## Supplementary Data

# Streptogramin A derivatives as mitochondrial translation inhibitors to suppress glioblastoma stem cells growth 

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Fig. S1. The molecular structures of each of the four possible diastereomers generated by the Diels-Alder reactions between dalfopristin (1) or virginiamycin M1 (2) and maleic anhydride or maleimide as dienophiles. The stereochemistry of each possible diastereomer obtainable from the reaction is highlighted.


Fig. S2. Per-residue interaction energy decomposition performed on the docking-predicted binding mode for C10/C13 derivatives. VM1 (2) binding mode from structure 6I9R is also reported for reference. In detail, the upper panel reports a per-residue decomposition of electrostatic interaction energy: values (in $\mathrm{Kcal} / \mathrm{mol}$ ) are plotted according to the colour scale reported on the right, ranging from red (positive value, repulsive interaction) to black (negative value, attractive interaction). The lower panel, instead, reports a per-residue decomposition of hydrophobic interactions: values (in arbitrary/adimensional units) are plotted according to the color scale reported on the right, ranging from white (zero, absence of interaction) to dark teal (indicating a good hydrophobic interaction).


Fig. S3. Per-residue interaction energy decomposition performed on the docking-predicted binding mode for C16 derivatives. VM1 (2) binding mode from structure 6I9R is also reported for reference. In detail, the upper panel reports a per-residue decomposition of electrostatic interaction energy: values (in $\mathrm{Kcal} / \mathrm{mol}$ ) are plotted according to the colour scale reported on the right, ranging from red (positive value, repulsive interaction) to black (negative value, attractive interaction). The lower panel, instead, reports a per-residue decomposition of hydrophobic interactions: values (in arbitrary/adimensional units) are plotted according to the color scale reported on the right, ranging from white (zero, absence of interaction) to dark teal (indicating a good hydrophobic interaction).


Fig. S4. Per-residue interaction energy decomposition performed on the docking-predicted binding mode for C26 derivatives. VM1 (2) binding mode from structure 6I9R is also reported for reference. In detail, the upper panel reports a per-residue decomposition of electrostatic interaction energy: values (in $\mathrm{Kcal} / \mathrm{mol}$ ) are plotted according to the colour scale reported on the right, ranging from red (positive value, repulsive interaction) to black (negative value, attractive interaction). The lower panel, instead, reports a per-residue decomposition of hydrophobic interactions: values (in arbitrary/adimensional units) are plotted according to the color scale reported on the right, ranging from white (zero, absence of interaction) to dark teal (indicating a good hydrophobic interaction).


Fig. S5. Western blot analysis for the evaluation of mitochondrial translation inhibition of $\mathrm{C} 10 / \mathrm{C} 13$ derivatives. $\mathrm{C} 10 / \mathrm{C} 13$ derivatives do not inhibit mitochondrial translation. Immunoblotting on COMI cells after 48 hours of treatment with increasing concentrations of $\mathbf{1 a}, \mathbf{1 b}, \mathbf{2 a}$, and $\mathbf{2 c}$, alone (left column) or in combination with $\mathrm{Q} \mathbf{( 3 )}$ (right column). Effects on COX1, COX4 and beta tubulin proteins are shown. One representative result, $\mathrm{n}=3$ biological replicates.


Fig. S6. Western blot analysis for the evaluation of mitochondrial translation inhibition of C16 hydroxyl derivatives. C16 hydroxyl derivatives inhibit mitochondrial translation. Immunoblotting on COMI cells after 48 hours of treatment with increasing concentrations of $(\mathbf{1 6 R})-\mathbf{1 c},(\mathbf{1 6 S})-\mathbf{1 c},(\mathbf{1 6 R})-\mathbf{2 c}$, and $\mathbf{( 1 6 S )} \mathbf{- 2} \mathrm{c}$, alone (left column) or in combination with Q (3) (right column). Effects on COX1, COX4 and beta tubulin proteins are shown. One representative result, $\mathrm{n}=3$ biological replicates.


