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GC/c-IRMS for improving the detection of authenticity of grape must

Journal:	<i>Journal of Agricultural and Food Chemistry</i>
Manuscript ID	jf-2019-059525.R3
Manuscript Type:	Article
Date Submitted by the Author:	n/a
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1 **GC/c-IRMS for improving the detection of authenticity of grape must**

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27 Abstract

28 Since ancient times, grape must and wine have been considered one of the most sophisticated
29 matrices and, in the last years, the continuous rise in volumes and prices of grapes and wine
30 has encouraged frauds and adulterations in the oenological field. One of the most common
31 adulteration is the sugar addition to grape must in the form of cane, beet sugar, or syrup coming
32 from vegetable sources like cereals or fruits. Since 1990, the International Organisation of Vine
33 and Wine (OIV) has issued specific official isotopic methods to fight against this practice, but
34 they are not always effective. With the aim to develop a new method able to identify the sugar
35 addition, we compared the $\delta^{13}\text{C}$ value of sugar extracted from grape must and analysed by EA-
36 IRMS to the $\delta^{13}\text{C}$ value of proline analysed by GC-c-IRMS, after extraction and derivatization.
37 $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ of proline have also been tested as potential geographical markers.

38 In addition, the carbon isotopic composition of two characteristic grapes must sugars (myo and
39 scyllo – inositol) was measured by GC-c-IRMS after derivatization, in order to identify the
40 illegal correction of their concentration.

41 On the basis of the obtained results, we can conclude that the compound specific isotope
42 analysis represents a novel analytical tool to support and improve certification and control
43 procedures.

44

45 **Keywords:** GC/c-IRMS, stable isotope analysis, proline, myoinositol and scylloinositol,
46 chaptalization, grape must

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48 **Declarations of interest:** none.

49 This research did not receive any specific grants from funding agencies in the public,
50 commercial, or non-profit sectors.

51 1. Introduction

52 According to the International Organisation of Vine and Wine (OIV) definition, wine is the
53 beverage resulting exclusively from the partial or complete alcoholic fermentation of fresh
54 grapes, whether crushed or not, or of grape must that is the liquid product obtained naturally or
55 by physical processes from fresh grapes.¹ OIV has established for wine a unique minimum
56 alcoholic strength of 8.5 % vol., with the flexibility to be reduced to 7 % vol to guarantee its
57 stability.² To react this limit the addition of sources of sugar to grape must before fermentation
58 is allowed. While the addition of rectified must to grape must or wine before or during
59 fermentation to increase the alcoholic strength is always permitted in all the countries, the use
60 of beet sugar and cane sugar is legal only for specific winegrowing regions and vintages e.g.
61 in Brazil, Canada, Chile, China, France, Germany, Japan, New Zealand, Switzerland, United
62 Kingdom, United States.³ In Italy the addition of exogenous sugar (beet or cane) is forbidden
63 and it constitutes a fraud (chaptalisation) by unscrupulous wine producers to increase profit.⁴
64 In 1990 the European Commission (EC) adopted isotopic methods as the first official analytical
65 methods (now OIV MA-AS-311-05, MA-AS-312-06, and MA-AS2-12) to detect and combat
66 these types of grape must and wine frauds.⁵ They are based on the analysis of the isotopic ratios
67 of hydrogen (D/H) and carbon (¹³C/¹²C, expressed as $\delta^{13}\text{C}$) in ethanol distilled from wine and
68 after must fermentation. Unfortunately, during the last decades, adulterations became
69 increasingly sophisticated,⁶ thus the development of even more powerful analytical methods
70 for must and wine authentication is a great challenge.

71 The direct stable isotope ratio analysis of single chemical compounds, normally measured by
72 a GC-c-IRMS technique, provide a means of obtaining a more in-depth understanding in
73 respect to the traditionally analysis of bulk products.⁷ Examples of application are reported for
74 single amino acids in wheat and durum wheat samples, to discriminate between organically
75 and conventionally agricultural practices,⁸ vanillin, to discriminate between natural and

76 synthetic ones,⁹ fatty acid, to identify the adulteration of high value oils with cheaper ones,⁹
77 wine volatile compounds, to reassess the water status in vineyards.¹⁰ In relation to wine, Spitzke
78 at al. developed a GC-c-IRMS method to analyse $^{13}\text{C}/^{12}\text{C}$ of higher alcohols, for example, 2-
79 methylpropan-1-ol, 2- and 3-methylbutan-1-ol; butan-2,3-diol, 2-phenyl-1-ethanol, and
80 glycerol.¹¹ By correlating the $\delta^{13}\text{C}$ of higher alcohol compounds (such as 2-/3-methylbutan-1-
81 ol) with that of wine ethanol ($R^2 = 0.829$), they were able to improve the detection of
82 chaptalisation. Other groups,^{12,13} have also investigated the isotopic composition of ethanol and
83 glycerol by GC-c-IRMS as alternative techniques to determine adulteration of wines.

84 This paper illustrates two studies that propose new specific compound methods to improve the
85 identification of grape must authenticity.

86 In the first one we compared, for the first time, the $\delta^{13}\text{C}$ of the sugar fraction extracted from
87 grape must using the official method UNI ENV 12140:1997, with the $\delta^{13}\text{C}$ of proline after
88 extraction and derivatization.¹⁴ Among the amino acids the most abundant in wine and grapes
89 is the proline with a content ranging between 30 and 85%,¹⁵ and this makes it possible to obtain
90 a sufficient quantity of it for derivatization and analysis. Moreover, yeast does not require
91 proline as a nitrogen source and it is therefore maintained in wine. The isotopic ratio of nitrogen
92 ($^{15}\text{N}/^{14}\text{N}$, expressed as $\delta^{15}\text{N}$) was also analysed using proline, as additional marker to trace the
93 geographical origin of grapes. As reported by Paolini et al.⁸ the isotopic ratio of nitrogen could
94 be a useful additional marker, because differently from H, O and C, nitrogen in grape
95 compounds derives directly from the soil, and therefore the factors affecting its isotopic
96 variability are different from those affecting the other 3 isotopic ratios.¹⁶

97 In the second study we tested two characteristic grape must sugars (myo and scyllo - inositol)
98 after derivatization, to see if the analysis of their $\delta^{13}\text{C}$ values could be useful in identifying the
99 illegal correction of their concentration in concentrated and rectified grape must (CRM).

