ± 0.008 a | 0.031 ± 0.006 a | 0.036 ± 0.009 a | 0.047 ± 0.004 a |
| 223.071 | 0.306 ± 0.047 a | 0.293 ± 0.047 a | 0.3 ± 0.088 a | 0.606 ± 0.126 a | 0.432 ± 0.03 a | 0.495 ± 0.206 a | 0.526 ± 0.104 a | 0.417 ± 0.074 a |
| 224.072 | 0.065 ± 0.013 a | 0.068 ± 0.012 a | 0.064 ± 0.021 a | 0.136 ± 0.03 a | 0.1 ± 0.013 a | 0.118 ± 0.063 a | 0.123 ± 0.034 a | 0.101 ± 0.02 a |
| 43.054 | 11.958 ± 1.197 a | 52.83 ± 3.456 a * | 21.216 ± 0.946 a | 15.492 ± 1.735 a | 21.484 ± 1 a | 32.341 ± 27.711 a | 51.222 ± 10.703 a | 35.167 ± 16.705 a |
| 44.998 | 30.733 ± 2.431 a | 26.507 ± 1.446 a | 24.363 ± 0.928 a | 25.237 ± 1.081 a | 49.036 ± 1.001 a | 35.609 ± 29.4 a | 47.339 ± 8.195 a | 63.426 ± 13.638 a |
| 46.995 | 0.38 ± 0.078 a | 0.366 ± 0.04 a | 9.6 ± 0.509 a * | 0.315 ± 0.022 a | 0.928 ± 0.185 a | 4.572 ± 9.753 a | 21.468 ± 17.341 a | 3.093 ± 5.477 a |
| 48.004 | 0.724 ± 0.139 a | 0.597 ± 0.051 a | 15.599 ± 0.745 a * | 0.66 ± 0.024 a | 1.706 ± 0.216 a | 7.359 ± 15.108 a | 32.77 ± 26.3 a | 4.654 ± 7.948 a |
| 49.011 | 8.302 ± 4.011 a | 2.959 ± 1.203 a | 314.259 ± 129.406 a | 4.045 ± 0.163 a | 101.948 ± 17.028 a | 634.38 ± 1387.185 a | 2747.29 ± 2248.65 a | 364.65 ± 683.445 a |
| 142.936 | 0.009 ± 0.002 a | 0.003 ± 0.001 a | 0.031 ± 0.004 a | 0.031 ± 0.002 a | 0.007 ± 0.006 a | 0.019 ± 0.024 a | 0.039 ± 0.02 a | 0.014 ± 0.007 a |

Headspace VOC analysis was carried out by proton transfer reaction-time of flight-mass spectrometry (PTR-ToF-MS) analysis for *Lysobacter antibioticus* DSM 2044^T, *L. capsici* DSM 19286^T, *L. enzymogenes* DSM 2043^T and *L. gummosus* DSM 6980^T grown for ten days on nutrient agar (NA) or potato dextrose agar (PDA). VOCs were identified by the mass/charge ratio (*m/z*). Mean \pm standard deviation values of the headspace VOC concentration (expressed as ppbv) from four to five replicates are reported for each compound, *Lysobacter* type strain and growth media. For each compound, different letters indicate significant differences according to Kruskal-Wallis test with Bonferroni correction ($p \le 0.05$). Only compounds with a background-corrected headspace concentration significantly emitted (Kruskal-Wallis with Bonferroni correction, $p \le 0.05$) in *Lysobacter*-inoculated headspace vials as compared with uninoculated vials for at least one strain and growth medium are reported.

VOCs were grouped based on their emission profiles in: VOCs with higher emission by all *Lysobacter* type strains on PDA as compared with NA (Cluster 1), VOCs with higher emission by some *Lysobacter* type strains on NA as compared with PDA (Cluster 2), VOCs with higher emission by all *Lysobacter* type strains on NA as compared with PDA (Cluster 3), VOCs with higher emission by all *Lysobacter* type strains on NA as compared with PDA (Cluster 4), VOCs with different (Cluster 5) or consistent (Cluster 6) emission by *Lysobacter* type strains on both growth media. Asterisks in bold (*) indicate VOC emission values associated to non-zero coefficients of the least absolute shrinkage and selection operator (LASSO) model calculated for the growth media prediction (column LASSO) and *Lysobacter* type strain prediction on PDA (columns Potato dextrose agar) and NA (columns Nutrient agar). Non-zero coefficients specify VOCs that can be used to predict the growth media and the *Lysobacter* type strain, respectively.