Fig. S7. Western blot analysis for the evaluation of mitochondrial translation inhibition of (16S) fluoro derivatives. (16S) fluoro derivatives inhibit mitochondrial translation. Immunoblotting on COMI cells after 48 hours of treatment with increasing concentrations of (16S)-1e, and (16S)-2e, alone (left column) or in combination with Q (3) (right column). Effects on COX1, COX4 and beta-tubulin proteins are shown. One representative result, $\mathrm{n}=3$ biological replicates.


Fig. S8. Western blot analysis for the evaluation of mitochondrial translation inhibition of (16R) NHMe derivatives. (16R) NHMe derivatives do not inhibit mitochondrial translation. Immunoblotting on COMI cells after 48 hours of treatment with increasing concentrations of (16R)-1d, and (16R)-2d, alone (left column) or in combination with Q (3) (right column). Effects on COX1, COX4 and beta-tubulin proteins are shown. One representative result, $\mathrm{n}=3$ biological replicates.


Fig. S9. Western blot analysis for the evaluation of mitochondrial translation inhibition of $\mathbf{6}$ and $\mathbf{7}$ derivatives. Compounds 6 and 7 inhibit mitochondrial translation only at high concentrations. Immunoblotting on COMI cells after 48 hours of treatment with increasing concentrations of 6 and 7. Effects on COX1, COX4 and beta tubulin proteins are shown. One representative result, $\mathrm{n}=3$ biological replicates.


Fig. S10. UHPLC chromatograms of a) drug-free sample (blank); b) standard sample used for calibration curve of VM1 (2) $(50 \mathrm{ng} / \mathrm{mL})$; c) sample treated with VM1 (2). UHPLC conditions: Hypersil Gold C18 column (Thermofisher, 100 x 2.1 mm , particle size: $1.9 \mu \mathrm{~m}$ ); linear gradient from 5 to $100 \%$ B (B: acetonitrile $0.1 \%$ of formic acid; A: water $+0.1 \%$ formic acid) over 15 min , followed by 4 min at $100 \%$; flow $0.2 \mathrm{~mL} / \mathrm{min}$; MS spray voltage 3500 V ; full scan at 120.000 FWHM ( $200 \mathrm{~m} / \mathrm{z}$ ); scan range $100-1000 \mathrm{~m} / \mathrm{z}$.


Fig. S11. Extracted ion chromatograms of the $[\mathrm{M}+\mathrm{H}]^{+}$adducts (Skyline software) of a representative standard sample of VM1 (2) conc. $50 \mathrm{ng} / \mathrm{mL}$ (on the left) and sample treated with VM1 (2) (on the right). The peak area of the precursor ions were extracted from chromatograms for both standards and treated samples to build the standard curves of each compound and to compute concentration values of the compound in samples, respectively.


Fig. S12. Calibration curve for VM1 (2); $\mathrm{R}^{2}=0.9917$. Five serial drug dilutions, from 0.005 to $50 \mathrm{ng} / \mu \mathrm{L}$, were prepared in non-treated metabolite extracts resuspended in $5 \%$ acetonitrile/ $0.1 \%$ formic acid aqueous solution starting from a 10 mM stock of VM1 (2). $20 \mu \mathrm{~L}$ of each dilution plus a blank samples were loaded into the LC autosampler. After the analysis, the log-transformed value of the total area of precursor ions of each standard sample was plotted in function of the concentration to construct the standard curve. The error on concentrations was calculated as the physical error deriving from the weighting of the compound plus the error from pipetting. For each serial dilution performed, the error was propagated using the variance formula.

Table S1
UHPLC-MS retention time (min) returned by Skyline software for extracted peak of each compound.

|  | RT |  | RT |
| :---: | :---: | :---: | :---: |
| $\mathbf{D ~ ( 1 ) ~}$ | 6.9 | VM1 (2) | 9.2 |
| $\mathbf{1 a}$ | 6.6 | $\mathbf{2 a}$ | 8.6 |
| $\mathbf{( 1 6 R ) - 1 \mathbf { c }}$ | 6.6 | $\mathbf{( 1 6 R ) - 2 c}$ | 8.3 |
| $\mathbf{( 1 6 R ) - 1 e}$ | 7.3 | $\mathbf{( 1 6 R )} \mathbf{- 2 e}$ | 9.3 |
| $\mathbf{5}$ | 7.0 |  |  |
| $\mathbf{6}$ | 6.2 |  |  |



${ }^{1} \mathrm{H}$ NMR（ $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ）

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${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$




Fig．S13．${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{1}$ ．
${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

$\mathrm{CH}_{2} \mathrm{Cl}_{2}$



Fig. S14. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 2.