100 These polyalcohols originate in the grape berry and their quantification has been already

101 proposed by Monetti et al. to control the authenticity of the CRM, because they are not retained
102 by the resins used for concentration process and they are not present in other purified
103 commercial sugars.¹⁷ A minimum content of myo-inositol (750 mg/kg sugar) and a myo- and
104 scyllo-inositol ratio of 20 or less have been suggested as authenticity indexes.¹⁷ In Regulation
105 479/2008 this proposal was only partially adopted and today official CRM controls focus on
106 the presence of myo-inositol alone, without considering its commercial availability, its levels
107 in musts, and the relationship between the two isomers.¹⁸ There is the suspect that fraudsters
108 correct the concentration of these two polyalcohols, in particular myo-inositol, by adding
109 commercial pure myo- and scyllo-inositol to fake grape must concentrate originated from other
110 fruits (e.g. date or tapioca) or from a mix of sugars. The $\delta^{13}\text{C}$ variability of authentic and fake
111 polyalcohol have been explored in this study and tested on samples coming from the market, in
112 order to verify their validity as fraud detectors.

113 **2. Materials and methods**

114 ***2.1. Reagents and solutions***

115 Proline $\geq 98\%$, myo-Inositol $\geq 99\%$, scyllo-Inositol $\geq 98\%$ and cation-exchange resin
116 (Amberlite® IR120) were purchased from Sigma Aldrich. All other solvents (acetone,
117 dichloromethane, ethanol, ethyl acetate and isopropanol) and reagents (acetic anhydride,
118 silylating agent HMDS+TMCS+Pyridine 3:1:9, and triethylamine) used were of analytical
119 grade and purchased from Sigma Aldrich and VWR.

120

121 ***2.2. Sampling***

122 ***2.2.1 Italian grape must, first study***

123 59 authentic Italian grape musts were sampled during 2016 (N=36) and 2017 (N=23) harvests
124 (Table 1). One to five samples were collected from 15 different Italian regions (Abruzzo,
125 Campania, Emilia Romagna, Friuli Venezia Giulia, Lazio, Liguria, Lombardy, Marche,

126 Piedmont, Puglia, Sardinia, Sicily, Tuscany, Trentino Alto Adige, Umbria and Veneto), at the
127 usual technological harvest time (early, medium and late harvesting) (Supplementary Figure
128 1S). 25 different grape varieties were considered in order to describe natural variability (Table
129 1). The sampling was supervised by the technicians of the Edmund Mach Foundation (San
130 Michele all'Adige - Italy), who personally followed all the harvesting and grape crushing stages
131 for must production (as part of the Italian Project 'Climaitalia2020'). Proline was extracted
132 from all the samples and derivatized and subjected to the analysis of $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$.

133 ***2.2.2 Rectified concentrated grape must, second study***

134 10 authentic rectified concentrated grape musts of the 2018 harvest were collected (Table 2).
135 In addition, 7 commercial CRM samples from Italy and Spain were also considered.

136 ***2.3 EA-IRMS analysis***

137 *2.3.1 Extraction of sugar fraction,*

138 The sugar fraction was extracted using the official method UNI ENV 12140. In brief, the solid
139 fraction of the must (50 mL) was removed by centrifugation at 1400 g for 10 min. 2 g of
140 powdered calcium hydroxide were added to the supernatant liquid and the solution was heated
141 in a bath at 90°C for 3 min. The precipitate was separated by centrifugation of the hot solution
142 (3 min at 1400 g) and the supernatant liquid was acidified with 0.1 mol/L sulfuric acid in order
143 to obtain a pH value of approximately 5. After a night in the refrigerator (4°C), the supernatant
144 liquid was freeze-dried to obtain the sugar fraction.

145 *2.3.2 EA-IRMS analysis of $\delta^{13}\text{C}$*

146 The $\delta^{13}\text{C}$ value of sugar fraction samples was measured using an elemental analyser (Flash EA
147 1112, Thermo Scientific, Bremen, Germany), coupled with an IRMS (DELTA V, Thermo
148 Scientific) through a ConFlo IV dilutor device (Thermo Finningan, Bremen, Germany).

149 **2.4 GC-c-IRMS analysis**

150 *2.4.1 Isolation and derivatization of proline*

151 The grape must sample was adjusted to pH 2.3 with HCl 0.01 M and 100 μ l of a norleucine
152 solution (8 **mg/mL** in 0.1M HCl) was added as internal standard. Proline was isolated using an
153 Amberlite IR120 cation-exchange resin, previously saturated with H⁺ at all exchange sites as
154 reported by Takano et al.¹⁹ 5 mL of grape must sample were passed through the resin column
155 and washed with water. Finally, proline was eluted with NH₄OH (10% w/w) and then dried
156 under N₂.²⁰

157 Proline was analysed after *N*-acetylisopropyl derivatization, following the method reported by
158 Corr et al.¹⁴ Briefly, proline was esterified with 1 mL of acetyl chloride:isopropanol mixture
159 (1:4 v/v) at 100 °C for 1h and then acylated with 1 mL of acetic anhydride-triethylamine-
160 acetone mixture (1:2:5 v/v) at 60°C for 10 min. The reagents were evaporated under a gentle
161 stream of nitrogen at room temperature, 1 mL of saturated sodium chloride-water solution and
162 1 mL of ethyl acetate were added and then mixed vigorously. The organic layer containing the
163 derivatized proline was dried under nitrogen, residual water was removed with
164 dichloromethane and finally dissolved in ethyl acetate (200 mL).

165 *2.4.2 Derivatization and quantification of myo and scyllo inositol*

166 Derivatization and quantification of myo- and scyllo-inositol in grape must were performed
167 following the official method RESOLUTION OIV-OENO 419C-2015. Briefly, 5 g of grape
168 must were weighted in a 50 mL volumetric flask, adding 1 mL of xylitol standard solution
169 (10000 mg/L in water) and then brought to volume with water. 100 μ l of the final solution were
170 dried under a gentle stream of nitrogen and 400 μ l of the derivatization mixture (HMDS +
171 TMCS + Pyridine, 3:1:9) were added. The vial was closed and placed in the oven at 80°C for
172 60 minutes.

173 Myo- and scyllo-inositol were quantified using an Agilent Intuvo 9000 GC-FID system,
174 injecting 3 μl in split mode (1:10) into a 30 m HP-5MS Ultra Inert (0.32 mm I.D. \times 0.25 μm
175 film thickness; Agilent) with H_2 as carrier gas (2 mL/min). The oven temperature was
176 programmed starting at 100°C, raised to 240°C by 10°C/min, then raised to 260°C by 2°C/min,
177 and finally raised to 310°C by 50°C/min and held at this temperature for 5 minutes. The injector
178 temperature was set at 270°C.

179 2.4.3 GC-c-IRMS analysis of proline

180 The $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ values of proline were determined using a Trace GC Ultra (GC IsoLink +
181 ConFlo IV, Thermo Scientific) interfaced with an IRMS (DELTA V, Thermo Scientific) and
182 with a single-quadrupole MS detector (ISQ Thermo Scientific). 0.5 μL of each sample was
183 injected in splitless mode and a 60 m HP-INNOWAX capillary column (0.32 mm I.D. \times 0.25
184 μm film thickness; Agilent) was used with He as carrier gas (1.4 mL/min). The injector was at
185 250 °C and the injector was at 250 °C and the oven temperature program was set as follows:
186 initial 40 °C held for 2 min, ramped to 140 °C at 40 °C/min, followed by ramped to 180 °C at
187 2.5 °C/min for, then ramped to 220 °C at 6 °C/min and finally ramped to 250 °C at 40 °C/min
188 for 15 min.

189 The eluted proline was combusted into N_2 and CO_2 in a combustion furnace reactor operated
190 at 1030°C. During $\delta^{15}\text{N}$ analysis, a liquid nitrogen trap was added after the reactor to block the
191 CO_2 .

192 In order to monitor derivatization step and instrumental performance, a standard proline was
193 derivatized and the $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ values were measured using GC-c-IRMS before and after
194 each analytical run and compared with the isotopic composition measured directly by EA-
195 IRMS ($\delta^{13}\text{C} = -24.5\text{‰}$ and $\delta^{15}\text{N} = +1.1\text{‰}$) without any derivatization step.

196 2.4.4 GC-c-IRMS analysis of myo and scyllo inositol

197 The $\delta^{13}\text{C}$ values of myo and scyllo inositol were determined injecting 2 μl in splitless mode in
198 a 30 m HP5-MS capillary column (0.32 mm I.D. \times 1.00 μm film thickness; Agilent) with He
199 as carrier gas at 1.5 mL/min. The injector was at 250 $^{\circ}\text{C}$ and the oven temperature program
200 used is as follows: held for 20 min at 150 $^{\circ}\text{C}$, increased to 220 $^{\circ}\text{C}$ at 10 $^{\circ}\text{C}/\text{min}$, and finally
201 ramped to 320 $^{\circ}\text{C}$ at 40 $^{\circ}\text{C}/\text{min}$ and held for 10 min..

202 In order to monitor the derivatization step and instrumental performance, a standard mix of
203 myo and scyllo-inositol with know carbon isotopic composition (-37.2‰ and -36.9‰
204 respectively) was derivatized and the $\delta^{13}\text{C}$ values were measured using GC-C-IRMS before
205 and after each analytical run.

206 2.5 Data expression

207 All the $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ values are reported in relation to the known isotopic composition of the
208 reference CO_2 and N_2 gasses introduced into the ion source of the IRMS at the beginning and
209 end of each EA and GC run. All samples were measured three times, and the isotope ratio was
210 expressed in ‰ versus V-PDB (Vienna - Pee Dee Belemnite) for $\delta^{13}\text{C}$ and atmospheric
211 nitrogen for $\delta^{15}\text{N}$ according to equation 1.

$$212 \quad \delta = \left[\frac{(R_s - R_{std})}{R_{std}} \right] (1)$$

213 where R_s is the isotope ratio of the sample and R_{std} is the isotope ratio of the internationally
214 accepted standard.

215 The $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ values of pure non-derivatized proline were determined by EA-IRMS. The
216 isotopic values $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ were calculated against two working in-house standards
217 (caseins), the first one with $\delta^{13}\text{C}=-21.98\text{‰}$ and $\delta^{15}\text{N}=7.38\text{‰}$ while the second one with
218 $\delta^{13}\text{C}=-30.60\text{‰}$ and $\delta^{15}\text{N}=-3.40\text{‰}$. They have themselves been calibrated against international
219 reference materials: fuel oil NBS-22 with $\delta^{13}\text{C}=-30.03\text{‰}$, sucrose IAEA-CH-6 with

220 $\delta^{13}\text{C}=-10.45\text{‰}$ (IAEA-International Atomic Energy Agency, Vienna, Austria), and L-
221 glutamic acid USGS 40 with $\delta^{13}\text{C}=-26.39\text{‰}$ and $\delta^{15}\text{N}=-4.52\text{‰}$ (U.S. Geological Survey,
222 Reston, VA, USA) for $^{13}\text{C}/^{12}\text{C}$ and $^{15}\text{N}/^{14}\text{N}$ and potassium nitrate IAEA- NO_3 ($\delta^{15}\text{N}=+4.7\text{‰}$)
223 from IAEA for $^{15}\text{N}/^{14}\text{N}$.

224 The $\delta^{15}\text{N}$ and $\delta^{13}\text{C}$ values of proline in grape must were calculated against the standard proline,
225 analysed before and after each sample. The instrumental data were corrected on the basis of
226 the difference between the $\delta^{15}\text{N}$ and $\delta^{13}\text{C}$ values of the standard proline in GC-c-IRMS (mean
227 of six results, three before and three after the samples) and in EA-IRMS, which was in any case
228 always lower than 0.5‰ and $\pm 1.6\text{‰}$ for $\delta^{15}\text{N}$ and $\delta^{13}\text{C}$, respectively. Likewise, the $\delta^{13}\text{C}$ value
229 of myo and scyllo inositol in must was calculated against the mixture of standard myo and
230 scyllo inositol, analysed before and after each sample.

231 The effective $\delta^{13}\text{C}$ values of proline, myo and scyllo inositol were obtained applying an
232 empirical correction in order to remove the contribution of the derivatization reagents.
233 Correction factor was calculated by determining the $\delta^{13}\text{C}$ value of the underivatized standard
234 (EA-IRMS) and the $\delta^{13}\text{C}$ value of the derivatized standard (GC-c-IRMS):

$$235 \quad n_{cd}\delta^{13}C_{cd} = n_c\delta^{13}C_c + n_d\delta^{13}C_d \quad (2)$$

236 where n is the number of moles of carbon, and the subscripts c , d , and cd represent the
237 compound of interest, the derivative group, and the derivatized compound, respectively.

238

239 **2.6 Repeatability limit and uncertainty**

240 One sample of grape must and one of grape CRM were treated and derivatized one time a
241 month for one year to calculate the within-laboratory reproducibility standard deviation (SR)
242 of $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ analysis of proline, myo and scyllo inositol. The analytical uncertainty (U) of
243 the $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ of proline, myo and scyllo inositol analysis, expressed as the coverage factor

244 $k=2$ multiplied for the SR (within-laboratory reproducibility standard deviation), was 0.4%,
245 whereas the reproducibility limit expressed as $k \cdot \text{rad } 2 \cdot \text{SR}$ (with $k=2$) was 0.6%.²¹

246 To determine the repeatability limit for $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ of proline, 10 replicates of a grape must
247 sample were derivatized, and each of the them were analysed using GC-c-IRMS. For the $\delta^{13}\text{C}$
248 of myo and scyllo-inositol a sample of grape CRM was considered. The standard deviation
249 obtained (1σ) was 0.5‰ for $\delta^{13}\text{C}$ of proline and 0.2‰ for all other parameters (Supplementary
250 Table 1S).