${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ )
 ,

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ )



Fig. S15. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{1 a}$.


Fig. S16. High resolution $\mathrm{ESI}(+)$ MS spectrum of compound 1a.


Fig. S17. HPLC chromatogram of compound 1a (RP18 column, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )




$\begin{array}{lllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & \mathrm{ppm}\end{array}$
Fig. S18. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{1 b}$.


Fig. S19. High resolution ESI $(+)$ MS spectrum of compound 1b.


Totals :
1.50717 e 42077.11069

Fig. S20. HPLC chromatogram of compound 1b (RP18 column, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).


Fig. S21. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 2a.


Fig. S22. High resolution $\operatorname{ESI}(+)$ MS spectrum of compound 2a.


Fig. S23. HPLC chromatogram of compound 2a (RP18 column, gradient water/acetonitrile 30:70 to 80:20).

$\left.\begin{array}{ccccccccc}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 \\ \hline\end{array} \begin{array}{llll}120\end{array}\right]$


Fig. S25. High resolution ESI $(+)$ MS spectrum of compound 2b.


Fig. S26. HPLC chromatogram of compound 2b (RP18, gradient water/acetonitrile 30:70 to 80:20).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )






${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$\mathrm{CDCl}_{3}$
$\begin{array}{lllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & \mathrm{ppm}\end{array}$

Fig. S27. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16R)-1c.


Fig. S28. High resolution ESI( + ) MS spectrum of compound (16R)-1c.


Fig. S29. HPLC chromatogram of compound (16R)-1c (RP18, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).

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${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


Fig. S30. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-1c.


Fig. S31. High resolution ESI $(+$ ) MS spectrum of compound (16S)-1c.


Fig. S32. HPLC chromatogram of compound (16S)-1c (RP18, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).
(
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(16R)-2c
$\mathrm{CDCl}_{3}$
$\begin{array}{llllllllllllllllllll} \\ 210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20\end{array}$

Fig. S33. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16R)-2c.


Fig. S34. High resolution $\mathrm{ESI}(+) \mathrm{MS}$ spectrum of compound (16R)-2c.


Signal 2: DAD1 B, Sig=254,16 Ref=off

| $\begin{gathered} \text { Peak } \\ \# \end{gathered}$ | $\begin{gathered} \text { RetTime } \\ {[\mathrm{min}]} \end{gathered}$ | Type | Width [min] | $\begin{gathered} \text { Area } \\ {\left[m A U^{\star} s\right]} \end{gathered}$ | Height [mAU] | $\begin{gathered} \text { Area } \\ \% \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 6.974 |  | 0.1502 | 1.11277 e 4 | 1222.59326 | 99.5675 |
| 2 | 7.533 | VV | 0.1482 | 48.33984 | 5.41012 | 0.4325 |

Totals : $\quad 1.11760 \mathrm{e} 41228.00338$

Fig. S35. HPLC chromatogram of compound (16R)-2c (RP18, gradient water/acetonitrile 30:70 to 80:20).
H NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(16S)-2c

## $\mathrm{CDCl}_{3}$

$\begin{array}{lllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & \mathrm{ppm}\end{array}$

Fig. S36. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-2c.


Fig. S37. High resolution $\mathrm{ESI}(+)$ MS spectrum of compound (16S)-2c.


Fig. S38. HPLC chromatogram of compound (16S)-2c (RP18, gradient water/acetonitrile 30:70 to 80:20).
${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )



Fig. S39. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16R)-1d.


Fig. S40. High resolution ESI(+) MS spectrum of compound (16R)-1d.


Fig. S41. HPLC chromatogram of compound (16R)-1d (RP18, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$\mathrm{CDCl}_{3}$



Fig. S42. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-1d.