251 **2.7 Statistical analysis**

252 The data were analysed with the Statistica software 13.1 (StatSoft Inc., Tulsa, OK, USA).
253 Statistically significant correlations were verified using the Pearson correlation test.
254 Statistically significant differences were observed using a Tukey HSD test. In all the statistical
255 analysis, the cutoff value was set at $p < 0,05$, which is associated with a significant difference
256 between groups of values.

257 **3. Results and discussion**

258

259 **Study 1**

260 In plants, the sugar fraction (mainly glucose and fructose) is the result of photosynthesis, that
261 takes place in the green plastids of plant cells using carbon dioxide and water as precursors.
262 Sugars, through glycolysis and the Krebs cycle, are used by the plants to synthesise 2-
263 oxoglutarate that thanks to the action of glutamate synthase is converted in glutamate.²² Proline
264 comes from glutamate, which is converted to proline by two consecutive reduction steps
265 catalyzed first by pyrroline-5- carboxylate synthase (P5CS) and then by pyrroline-5-
266 carboxylate reductase (P5CR).²³ A strict correlation between the isotopic composition of sugars
267 and amino acids is expected, given the biosynthetic path described above.

268 To investigate this relationship in real must samples, we considered 59 authentic musts cover
269 all of Italy and from two harvest years. The $\delta^{13}\text{C}$ values of both sugar and proline ranged
270 between -30‰ and -23‰, in line with the botanical origin of the matrix. Indeed, the *Vitis*
271 *Vinifera* species belongs to plants with a C3 photosynthetic cycle and its $\delta^{13}\text{C}$ normally ranges
272 from -29‰ to -25‰.³⁰

273 As reported in Table 1, it seems there is not a big isotopic fractionation between sugar fraction
274 and proline. The $\Delta^{13}\text{C}$ ($\delta^{13}\text{C}$ sugar - $\delta^{13}\text{C}$ proline) varies in a narrow range between -1.7‰ and
275 +1.6‰.

276 By comparing the $\delta^{13}\text{C}$ values of the sugar fraction with the $\delta^{13}\text{C}$ of the amino acid proline, we
277 obtained a significant ($P < 0.01$, $R^2=0.71$) linear relationship ($\delta^{13}\text{C}$ sugar = $0.70 * \delta^{13}\text{C}$ proline -
278 7.65; Figure 1). We can define a threshold value for the relationship, calculating 95% of the
279 confidence interval of the regression line from the following equation:

$$280 \quad y = 0.70x - 7.65 - 2*s$$

281 where “ $y = 0.70x - 7.65$ ” is the linear regression model obtained from the 59 data points, “2”
282 is the Student t and “s” is the standard deviation of the residues (difference between calculated
283 and observed value), which in this case is 1.59.

284 As the addition of exogenous cane sugar to must changes only the $\delta^{13}\text{C}$ of the sugar fraction
285 and not that of proline, as here demonstrated (Supplementary Table 2S), the fraudulent practice
286 of sugar addition changes this relationship, which could go beyond the threshold value, even if
287 the $\delta^{13}\text{C}$ of the sugar fraction is not outside the upper limit defined by the wine databank (EU
288 Regulations 273 and 274/2018) and by the natural grape variability (Guideline for Grape Juice
289 of the Association of the Industries of Juices and Nectars from Fruits and Vegetables (AIJN))
290 and equal to -23‰.

291 To demonstrate the possibility of improving chaptalisation detection, we simulated the
292 adulteration of the 59 musts by adding an increasing % of cane sugar ($\delta^{13}\text{C} = +12\text{‰}$) and
293 calculated the number of samples identified as adulterated with cane sugar, both on the basis
294 of the $\delta^{13}\text{C}$ value of the sugar fraction (which must be lower than -23‰) and that of the
295 relationship between the $2 \delta^{13}\text{C}$ (Figure 2). In all cases, the relationship improved detection of
296 the chaptalisation of must. With an addition of 20% cane sugar this new method made it
297 possible to identify all of our adulterated samples as actually adulterated, while with an addition
298 between 5 and 20% the detection increases from a minimum of 6% (5 samples rather than 1
299 with 5% addition) to 72% (59 samples rather than 16 with 20% addition) (Figure 2).

300 $\delta^{13}\text{C}$ analysis of both the sugar fraction and the amino acid proline and their correlation can
301 therefore be considered as a reliable internal standard for improving detection of the fraudulent
302 addition of sugar to must.

303 Our results can be used to check the authenticity of both grape must and wine. Indeed, the
304 proline utilization by *Saccharomyces cerevisiae* as source of nitrogen requires the presence of
305 oxygen and therefore, in anaerobic fermentation conditions, as happens in wine production,
306 proline is conserved in wine without isotopic fractionation.⁸ Moreover, the $\delta^{13}\text{C}$ of wine
307 ethanol is strictly correlated with that of the relevant must sugar with a mean difference of
308 1.7‰ between them.²⁴ Therefore we can calculate both regression line and 95% confidence
309 limits also for $\delta^{13}\text{C}$ of ethanol vs $\delta^{13}\text{C}$ of proline in wine, by predicting $\delta^{13}\text{C}$ of ethanol from
310 $\delta^{13}\text{C}$ of sugar (Figure 1).

311 The isotopic ratio of carbon ($^{13}\text{C}/^{12}\text{C}$, expressed as $\delta^{13}\text{C}$) and nitrogen ($^{15}\text{N}/^{14}\text{N}$, expressed as
312 $\delta^{15}\text{N}$) were also analysed in proline, with the aim to evaluate the power of this marker to trace
313 the geographical origin of grapes. We investigated $\delta^{13}\text{C}$ of proline patterns in grape must across
314 Italy (Figure 3A). Carbon isotope measurements described a gradient of more depleted values
315 in the north Italy to more enriched values in the south Italy (Figure 3A). This was not surprising

316 since stable carbon isotope ratios of plant materials, are primarily related to the photosynthetic
317 pathway used by a plant, even if $\delta^{13}\text{C}$ in foodstuffs exhibits some geographical dependence
318 linked to water stress and humidity during cultivation, although these differences are often very
319 small in comparison to other isotopes.²⁵

320 As reported in Figure 3B it is not possible to identify a strictly correlation between $\delta^{15}\text{N}$ of
321 proline and the region from which the sample comes from. This could be due to the different
322 agricultural practices adopted even within neighboring areas (e.g. organic or chemical
323 fertilisation) or to the pedological characteristics of the soil. Indeed, inorganic fertilisers have
324 $\delta^{15}\text{N}$ values close to those of atmospheric N_2 (from -6‰ to +6‰), whereas manure and other
325 organic fertilisers can be substantially enriched (from +1‰ to +37‰).²⁶

326 Paolini et al. reported that, despite nitrogen isotope fractionation through the chain soil - wine,
327 the $\delta^{15}\text{N}$ values of leaves, grapes, wine and in particular of proline in must and wine maintain
328 the variability of $\delta^{15}\text{N}$ in the growing soil.⁸ Samples from Sardinia have the higher $\delta^{15}\text{N}$ values
329 (+9.8‰). This is in line with previous data reported for other matrices coming from this region,
330 such as casein from pecorino cheese.²⁷ The main reason could be the water stress due to the
331 high temperature, which affects the biological turnover of nitrogen isotopes.²⁸

332 **Study 2**

333 Myo-inositol, a carbocyclic sugar, is synthesized in grapes from glucose-6-phosphate (G-6-P)
334 in two steps. First, an inositol-3-phosphate synthase enzyme (e.g. ISYNA1) isomerize the G-
335 6-P to myo-inositol 1-phosphate, which is then finally dephosphorylate to give free myo-
336 inositol by an inositol monophosphatase enzyme (e.g. IMPA1). It is normally abundant in must
337 grape (more than 750 mg/kg sugar) while its stereoisomer scyllo-inositol is less abundant (more
338 than 38 mg/kg sugar).^{17,29} In addition, a ratio of 20 or lower between myo- and scyllo-inositol
339 has been suggested as authenticity index.¹⁷ These limits of content are a useful routine control
340 tool of CRM, but they can be easily falsified by fraudulent addition to the must of pure

341 commercial polyalcohols in the right concentration. To describe the natural $\delta^{13}\text{C}$ variability of
342 these must components, several samples of authentic CRM were analysed. In all these samples
343 the myo- and scyllo-inositol content was measured and all of them fell within the limits
344 suggested by Monetti et al. (Table 2).¹⁷

345 As reported in Table 2 myo-inositol and scyllo-inositol, that differ chemically only in the three-
346 dimensional orientations of their atoms in space, showed a similar $\delta^{13}\text{C}$ range between -29.4‰
347 and -25.0‰, which is typical for plants with a C3 photosynthetic cycle.³⁰ The maximum
348 difference between the two $\delta^{13}\text{C}$ is 2.4‰ and it can be used as a limit beyond which an addition
349 of one of the two polyalcohols can be suspected. In addition, a limit of -25‰ for both the
350 polyalcohols could be proposed (Figure 4). While commercial myo-inositol shows an average
351 value of $\delta^{13}\text{C}$ of $-29.0 \pm 0.3\text{‰}$ probably due to the origin of the grapes from which the
352 commercial product is extracted at low cost, the $\delta^{13}\text{C}$ values of scyllo-inositol is typical of C4
353 plants ($-11.8 \pm 0.3\text{‰}$). Scyllo-inositol is rare and expensive, not being widely available from
354 commercial sources, and for this reason it must be synthesized. Several synthetic approaches
355 to produce scyllo-inositol are known.³¹ Rodriguez et al. present a concise synthesis of scyllo-
356 inositol starting from inexpensive D-glucose.³²

357 Commercial glucose is normally produced via the enzymatic hydrolysis of starch,³³ that
358 belongs to the C4 plant with a range of variability between -10 and -16‰,³⁰ and this could
359 justify the values found for scyllo-inositol. The illegal addition of this polyalcohol should be
360 easily identified. A difference between the two $\delta^{13}\text{C}$ higher than 2.4‰ or a value of $\delta^{13}\text{C}$ scyllo-
361 inositol higher than -25‰ could be interpreted as an index of adulteration. Figure 4 shows the
362 isotopic composition of myo- and scyllo-inositol of commercial CRM samples compared with
363 the authentic one. Only one sample falls within the variability limit, while five samples showed
364 higher $\delta^{13}\text{C}$ scyllo-inositol values (higher than -25‰). Sample Q (Table 2) is characterized by

365 a high $\delta^{13}\text{C}$ myo-inositol value. Maybe in this case, too, commercial myo-inositol from a C4
366 plant source was used.

367 $\delta^{13}\text{C}$ analysis of proline by GC/c-IRMS combine with the analysis of the sugar fraction
368 represents a novel analytical tool to support and improve the detection of fraudulent addition
369 of cane sugar to must and wine. While $\delta^{13}\text{C}$ of proline could be useful as geographical indicator
370 while $\delta^{15}\text{N}$ of proline seems too correlate with the agronomic practices adopted. Moreover, the
371 compound-specific analysis of $\delta^{13}\text{C}$ of myo and scyllo-inositol represents a useful tool to
372 identify the illicit addition of these two polyalcohol in must concentrated obtained not from
373 grape (e.g. from cereal) to mime the composition of an authentic grape CRM.

374 **Abbreviation used**

375 CRM – concentrated rectified must

376 GC - gas chromatographyI

377 RMS - Isotopic Ratio Mass Spectrometry

378 **Supporting information description**

379 Table 1S: Repeatability of methods ($\delta^{13}\text{C}$ data corrected)

380 Table 2S: Effect of addition of sugar cane to a grape must on the $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ value of proline

381 Figure 1S: Map of sampling

382 Novelty statement

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486 Figure caption.

487 Figure 1: Correlation between $\delta^{13}\text{C}$ of proline and sugar fraction. The limit accepted by the
488 Association of the Industries of Juices and Nectars from Fruits and Vegetables (AIJN)
489 and the limit calculated based on the 95% of the confidence intervals were reported as
490 dotted line.

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492 Figure 2: Improvement in must chaptalization detection by the number of samples identified
493 as adulterated with cane sugar.

494 Figure 3: Geographical variability of $\delta^{13}\text{C}$ (A) and $\delta^{15}\text{N}$ (B) of proline in authentic Italian
495 grape must.

496 Figure 4: Variability of $\delta^{13}\text{C}$ of myo and scyllo inositol of authentic grape CRM and samples
497 collected on the market.