Fig. S43. High resolution $\mathrm{ESI}(+)$ MS spectrum of compound (16S)-1d.


Fig. S44. HPLC chromatogram of compound (16S)-1d (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).

${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


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Fig. S46. High resolution ESI(+) MS spectrum of compound (16R)-2d.


Fig. S47. HPLC chromatogram of compound (16R)-2d (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).

$\mathrm{CHCl}_{3}$


${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )
$\mathrm{CD}_{3} \mathrm{OD}$



Fig. S48. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-2d.


Fig. S49. High resolution $\mathrm{ESI}(+)$ MS spectrum of compound (16S)-2d.


| Peak RetTime Type \# [min] | Width [min] | $\begin{gathered} \text { Area } \\ {\left[\mathrm{mAU}^{\star} \mathrm{S}\right]} \end{gathered}$ | Height [mAU] | Area $\%$ |
| :---: | :---: | :---: | :---: | :---: |
| $1 \quad 15.756 \mathrm{BB}$ | 0.2148 | 899.62329 | 64.46941 | 100.0000 |
| Totals: |  | 899.62329 | 64.46941 |  |

Fig. S50. HPLC chromatogram of compound (16S)-2d (RP18, isocratic water/acetonitrile 30:70 $+1 \%$ trifluoroacetic acid).



Fig. S51. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16R)-1e.


Fig. S52. High resolution $\mathrm{ESI}(+) \mathrm{MS}$ spectrum of compound (16R)-1e.


Fig. S53. HPLC spectrum of compound (16R)-1e (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


$\qquad$





Fig. S54. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-1e


Fig. S55. High resolution $\mathrm{ESI}(+)$ MS spectrum of compound (16S)-1e.


Fig. S56. HPLC chromatogram of compound (16S)-1e (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).


${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(16R)-2e
$\mathrm{CHCl}_{3}$
$\mathrm{H}_{2} \mathrm{O}$
 In 1
C NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(16R)-2e


Fig. S58. High resolution $\mathrm{ESI}(+) \mathrm{MS}$ spectrum of compound (16R)-2e.


Fig. S59. HPLC chromatogram of compound (16R)-2e (RP18, gradient water/acetonitrile 30:70 to 80:20).


${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(16S)-2e
$\mathrm{CDCl}_{3}$


Fig. S60. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound (16S)-2e.


Fig. S61. High resolution ESI $(+)$ MS spectrum of compound (16S)-2e.


Fig. S62. HPLC chromatogram of compound (16S)-2e (RP18, gradient water/acetonitrile 30:70 to 80:20).


Fig. S63. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{1 i}$.


Fig. S64. High resolution $\operatorname{ESI}(+)$ MS spectrum of compound $\mathbf{1 i}$.


Fig. S65. HPLC chromatogram of compound $\mathbf{1 i}$ (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

$2 i$


Fig. S66. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{2 i}$.


Fig. S67. High resolution ESI $(+)$ MS spectra of compound $\mathbf{2 i}$.


Fig. S68. HPLC chromatogram of compound $\mathbf{2 i}$ (RP18, gradient water/acetonitrile 30:70 to 80:20).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


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${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

4

| 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 | 1 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Fig. S69. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 4 .


Fig. S70. High resolution ESI $(+)$ MS spectrum of compound 4.


Fig. S71. HPLC chromatogram of compound 4 (RP18, gradient water/acetonitrile 30:70 to 80:20).





Fig. S72. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 5 .


Fig. S73. High resolution ESI $(+)$ MS spectrum of compound 5.


Fig. S74. HPLC chromatogram of compound 5 (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).

${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$\mathrm{CHCl}_{3}$



Fig. S75. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 6 .


Fig. S76. High resolution ESI( + ) MS spectrum of compound 6 .


Fig. S77. HPLC chromatogram of compound 6 (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).


Fig. S78. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound 7.


Fig. S79. High resolution ESI( + ) MS spectrum of compound 7.


Fig. S80. HPLC chromatogram of compound 7 (RP18, isocratic water/acetonitrile $30: 70+1 \%$ trifluoroacetic acid).