498 Figure 1S: Map of sampling

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525 Table 1 Experimental values of the isotope composition $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ of proline, $\delta^{13}\text{C}$ of the
 526 sugar fraction and variation between the two $\delta^{13}\text{C}$

Number	Year of Harvest	Italian region	Variety of grape	$\delta^{15}\text{N}$ proline (‰. vs AIR)	$\delta^{13}\text{C}$ proline (‰. vs V-PDB)	$\delta^{13}\text{C}$ sugar (‰. vs V-PDB)	Δ $\delta^{13}\text{C}$ sugar - $\delta^{13}\text{C}$ proline
1	2016	Veneto	Oseleta	5.6	-24.2	-24.7	-0.5
2	2016	Veneto	Corvinone	4.2	-25.2	-24.9	0.3
3	2016	Veneto	Manzoni	4.7	-27.0	-28.1	-1.1
4	2016	Veneto	Chardonnay	4.8	-27.0	-26.5	0.5
5	2016	Friuli VG	Sauvignon	0.2	-27.6	-27.7	-0.1
6	2016	Friuli VG	Malvasia	0.0	-29.0	-28.2	0.8
7	2016	Piedmont	Nebbiolo	0.6	-28.8	-27.1	1.7
8	2016	Piedmont	Nebbiolo	-0.5	-28.8	-27.4	1.4
9	2016	Piedmont	Nebbiolo	-0.1	-28.8	-27.6	1.2
10	2016	Piedmont	Nebbiolo	-0.3	-26.2	-24.7	1.5
11	2016	Piedmont	Riesling	1.6	-27.8	-26.2	1.6
12	2016	Lombardy	Pinot	4.9	-27.4	-28.2	-0.8
13	2016	Lombardy	Chardonnay	0.5	-28.6	-27.6	1.0
14	2016	Emilia Romagna	Barbera	-1.5	-25.8	-26.1	-0.3
15	2016	Emilia Romagna	Cabernet	-2.5	-26.6	-26.6	0.0
16	2016	Toscany	Sangiovese	4.7	-25.8	-25.7	0.1
17	2016	Toscany	Sangiovese	4.0	-26.2	-25.6	0.6
18	2016	Lazio	Syrah	2.6	-26.2	-25.7	0.5
19	2016	Lazio	Vermentino	0.6	-27.0	-26.0	1.0
20	2016	Lazio	Malvasia	2.4	-25.0	-26.1	-1.1
21	2016	Lazio	Montepulciano	0.8	-28.8	-28.1	0.7
22	2016	Marche	Verdicchio	4.4	-28.8	-27.2	1.6
23	2016	Marche	Verdicchio	6.5	-30.0	-28.4	1.6
24	2016	Marche	Montepulciano	4.5	-29.4	-28.7	0.7
25	2016	Marche	Sangiovese	3.7	-29.8	-28.8	1.0
26	2016	Marche	Verdicchio	2.8	-28.4	-27.7	0.7
27	2016	Abruzzo	Trebbiano	3.2	-28.6	-27.3	1.3
28	2016	Umbria	Grechetto	3.7	-27.6	-27.6	0.0
29	2016	Umbria	Sagrantino	2.4	-28.8	-28.9	-0.1
30	2016	Campania	Aglianico	4.5	-25.2	-25.9	-0.7
31	2016	Puglia	Primitivo	6.5	-26.0	-25.4	0.6
32	2016	Puglia	Primitivo	5.0	-26.0	-25.6	0.4
33	2016	Sicily	Insolia	3.7	-24.2	-25.7	-1.5
34	2016	Sicily	Insolia	2.0	-26.4	-25.4	1.0
35	2016	Sardinia	Vermentino	6.4	-26.6	-25.6	1.0
36	2016	Sardinia	Monica	9.8	-24.2	-23.3	0.9
			Mean	3.0	-27.2	-26.7	0.5
			SD	2.6	1.6	1.4	0.8
			Min	-2.5	-30.0	-28.9	-1.5
			Max	9.8	-24.2	-23.3	1.7

Number	Year of Harvest	Italian region	Variety of grape	$\delta^{15}\text{N}$ proline (‰. vs AIR)	$\delta^{13}\text{C}$ proline (‰. vs V-PDB)	$\delta^{13}\text{C}$ sugar (‰. vs V-PDB)	Δ $\delta^{13}\text{C}$ sugar - $\delta^{13}\text{C}$ proline
1	2017	Trentino AA	Muller Thurgau	8.0	-28.4	-28.7	-0.3
2	2017	Trentino AA	Pinot grigio	6.5	-26.4	-27.5	-1.1
3	2017	Trentino AA	Kerner	9.2	-27.2	-28.4	-1.2
4	2017	Veneto	Chardonnay	2.8	-27.0	-26.6	0.4
5	2017	Veneto	Chardonnay	1.2	-24.6	-25.8	-1.2
6	2017	Friuli VG	Chardonnay	-1.0	-28.4	-27.9	0.5
7	2017	Piedmont	Riesling	2.4	-26.2	-27.2	-1.0
8	2017	Piedmont	Nebbiolo	0.4	-25.4	-26.7	-1.3
9	2017	Piedmont	Nebbiolo	2.9	-27.8	-27.4	0.4
10	2017	Piedmont	Nebbiolo	4.5	-27.2	-26.8	0.4
11	2017	Piedmont	Nebbiolo	5.4	-25.8	-24.9	0.9
12	2017	Lombardy	Pinot	5.8	-27.0	-26.8	0.2
13	2017	Toscany	Sangiovese	2.6	-23.6	-23.4	0.2
14	2017	Lazio	Malvasia	5.2	-27.0	-26.8	0.2
15	2017	Lazio	Montepulciano	2.6	-28.0	-27.7	0.3
16	2017	Marche	Verdicchio	6.0	-27.0	-25.6	1.4
17	2017	Marche	Verdicchio	8.0	-27.2	-25.5	1.7
18	2017	Marche	Verdicchio	3.7	-24.6	-26.2	-1.6
19	2017	Abruzzo	Montepulciano	4.1	-24.6	-24.6	0.0
20	2017	Abruzzo	Montepulciano	6.2	-23.4	-23.0	0.4
21	2017	Abruzzo	Montepulciano	6.2	-25.2	-23.8	1.4
22	2017	Umbria	Grechetto	5.4	-24.4	-25.4	-1.0
23	2017	Campania	Aglianico	2.1	-24.4	-24.2	0.2
			Mean	4.4	-26.1	-26.1	0.0
			SD	2.6	1.5	1.6	0.9
			Min	-1.0	-28.4	-28.7	-1.6
			Max	9.2	-23.4	-23.0	1.7

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543 Table 2 Experimental values of the isotope composition $\delta^{13}\text{C}$ of myo and scyllo inositol and
 544 variation between the two $\delta^{13}\text{C}$ in authentic samples of CRM and in samples from the market.

Sample	Geographical origin	$\delta^{13}\text{C}$ scyllo inositol (‰. vs V-PDB)	$\delta^{13}\text{C}$ myo-inositol (‰. vs V-PDB)	Δ $\delta^{13}\text{C}$ scyllo - $\delta^{13}\text{C}$ myo	
Authentic CRM samples	A	Puglia	-28.6	-28.2	-0.4
	B	Italy	-28.6	-26.2	-2.4
	C	Italy	-28.6	-29.4	0.8
	D	Italy	-28.2	-29.4	1.2
	E	Italy	-27.0	-27.4	0.4
	F	Italy	-27.0	-28.2	1.2
	G	Italy	-26.6	-29.0	2.4
	H	Italy	-25.0	-25.8	0.8
	I	Sicily	-25.0	-26.2	1.2
	Mean	-27.2	-27.8	0.6	
	SD	1.5	1.4	1.3	
	Min	-28.6	-29.4	-2.4	
	Max	-25.0	-25.8	2.4	
Market's CRM	L	Italy	-23.4	-25.8	2.4
	M	Italy	-22.2	-27.8	5.6
	N	Italy	-21.8	-27.8	6.0
	O	Spain	-21.8	-27.4	5.6
	P	Spain	-19.0	-23.4	4.4
	Q	Spain	-25.0	-21.0	-4.0
	R	Spain	-25.8	-25.8	0.0

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566 Table 1S: Repeatability of methods ($\delta^{13}\text{C}$ data corrected)

	$\delta^{13}\text{C}$ proline (‰, vs V- PDB)	$\delta^{15}\text{N}$ proline (‰, vs AIR)	$\delta^{13}\text{C}$ scyllo inositol (‰, vs V- PDB)	$\delta^{13}\text{C}$ myo- inositol (‰, vs V- PDB)
1	-24.4	5.3	-28.6	-29.0
2	-25.4	5.5	-28.7	-29.1
3	-25.4	5.2	-29.0	-29.0
4	-25.0	5.3	-28.5	-28.6
5	-24.4	5.3	-28.5	-28.6
6	-24.4	5.7	-29.0	-28.5
7	-25.4	5.2	-28.6	-28.7
8	-24.8	5.3	-28.2	-28.7
9	-25.0	5.4	-28.5	-29.0
10	-25.8	5.5	-28.6	-29.0
Mean	-25.0	5.4	-28.6	-28.8
SD	0.5	0.2	0.2	0.2

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596 Table 2S: Effect of addition of sugar cane to a grape must on the $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$ value of
597 proline

% w/w of cane added to must	$\delta^{13}\text{C}$ proline (‰, vs V- PDB)	$\delta^{15}\text{N}$ proline (‰, vs AIR)
0	-24.4	5.4
14	-25.0	5.3
28	-24.4	5.5
32	-24.4	5.2

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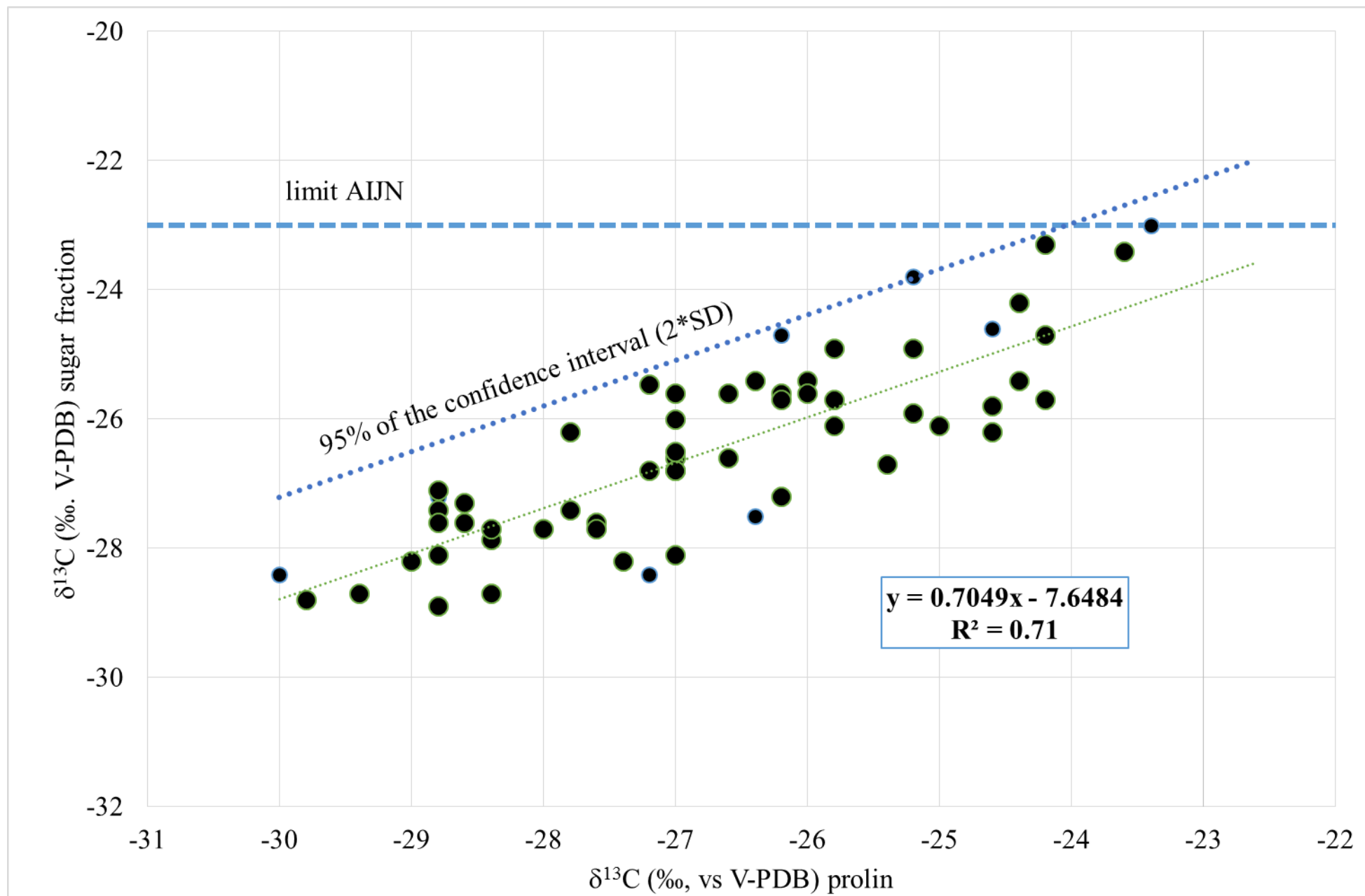
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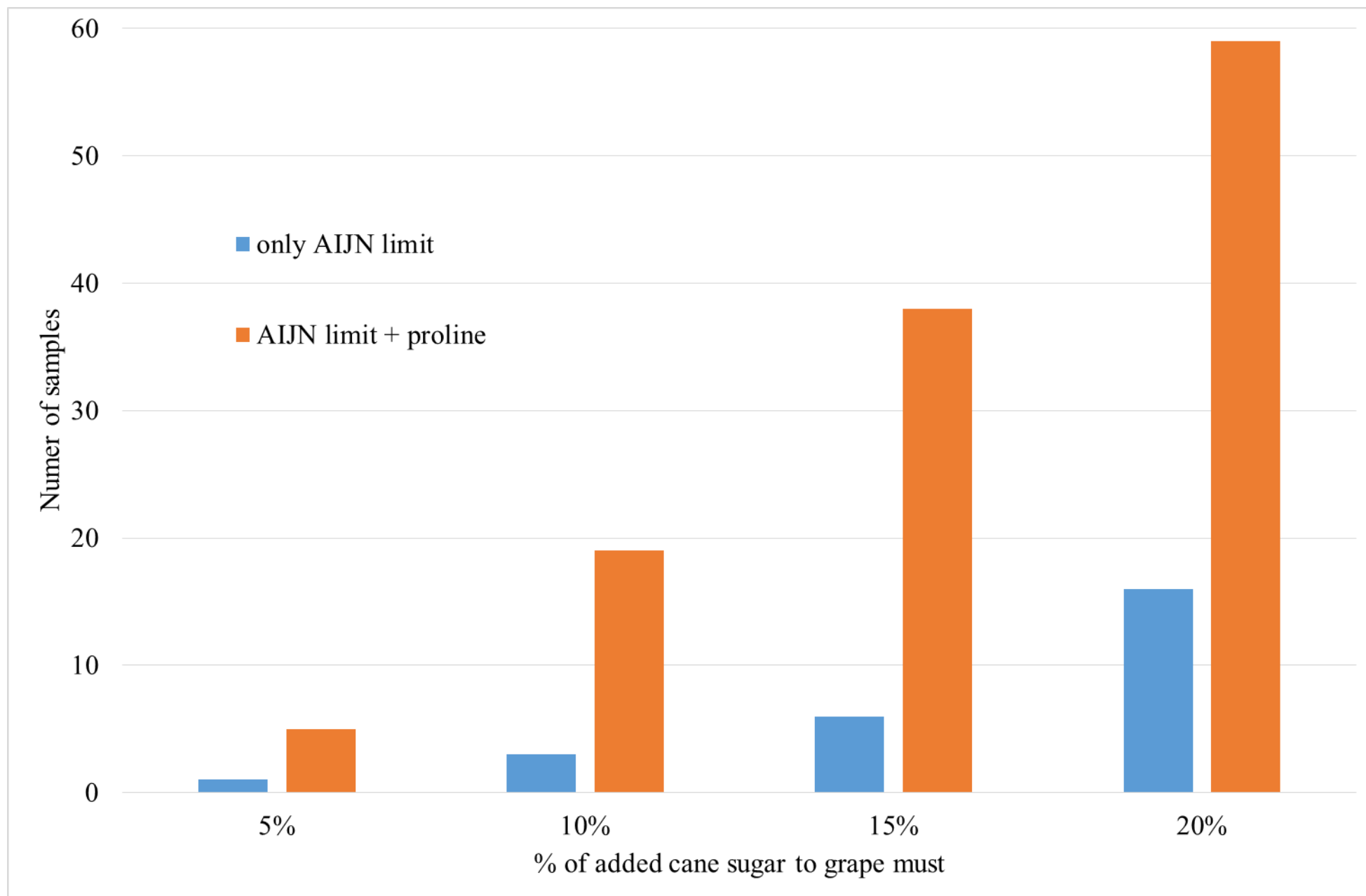
633 **Graphic for table of content**

634

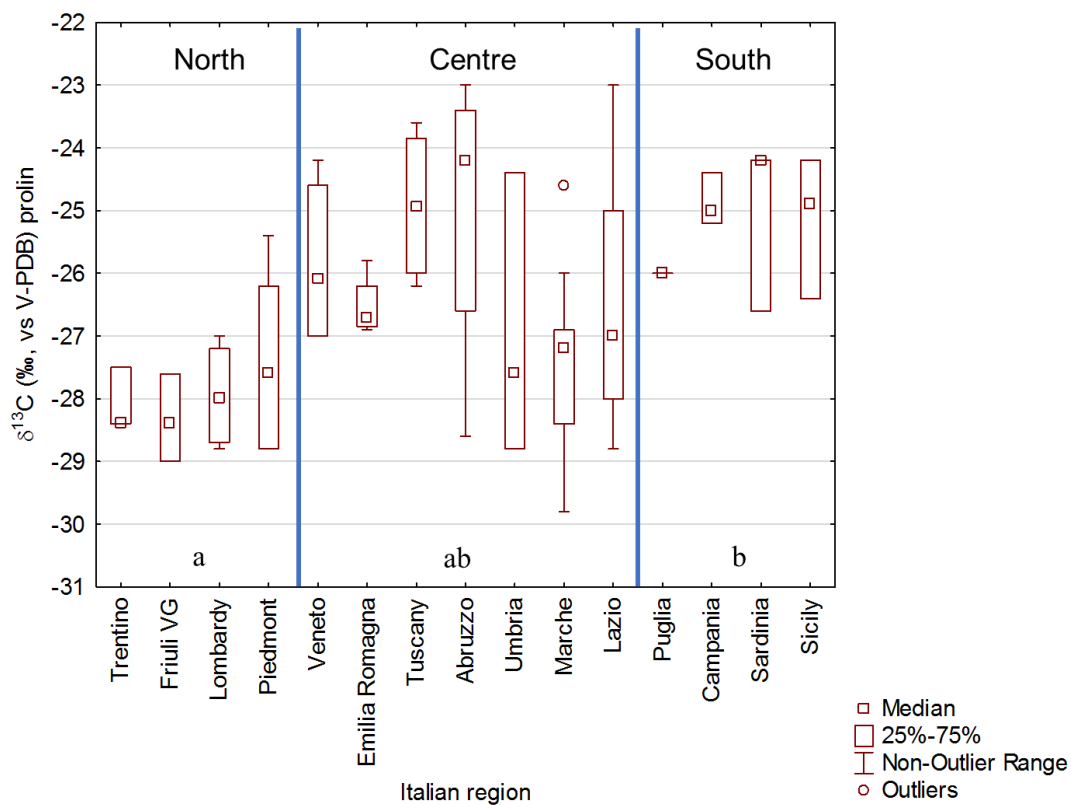


635





A



B

